

Deep ocean ventilation, carbon isotopes, marine sedimentation and the deglacial CO₂ rise

T. Tschumi¹, F. Joos^{1,2}, M. Gehlen³, and C. Heinze^{4,5,6}

¹Climate and Environmental Physics, Physics Institute, University of Bern, Sidlerstrasse 5, 3012 Bern, Switzerland

²Oeschger Centre for Climate Change Research, University of Bern, 3012 Bern, Switzerland

³Laboratoire du Climat et de l'Environnement (LSCE), L'Orme des Merisiers Bât. 712, 91191 Gif-sur-Yvette, France

⁴Geophysical Institute, University of Bergen, Allegaten 70, 5007 Bergen, Norway

⁵Bjerknes Centre for Climate Research, Bergen, Norway

⁶Uni Bjerknes Centre, Uni Research, Bergen, Norway

Received: 25 August 2010 – Published in Clim. Past Discuss.: 27 September 2010

Revised: 14 May 2011 – Accepted: 13 June 2011 – Published: 22 July 2011

Abstract. The link between the atmospheric CO₂ level and the ventilation state of the deep ocean is an important building block of the key hypotheses put forth to explain glacial-interglacial CO₂ fluctuations. In this study, we systematically examine the sensitivity of atmospheric CO₂ and its carbon isotope composition to changes in deep ocean ventilation, the ocean carbon pumps, and sediment formation in a global 3-D ocean-sediment carbon cycle model. Our results provide support for the hypothesis that a break up of Southern Ocean stratification and invigorated deep ocean ventilation were the dominant drivers for the early deglacial CO₂ rise of ~35 ppm between the Last Glacial Maximum and 14.6 ka BP. Another rise of 10 ppm until the end of the Holocene is attributed to carbonate compensation responding to the early deglacial change in ocean circulation. Our reasoning is based on a multi-proxy analysis which indicates that an acceleration of deep ocean ventilation during early deglaciation is not only consistent with recorded atmospheric CO₂ but also with the reconstructed opal sedimentation peak in the Southern Ocean at around 16 ka BP, the record of atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$, and the reconstructed changes in the Pacific CaCO₃ saturation horizon.

1 Introduction

Ancient air samples trapped in polar ice reveal that atmospheric CO₂ has undergone large natural fluctuations with amplitudes of 90–100 ppm between glacial and interglacial periods. Antarctic temperature closely parallels the recorded atmospheric CO₂ concentration during at least the last eight glacial cycles (e.g., Lüthi et al., 2008). The global oceans have been identified

as the key regulator of the atmospheric carbon content on glacial-interglacial time-scales (e.g., Sigman and Boyle, 2000; Archer et al., 2000a; Siegenthaler and Wenk, 1984; Sarmiento and Toggweiler, 1984; Knox and McElroy, 1984). The ocean carbon cycle is driven by a multitude of different processes such as ocean circulation, the ocean carbon pumps and marine sediment formation which equilibrate with atmospheric CO₂ on time-scales of months to millennia. The main focus of this study is to examine the millennial-scale response in atmospheric CO₂ to changes in the ventilation rate of the deep ocean considering the carbon cycle feedbacks from implied modifications in marine biological production and sediment formation.

We perform a series of illustrative sensitivity experiments with the coupled Bern3D+C ocean-sediment carbon cycle model in which the rate of deep ocean ventilation is systematically varied by changing wind forcing. The time-dependent response for multiple carbon cycle proxies, such as $\delta^{13}\text{C}$ - and $\Delta^{14}\text{C}$ -signatures in the ocean and the atmosphere but also



Correspondence to: T. Tschumi
(tschumi@climate.unibe.ch)

the depth of the CaCO₃ saturation horizon and the sedimentation rate of CaCO₃, organic carbon and opal are compared to available paleo-data to assess the importance of changes in deep ocean ventilation for glacial-interglacial CO₂ variations.

Paleo-records of CO₂ carbon isotope composition provide valuable insight into past sources and sinks of the atmospheric carbon reservoir. This information can be used to constrain contributions from the individual processes made responsible for the CO₂ fluctuations (e.g., Elsig et al., 2009; Joos et al., 2004; Hughen et al., 2004; Smith et al., 1999; Leuenberger et al., 1992). Unfortunately, during glacial-interglacial transitions, there were generally several simultaneous processes generating similar imprints on CO₂, $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$. So far, the isotope records of atmospheric CO₂ alone have not allowed for a unique quantitative process identification (Köhler et al., 2006a).

The timing and coevolution of the CO₂ carbon isotope records over the last deglaciation nevertheless indicate that the initial CO₂ increase at the beginning of deglaciations was caused by the injection of some old carbon stock with low $\delta^{13}\text{C}$ signature into the atmosphere, as both records of atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ exhibit a relatively rapid decline during the early deglaciation. Such a carbon release could in principle come from terrestrial or oceanic sources. Zeng (2003) proposes that there is a 547 GtC carbon release from glacial to interglacial originating from organic carbon under the great ice sheets of the Northern Hemisphere. However, the observed atmospheric CO₂ variations are not in phase with reconstructed sea level fluctuations. Recorded atmospheric CO₂ concentrations show instead an extraordinary correlation with Southern Ocean (SO) climate changes. It thus appears more plausible that the dominant internal forcing feedback driving the deglacial CO₂ rise is located in the Southern Ocean (e.g., Fischer et al., 2010).

There is evidence that low glacial CO₂ concentrations can be at least partly attributed to an increased isolation of deep water masses from the atmosphere (e.g., Skinner et al., 2010; Boyle, 1988). Many authors postulate that during glacial times, there existed a relatively large and poorly ventilated abyssal water mass which led to a relocation of carbon from the atmosphere to the deep ocean. As most deep ocean water-masses are ventilated through the SO surface (e.g., Primeau, 2005), physical processes in this region are put forth as the responsible drivers of the postulated changes in deep ocean overturning (e.g., Fischer et al., 2010; Sigman and Boyle, 2000). It has been proposed that variations in deep ocean ventilation have resulted from (i) changes in SO wind forcing (Toggweiler et al., 2006), (ii) changes in SO buoyancy forcing (Watson and Naveira-Garabato, 2006) or (iii) changes in SO water column stratification (Schmittner and Galbraith, 2008; Francois et al., 1997). However, there is no broadly accepted consensus so far on the individual role of the different physical mechanisms nor on the magnitude of the associated CO₂ variations during glacial-interglacial transitions.

Reconstructions of the glacial $\delta^{13}\text{C}$ distribution from benthic foraminifera in the Pacific Ocean show that – relative to the Holocene – deep water below 2000–2500 m was more depleted in $\delta^{13}\text{C}$ compared to the upper water mass (Matsumoto et al., 2002; Keigwin, 1998). In the Indian, Atlantic and Southern Oceans, reconstructed glacial-age vertical $\delta^{13}\text{C}$ gradients point at the existence of a sharp chemocline at similar depths separating the well-ventilated water above from poorly ventilated water below (e.g., Hodell et al., 2003; McCorkle et al., 1998; Curry et al., 1988; Duplessy et al., 1988). These $\delta^{13}\text{C}$ -data support the “nutrient deepening” hypothesis (Boyle, 1988) calling for a transfer of nutrients and DIC from upper- and mid-depth waters to the deep ocean driven by a more efficient marine biological pump during glacial times. Several potential triggers for an increased biological pump efficiency during glacials have been suggested in the literature. For instance, reduced deep-water ventilation rates (e.g., Hodell et al., 2003), slowed organic matter degradation under colder water temperatures (Matsumoto et al., 2007; Kwon et al., 2009) or significantly increased aeolian iron supply to the ocean surface (Martin, 1990).

Jaccard et al. (2009) and Bradtmiller et al. (2010) have recently presented evidence for increased sequestration of remineralized carbon into the deep Pacific during the LGM. Their reasoning is based on the argument that a more efficient biological pump during glacial periods does not necessarily imply that total (= preformed + regenerated) nutrient concentrations in the deep ocean were significantly higher than during interglacials. Rather, the increase in biological pump efficiency would be mirrored in lower oxygen concentrations in the deep ocean together with an increase in regenerated nutrients which might be at least partially offset by a decrease in the preformed nutrient component (e.g., Ito and Follows, 2005).

Further evidence for a slowed deep ocean ventilation during glacial periods comes from recorded variations of opal burial in SO sediments (Anderson et al., 2009) and the occurrence of $\delta^{13}\text{C}$ carbon isotope minima in planktic foraminifera at the beginning of glacial terminations (Spero and Lea, 2002). These signals have been interpreted to reflect an invigoration in SO ventilation during the last glacial-interglacial transition. As a consequence, the breakdown of surface water stratification and renewed Circumpolar Deep Water (CDW) upwelling in the SO would bring the postulated abyssal watermass – which was isolated from the atmosphere during the glacial period – to the ocean surface.

The large recorded drop in atmospheric $\Delta^{14}\text{C}$ of about 300 ‰ over the course of the last deglaciation may further support this interpretation as the drop cannot be attributed solely to a decrease in stratospheric ¹⁴C-production (Hughen et al., 2004; Muscheler et al., 2004). Radiocarbon measurements on glacial-age sediments from the North Pacific (Galbraith et al., 2007; Marchitto et al., 2007) and from the deep Atlantic (Keigwin and Schlegel, 2002; Boyle, 1988) indicate that abyssal waters were indeed less ventilated during

glacial times. However, surface-to-deep ¹⁴C-age differences reconstructed from a sediment core in the western equatorial Pacific show that late-glacial ventilation at a depth of 2.8 km was not significantly lower than today (Broecker et al., 2007). Summarizing the available evidence from carbon isotopes, Broecker and Barker (2007) come to the conclusion that the volume of an isolated abyssal reservoir during glacials is limited to no more than half of the ocean.

The “shelf hypothesis” (Broecker, 1982) is an alternative oceanic mechanism that has been put forth to explain the recorded glacial-interglacial CO₂ fluctuations. This mechanism relies on changes in the biological pump turnover driven by shifts in organic matter accumulation on shallow shelves as sea level rises and falls. Atmospheric *p*CO₂ would be reduced during glacial times through increased biological productivity promoted by an enhanced whole-ocean nutrient inventory. Paleo-data have revealed, however, that several prerequisites are not met which are required for the shelf hypothesis to explain the full glacial CO₂ drawdown: For instance, there is no sufficiently large amount of organic sediments on shelf areas to provide additional nutrients during glacials (Peacock et al., 2006). Recent studies have suggested that an increase in the glacial nutrient inventory might be linked to changes in ventilation at intermediate depths rather than to changes in sea-level. A decreased volumetric extent of oxygen minimum zones would have reduced both the marine sinks for NO₃⁻ (Deutsch et al., 2004) and for PO₄³⁻ (Wallmann, 2010).

Besides reorganizations of the ocean-internal carbon cycle and the terrestrial biosphere, transient imbalances between marine sediment formation and continental weathering fluxes may also have contributed significantly to the recorded millennial-scale variations in atmospheric CO₂, δ¹³C_{CO₂} and Δ¹⁴C_{CO₂} (e.g., Archer et al., 2000a). Most prominently, the mechanism of carbonate compensation may have accounted for up to a third of the glacial-interglacial CO₂ change without conflicting with reconstructed seafloor CaCO₃ preservation patterns (Marchitto et al., 2005). Imbalances in the marine organic matter budget have received less attention in previous studies. Nevertheless, there is a priori no reason why excess burial or excess weathering of organic matter should not also have affected atmospheric CO₂ levels in the past.

Numerical models allow a process-based interpretation of the recorded CO₂ variations and to quantitatively test different scenarios of glacial-interglacial CO₂ variations. A dilemma is that the application of comprehensive climate-carbon models for multi-millennial simulations and for exploring the sensitivity of results to uncertain parameters is restricted by their large CPU-demand. On the other hand, box models are inherently limited as they do not include ocean dynamics and only poorly represent the 3-D geometry of the ocean and ocean-sediment interactions.

The goal of this study is to explore the carbon cycle implications of altered deep ocean ventilation and biological carbon pumps together with the millennial-scale effects of implied changes in marine sediment formation. For this purpose, we use the global coarse-resolution 3-D Bern3D+C ocean model (Müller et al., 2006; Tschumi et al., 2008), which is coupled to a spatially-resolved dynamic model for marine sediment formation (Tschumi, 2009). This modelling framework allows for the simulation of multiple tracers over long time periods of several 10 000 years which are required for sediment formation processes to relax to equilibrium.

The model simulates the cycling of total carbon (¹²C + ¹³C + ¹⁴C) and the carbon isotopes ¹³C and ¹⁴C between the atmosphere, the terrestrial biosphere, the ocean reservoir, and the marine sediments. The ocean circulation field, the ocean carbon pumps and the formation of organic, opal, and CaCO₃ sediments at the seafloor are calculated from prognostic dynamical equations discretized on 3-D grids. The terrestrial reservoir and the atmosphere are represented rather simplistically as single well-mixed boxes. The carbon stock on land is fixed (2220 GtC), but not its isotopic composition.

As a starting point of this study, we have calculated the response of the coupled model to a change in the export ratio of organic matter to calcite (Sigman et al., 1998). This experiment served as a test for the model representation of carbonate compensation and allow us to compare our modelling framework to others. The second model experiment was an increase of the whole-ocean nutrient inventory by 30 % (Sigman et al., 1998; Broecker and Peng, 1987) serving as a test for the “shelf hypothesis”. Note that a 30 %-increase in whole-ocean nutrients during glacials is likely to be an upper limit of a realistic change (Deutsch et al., 2004; Wallmann, 2010). In contrast to previous similar experiments, in our model stimulated export production of organic matter will eventually relaxed back to initial rates as excess organic matter sedimentation removes the injected nutrients from the ocean in the long term. Therefore, the modeled shift in the Pacific CaCO₃ saturation horizon in our experiment turned out to be small as opposed to previous studies (Heinze et al., 1991; Sigman et al., 1998). Finally, but as the main interest of this study, we have probed the response to variations in ocean deepwater ages driven by changes in SO wind forcing and the implied changes in SO deep convection for a range of different ocean ventilation time-scales. All of these sensitivity experiments started from the same model steady-state that is compatible with preindustrial boundary conditions for ocean physics and the biogeochemical cycles as described in Sect. 1.2.

The novel aspect of our analysis is the simulation of the time-dependent response to prototype scenarios considering ¹³C- and ¹⁴C-cycling explicitly in a fully-coupled dynamical ocean-sediment model with sedimentation of all particulate species, i.e. the marine organic matter, CaCO₃ and opal cycles are all represented as “open cycles” in this study. The

broad range of processes and variables simulated increases the complexity of the ocean carbon cycle in the model, but it also allows for a more robust simultaneous comparison of the model results with multiple proxies. The systematic variations in deep ocean water ages through changes in the wind forcing and deep convection in the Southern Ocean represents a new approach to explore the sensitivity of atmospheric CO₂ to deep ocean ventilation and marine sediment formation in 3-D models.

1.1 Model description

The version of the Bern3D+C carbon cycle model used here comprises an atmospheric, a terrestrial, an oceanic and a sedimentary reservoir which are coupled through the exchange of carbon and its isotopes ¹³C and ¹⁴C. The model's ocean physics is based on the work of Edwards et al. (1998) as updated by Edwards and Marsh (2005) and Müller et al. (2006). The marine biogeochemical cycles are described by Parekh et al. (2008) and Tschumi et al. (2008) and the coupling to the sedimentary component in Tschumi (2009). Unless explicitly stated otherwise, the model is run under standard parameters as mentioned in these references.

Marine biogeochemical cycles and air-sea gas-exchange are implemented following OCMIP-2 (Orr et al., 1999; Najjar et al., 1999) with the addition of prognostic formulations for marine biological productivity as well as representations for the cycling of iron (Parekh et al., 2008), silica (Tschumi et al., 2008), ¹³C and ¹⁴C.

Export production is a function of temperature, light, phosphate and iron following Doney et al. (1996). The competition between opal and calcite producers is modeled following formulations of the HAMOCC5 model of Maier-Reimer et al. (2005). The half-saturation constant for silica-uptake is a function of local silica concentration and the ratio of silica to phosphorous uptake by opal producers varies with silica, light, iron and phosphate following Aumont and Bopp (2006).

The atmosphere is considered as a single well-mixed box. To calculate air-sea gas-exchange, a constant value for the square of the wind-speed \bar{u}^2 is employed which is set equal to the spatio-temporal mean of the climatology for u^2 provided by OCMIP-2 (Najjar et al., 1999).

The land biosphere model taken from Siegenthaler and Oeschger (1987) is composed of 4 well-mixed compartments with fixed carbon exchange fluxes and stocks: ground vegetation and leaves (100 GtC), wood (500 GtC), detritus (120 GtC) and soils (1500 GtC).

The sedimentary component is based on the model described by Gehlen et al. (2006) and Heinze et al. (1999) which represents sediment diagenesis in the top 10 cm beneath seafloor. Sediment formation, i.e. the accumulation of opal, CaCO₃, organic matter and clay sediments is calculated on the basis of a set of dynamical equations for the sedimentary diagenetic processes. The redissolution and

remineralization reactions all follow first order kinetics. Two types of remineralization reactions are considered, oxic respiration and denitrification. For more details on the sediment model component and model equations, please refer to the appendix.

In the model, cycling of ¹³C is subject to the following fractionation effects: in the ocean, equilibrium fractionation is considered in the chemical carbonate equilibrium system (Mook, 1988). For air-sea gas-exchange, a kinetic fractionation is computed (Siegenthaler and Muennich, 1981). Further fractionations occur during marine photosynthesis (Freeman and Hayes, 1992) and the formation of calcium carbonate (Mook, 1988). In the land biosphere, isotopic fractionation is accounted for during photosynthesis.

Cycling of ¹⁴C is simulated explicitly, i.e. the actual concentrations of ¹⁴C are transported and not the ¹⁴C/¹²C-ratio (Orr et al., 1999).

1.2 Model spin-up and preindustrial steady state

The coupled carbon cycle model was spun up by bringing the terrestrial biosphere, the ocean and the sediments into steady state under fixed atmospheric values for *p*CO₂ (278.00 ppm), $\delta^{13}\text{C}$ (−6.305 ‰) and $\Delta^{14}\text{C}$ (0.00 ‰). A fixed production rate for ¹⁴C was then prescribed and *p*CO₂, $\delta^{13}\text{C}$ and $\Delta^{14}\text{C}$ were let evolve dynamically in the atmosphere to perform the subsequent sensitivity simulations. The remaining drifts in the control simulation was less than 3×10^{-3} ppm kyr^{−1} in *p*CO₂, less than 2×10^{-5} ‰ kyr^{−1} in $\delta^{13}\text{C}$ and less than 3×10^{-3} ‰ kyr^{−1} in $\Delta^{14}\text{C}$. As the typical model response to the perturbations applied was much larger than that, the spun up model state is sufficiently close to equilibrium for our purposes.

The combined ocean/sediment compartment was separately initialized as follows before it was coupled interactively to the atmosphere. To spin up ocean physics in a first step, sea surface temperatures (SST) and salinities (SSS) were restored towards modern monthly observations (Levitus et al., 1994; Levitus and Boyer, 1994) and wind stress was prescribed according to the monthly climatology from NCEP/NCAR (Kalnay and et al., 1996). After 10 000 model years, mixed boundary conditions were imposed for SST and SSS following Tschumi et al. (2008). Inspection of the modeled ocean circulation field and the temperature and salinity distributions at steady state revealed that the primary flow paths of the global ocean circulation and the important water masses were represented adequately (see Müller et al., 2006; Tschumi et al., 2008). The annual mean maximum Atlantic meridional overturning circulation (AMOC) amounted to 14.7 Sv, comparing favorably to the estimate of 17 ± 5 Sv (Talley et al., 2003).

In a second step, the steady state ocean circulation was then used to spin up marine biogeochemical cycles during 10 000 model years. The resulting steady state fields of the modeled biogeochemical tracers successfully reproduced the

large-scale patterns of the observed distributions in the modern ocean (see Tschumi et al., 2008; Müller et al., 2006, and Fig. 3). Simulated export fluxes for particulate organic matter (12.76 GtC yr⁻¹), CaCO₃ (1.13 GtC yr⁻¹) and opal (95 Tmol Si yr⁻¹) all lay within available estimates (Jin et al., 2006; Sarmiento and Gruber, 2006).

In a next phase, the sediments were “filled up” during 50 000 years by imposing spatially uniform, but temporally variable weathering fluxes to the sea surface for phosphate, silicic acid, alkalinity, carbon, ¹³C and ¹⁴C. The weathering rates were set equal to the net flux of the respective tracers across the ocean-sediment interface as diagnosed at each time-step such that annual mean ocean inventories are kept practically unchanged. As the sediment inventories converged to their steady state values, weathering rates stabilized to balance with the loss of solid matter that results from sediment burial. These values were diagnosed and prescribed subsequently as temporally constant weathering input. The ocean-sediment component was run into a new equilibrium during 30 000 years in this new setup.

Simulated sediment accumulation amounted to 0.14 GtC yr⁻¹ for organic matter and to 0.17 GtC yr⁻¹ for CaCO₃. These rates are roughly compatible with data-based estimates of 0.12–0.26 GtC yr⁻¹ for organic matter burial and of 0.1–0.14 GtC yr⁻¹ for CaCO₃ burial (Sarmiento and Gruber, 2006; Feely et al., 2004). In a series of carbon pulse release experiments, CO₂ neutralization time scales were found to be between 5–8 kyr for seafloor CaCO₃ neutralization and between 8–13 kyr for terrestrial CaCO₃ neutralization (Tschumi, 2009). These values are compatible with previous studies (e.g., Archer et al., 1997). The model equations, the model parameters and the details of the ocean-sediment coupling can be found in Tschumi (2009).

Simulated sediment composition in the spun up coupled steady state is shown in Fig. 1a–c. Table 1 displays bulk numbers of the simulated particulate fluxes exported from the sea surface, deposited at the sea floor and accumulated in the sediments along with available estimates for the preindustrial time

2 Model experiments and results

2.1 Rain ratio reduction

The marine biological pumps (Volk and Hoffert, 1985), i.e. the organic matter and the carbonate pumps, are central drivers of the ocean carbon cycle. For two separate reasons, they are of great importance for the regulation of the atmospheric CO₂ level by the oceans. First, they establish systematic gradients in dissolved inorganic carbon (DIC) and alkalinity between the surface and the deep ocean. The overall impact of both pumps together is lower *p*CO₂ levels in the surface ocean compared to an abiotic ocean.

Second, accumulation of organic matter and CaCO₃ sediments is driven by the deposition of biogenic material on the seafloor. Any change in the marine biological pumps thus also affects organic matter and CaCO₃ sedimentation with potentially significant impacts on atmospheric CO₂ through perturbations of the marine organic matter and CaCO₃ balances.

The ocean carbonate pump is driven by the production of biogenic CaCO₃ shells in the sea surface. The sinking CaCO₃ shells entrain alkalinity to the deep ocean and hence increase sea surface *p*CO₂ by shifting the speciation of the surface ocean DIC towards dissolved CO₂. A reduced export ratio between CaCO₃ and organic matter export (rain ratio) would reduce sea surface *p*CO₂, not only due to an ocean-internal redistribution of alkalinity but also as a result of a lower carbonate concentration in the deep ocean triggering redissolution of marine CaCO₃ sediments. The resulting imbalance between the weathering flux of dissolved CaCO₃ into the ocean and marine CaCO₃ sedimentation would drive a rise in the ocean's total alkalinity and additionally contribute to lower sea surface *p*CO₂. Carbonate compensation is the feedback mechanism responsible for re-establishing the ocean CaCO₃ balance on a time-scale of order 10 kyr through shifts in the depth of the CaCO₃ saturation horizon (Archer et al., 2000a).

By reason of their central role in the regulation of the ocean's alkalinity budget, rain ratio mechanisms were among the first candidates made responsible for driving glacial-interglacial CO₂ fluctuations in the atmosphere (e.g., Archer et al., 2000a; Berger and Keir, 1984). A reduced rain ratio during glacial times may have resulted, for instance, from increased diatom productivity in lower latitudes at the expense of coccolithophorids (silica leakage hypothesis, Matsumoto et al., 2002) or from relaxed nutrient-limitation in the sea surface (Margalef, 1978). The inferred CaCO₃ preservation history recorded in marine sediments, however, constrains the contribution of whole-ocean alkalinity changes to roughly a third of glacial-interglacial CO₂ variations (Marchitto et al., 2005).

In our first model experiment, we attempt to disentangle and quantify the complex response in ocean biogeochemistry to a reduced rain ratio. We perform parallel simulations in a coupled ocean-sediment setup and in an ocean-only setup to separate ocean-internal mechanisms from interactions with the sediments. Following Sigman et al. (1998), we abruptly reduce CaCO₃ export by 20 % uniformly in the global sea surface. This corresponds in our model to a decline in the global rain ratio from 0.088 to 0.071.

2.1.1 Results

Figure 3 shows the time-dependent carbon cycle responses as simulated with the ocean-only model (black lines) and the ocean-sediment model (red lines). In response to decreased CaCO₃ export atmospheric CO₂ drops from 278 ppm to

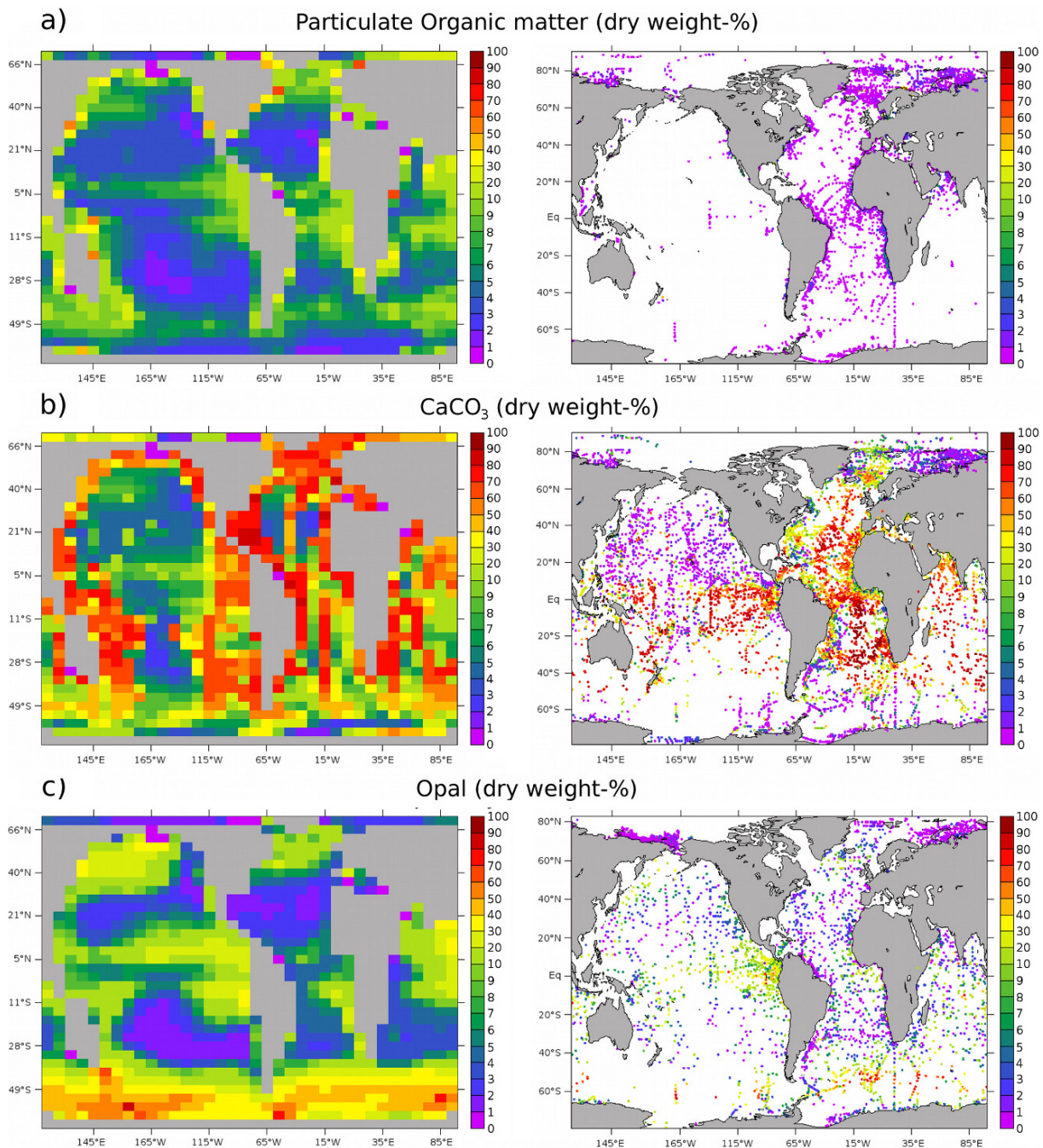


Fig. 1. Modelled (left row) and observed (right row) composition of surface sediments on the ocean bottom: **(a)** particulate organic matter, **(b)** CaCO₃ and **(c)** opal. Modelled sediment composition is averaged over the top 1.1 cm of the sediment column. Sediment data is from the Pangaea database (publishing Network for Geoscientific & Environmental Data, <http://orfois.pangaea.de>). Note that the color scale is not linear.

268 ppm within centuries (e-folding time scale: 390 yr) in the ocean-only model (Fig. 3a, black line). Mean surface alkalinity rises as less CaCO₃ shells sink from the surface to the ocean interior (Fig. 3b, black line). Reduced alkalinity in the ocean interior in turn lowers [CO₃²⁻] pushing the CaCO₃ saturation horizon upwards by approximately 300 m (Fig. 3d, black line).

In the ocean-sediment model, the CO₂ drawdown is strongly amplified to 237 ppm due to the increase in whole-ocean alkalinity (Fig. 3a and b, red line). Carbonate compensation responds rather slowly with an e-folding time scale of 11.7 kyr. Reduced CaCO₃ export and deposition on the seafloor drive a deepening in the CaCO₃ saturation horizon as to provide a larger area on the seafloor where the rain of

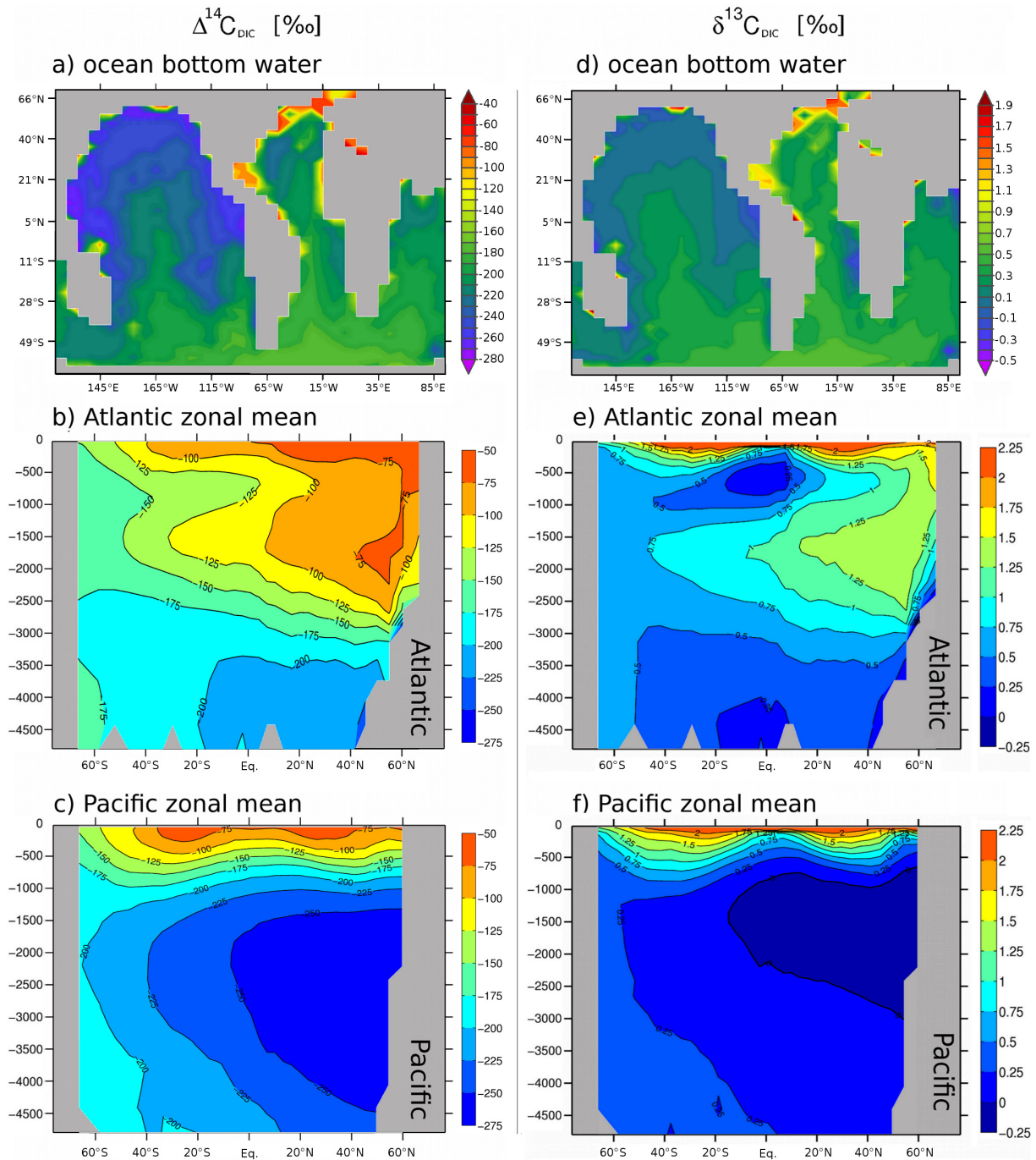


Fig. 2. Modelled distribution of the isotope signatures in ocean dissolved inorganic carbon for preindustrial boundary conditions. $\Delta^{14}\text{C}_{\text{DIC}}$ is shown in the left row and $\delta^{13}\text{C}_{\text{DIC}}$ in the right row. Top panels (a) and (d) show bottom water signatures, the middle row panels (b) and (e) display Atlantic and bottom panels (c) and (f) Pacific zonal mean sections in preindustrial steady state.

CaCO_3 is efficiently preserved (Fig. 3d, red line). As a consequence, the simulated CaCO_3 sedimentation rate starts to recover after the initial abrupt decline in order to restore the ocean's CaCO_3 balance in the model (Fig. 3c). The final total shift in steady state saturation depth in the ocean-sediment model amounts to a deepening of 700 m in the Atlantic and of 800 m in the Pacific basin (Fig. 3d, red line).

The simulated responses in atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ to changes in the rain ratio are small. The ocean-internal distribution of $\delta^{13}\text{C}_{\text{DIC}}$ and $\Delta^{14}\text{C}_{\text{DIC}}$ is only marginally affected by the marine CaCO_3 cycle. In the ocean-only experiment, we therefore simulate negligible changes in atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ as CaCO_3 export is reduced. In fact, there is virtually no response in

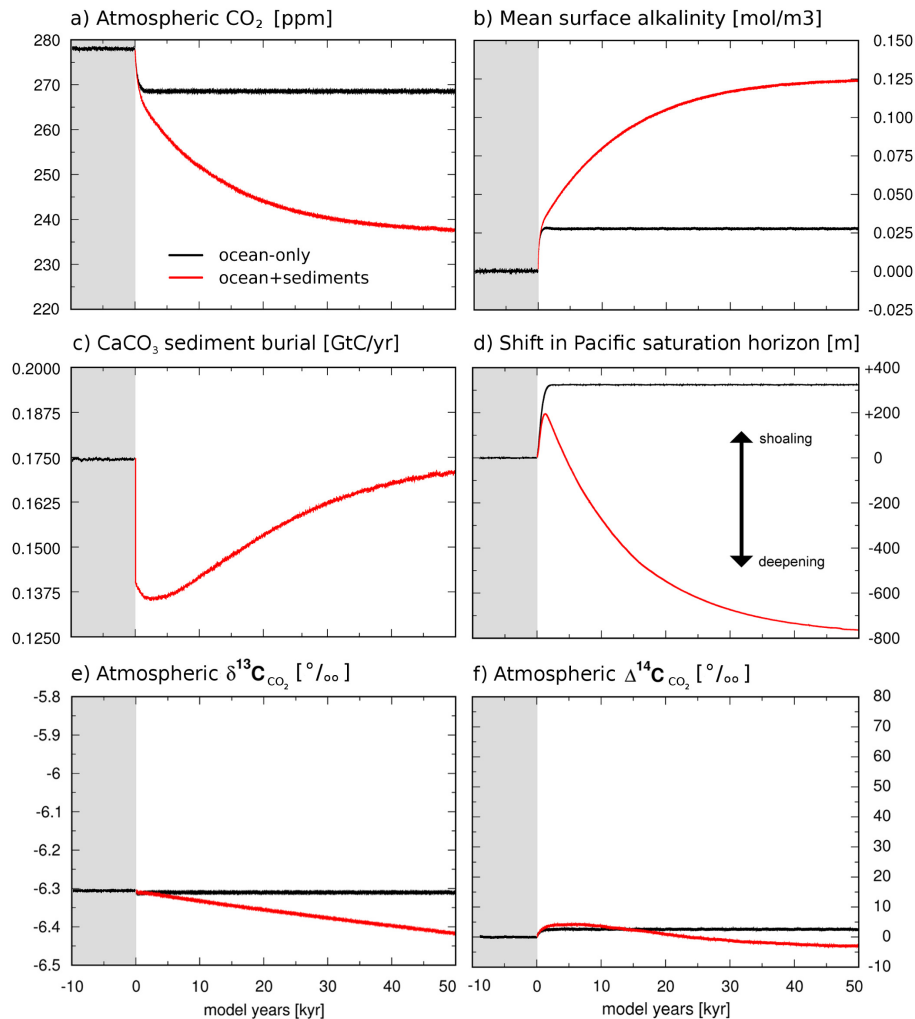


Fig. 3. Rain ratio reduction: Simulated response to a 20%-reduction in CaCO₃ export. **(a)** Atmospheric CO₂, **(b)** mean surface alkalinity, **(c)** CaCO₃ sedimentation, **(d)** shift in Pacific CaCO₃ saturation horizon, **(e)** atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ and **(f)** atmospheric $\Delta^{14}\text{C}_{\text{CO}_2}$.

atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ (Fig. 3e, black line) and $\Delta^{14}\text{C}_{\text{CO}_2}$ is raised by a modest 3 ‰ (Fig. 3f, black line).

In the coupled ocean-sediment experiment, the system response in carbon isotope signatures of atmospheric CO₂ is additionally sensitive to the decline in marine CaCO₃ sedimentation. However, the responses of $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ to carbonate compensation are rather slow and remain small (Fig. 3e and f, red lines).

2.1.2 Comparison to other studies

Heinze et al. (1991) have simulated a relatively modest sensitivity of atmospheric CO₂ to changes in the rain ratio. They have found a CO₂ drawdown of 28.5 ppm when the rain ratio is reduced by 50 % in the 3-D Hamburg Carbon Cycle Model which was coupled to a simplistic model for sediment diagenesis with fixed sediment resuspension rates. Archer and Maier-Reimer (1994) have shown that including organic-carbon-driven dissolution of CaCO₃ sediments substantially

increases the $p\text{CO}_2$ response to rain ratio changes in their model. They reach glacial CO₂ levels by reducing the rain ratio by 40 %, which amounts to a CO₂ sensitivity comparable to this study.

Sigman and Boyle (2000) have reduced the ratio of organic matter to CaCO₃ export in the low latitudes by 50 % in the CYCLOPS box model. They obtain a drawdown of 22 ppm in the ocean-only model and of 63 ppm in the ocean-sediment model. CaCO₃ compensation amplifies the CO₂ reduction by a factor of three in their study compared to a factor of four in this study. The reason for the comparatively lower impact of CaCO₃ compensation in the box model study is probably the shorter relaxation time of CaCO₃ burial. The original rate is restored after about 15 kyr, whereas in the Bern3D+C model it takes roughly 50 kyr. The longer relaxation time in our study results in a larger imbalance between CaCO₃ burial and weathering when integrated over time such that the increase in whole-ocean alkalinity is larger in response to carbonate compensation.

Chikamoto et al. (2009) have lowered the rain ratio by 25 % in the High-Latitude Exchange/Interior Diffusion Advection (HILDA) model coupled to a similar dynamical sediment model as used in this study. They find a drawdown of 59 ppm in atmospheric CO₂ together with a deepening of the calcite saturation horizon by 1870 m. The response time scale of atmospheric CO₂ and the magnitude of the drawdown are comparable to the results of our study; however, we find a deepening of the order of only 800 m in the CaCO₃ saturation horizon.

In conclusion, it appears to be challenging to robustly quantify the effect of rain ratio changes on atmospheric CO₂ by means of carbon cycle models. The simulated CO₂ response depends on the specific biogeochemical model state, such as the initial rain ratio or the CaCO₃ throughput in the oceans, as well as on the model parametrization of the ocean-sediment interactions. Our results are consistent with current understanding, suggesting that carbonate compensation would amplify the CO₂ response to changes in the ocean carbonate pump by a factor of as much as 3–4.

2.2 Strengthening of the marine biological cycle

As an alternative to a reduced rain ratio, low atmospheric CO₂ during glacial periods might have resulted from more vigorous biological carbon cycling in the ocean. A strengthened marine biological cycle might be driven for instance by increased nutrient availability (Broecker and Peng, 1982), iron fertilization (Martin and Fitzwater, 1988) or changes in nitrogen cycling (Falkowski, 1997).

In the next experiment, we examine the response in atmospheric CO₂, $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ to a stimulated ocean biology by increasing the marine nutrient inventory in the Bern3D+C model. The experiment is again performed with and without coupling to sediments to isolate the effect of ocean-sediment interactions.

Following the procedure of Archer et al. (2000a); Sigman et al. (1998) and Broecker and Peng (1987), we have abruptly increased PO₄ concentrations uniformly by 30 % to stimulate simultaneously both the organic matter and the carbonate pumps. With our model biology the global rain ratio remained virtually unchanged. Therefore, the sedimentary response on atmospheric CO₂ is dominated in our model by excess organic matter sedimentation rather than by carbonate compensation.

2.2.1 Atmospheric CO₂

In the ocean-only model, the augmented PO₄ inventory drives a persistent increase in organic matter and CaCO₃ export production both by roughly 11 % (Fig. 4c and d, black lines). The elevated rates of organic matter and CaCO₃ export have opposite effects on sea surface *p*CO₂. As the strengthened organic matter pump dominates the net ocean response, atmospheric CO₂ gets reduced to 255 ppm with an

e-folding time scale of 470 years in the ocean-only experiment (Fig. 4a, black line). As the concurrent changes in downward fluxes of POC and CaCO₃ have roughly opposing effects on [CO₃²⁻] at the depth of the saturation horizon, there is only a small shift in calcite saturation depth of less than 20 m in the ocean-only model (Fig. 4b, black line).

Biological export of carbon from the sea surface augments primarily in the Atlantic and in the Indian oceans, and to a lesser degree in the Northern Pacific. There is almost no change in export production in the equatorial and South Pacific as well as in the Southern Ocean (south of 51° S) because biological productivity in these regions is limited by iron or light availability. Note that a glacial increase in export production in the North Pacific is at odds with paleoproductivity reconstructions (see Jaccard et al., 2009; Galbraith et al., 2007; Crusius et al., 2004), suggesting that the history of export production from the North Pacific surface was predominantly controlled by processes other than changes in the whole-ocean nutrient inventory such as changes in water column stratification and associated upwelling of nutrients from the deep (Galbraith et al., 2007).

In the coupled ocean-sediment setup, atmospheric CO₂ is reduced to 232 ppm within 30 kyr (Fig. 4a, red line). Thereafter, CO₂ slightly increases by 2 ppm until the full balance between weathering and burial of organic matter and CaCO₃ is re-established after 50 ky (Fig. 4e and f). The effect of ocean-sediment interactions thus enhances the CO₂ drawdown in the coupled system by a factor of two.

The nutrient increase abruptly boosts the export of organic matter and CaCO₃ by 8.5 % in the ocean-sediment model (Fig. 4c and d, red lines). Subsequently, both export rates decline and reach almost original values after 50 kyr. As both changes in CaCO₃ and organic matter burial are taken into consideration, elevated organic matter export is not sustained in the long run here, in contrast to previous studies (Sigman et al., 1998; Broecker and Peng, 1987) which have accounted solely for changes in CaCO₃ sedimentation.

Excess organic matter burial draws nutrients out of the ocean to bring organic sedimentation back to balance the riverine input of nutrients. The initial increase in the nutrient inventory is thereby neutralized in the long run. The simulated decline in CaCO₃ export also follows directly from the gradual decrease in nutrient availability. In our model biology, CaCO₃ export is computed from primary productivity and is not affected by changes in the calcite saturation state. The effect of CaCO₃ compensation on restoring CaCO₃ burial is therefore only of secondary importance, in contrast to the rain ratio reduction experiment.

The modeled CaCO₃ saturation horizon deepens by about 150 m in the Atlantic and by 300 m in the Pacific during the first 15 000 years with a subsequent shoaling in both basins thereafter (Fig. 4b, red line). The first part of the response is dominated by excess POC burial, the second part by excess CaCO₃ burial. Since excess POC burial implies a loss of whole-ocean DIC, it tends to push the saturation horizon

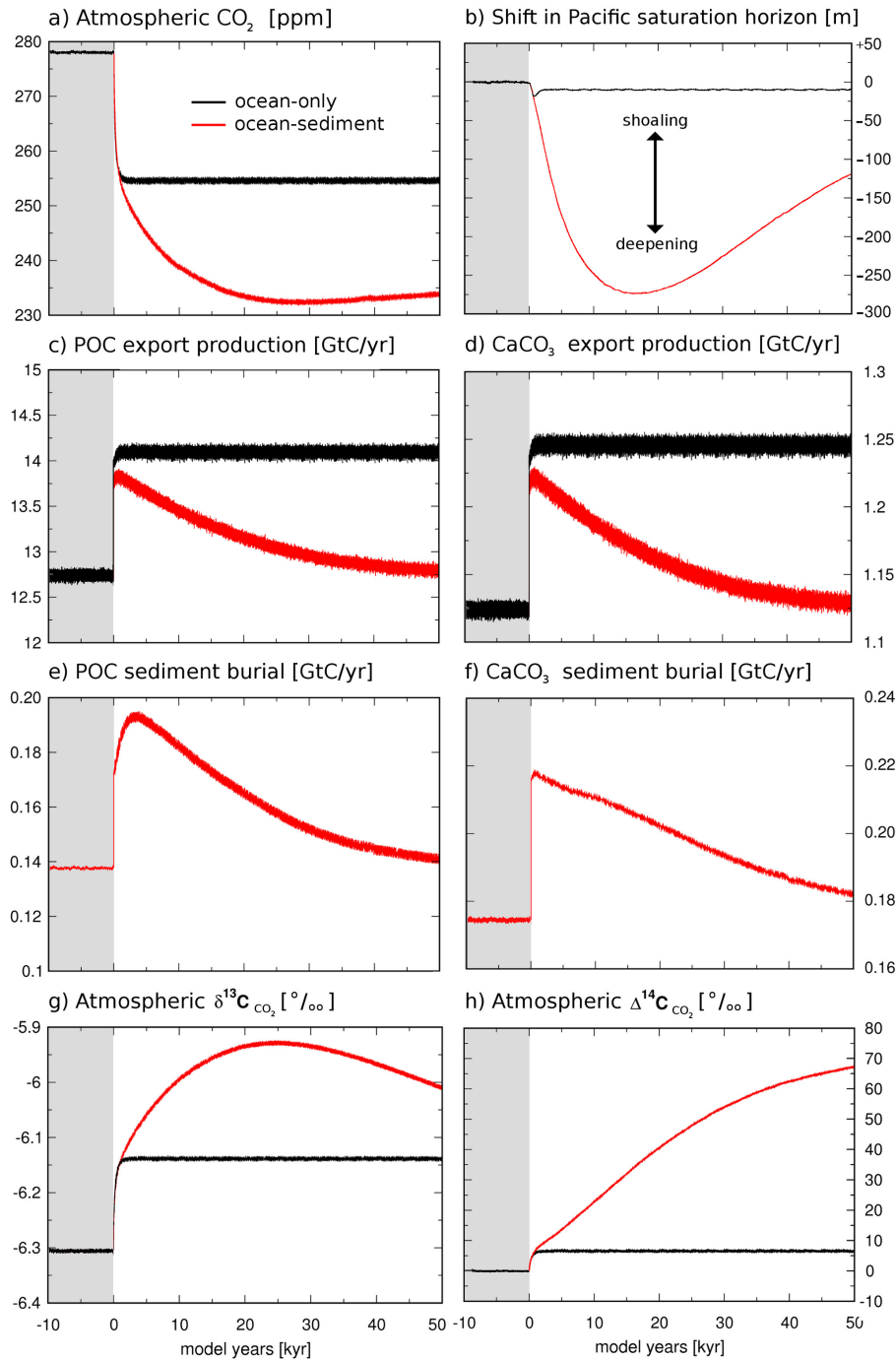


Fig. 4. Strengthening of the marine biological cycle: Simulated response to an abrupt 30%-increase in the ocean's phosphate content. (a) Atmospheric CO₂, (b) shift in Pacific CaCO₃ saturation horizon, (c) POC export production, (d) CaCO₃ export production, (e) POC sedimentation, (f) CaCO₃ sedimentation, (g) atmospheric δ¹³C_{CO₂} and (h) atmospheric Δ¹⁴C_{CO₂}.

downwards. On the contrary, excess CaCO₃ burial leads to a shoaling of the saturation horizon as alkalinity and DIC are removed from the ocean in a 2:1-ratio. The final shift in the calcite saturation horizon after 50 kyr is a deepening of less than 100 m.

2.2.2 Atmospheric δ¹³C_{CO₂} and Δ¹⁴C_{CO₂}

The simulated changes in atmospheric δ¹³C_{CO₂} and Δ¹⁴C_{CO₂} are predominantly controlled by the response in organic matter cycling. In the ocean-only model, atmospheric

$\delta^{13}\text{C}_{\text{CO}_2}$ rises by 0.16‰ and $\Delta^{14}\text{C}_{\text{CO}_2}$ by 7 ‰ in response to the strengthened marine biological cycle (Fig. 4g and h, black lines). $\delta^{13}\text{C}_{\text{CO}_2}$ rises as surface ocean $\delta^{13}\text{C}_{\text{DIC}}$ is shifted towards more positive values due to enhanced production of organic matter. The increase in $\Delta^{14}\text{C}_{\text{CO}_2}$ is the direct consequence of reduced atmospheric CO₂ under a constant ¹⁴C-production rate.

In the ocean-sediment model, $\delta^{13}\text{C}_{\text{CO}_2}$ increases by ~0.4‰ and $\Delta^{14}\text{C}_{\text{CO}_2}$ rises by roughly 70‰ in response to a strengthened marine biological cycle (red lines in Fig. 4g and h). The time-dependent response in the carbon isotope signatures of atmospheric CO₂ is more complex than in the ocean-only system. In particular, relaxation times are much longer due to the slow ocean-sediment interactions.

Next to the (transient) changes in the individual pumps' strengths, elevated sedimentation rates of organic matter and CaCO₃ additionally affect the response in isotopic carbon signatures in the ocean-sediment setup. The sedimentary response in atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ is rather involved since excess organic matter burial removes carbon with negative $\delta^{13}\text{C}$ and excess CaCO₃ burial draws carbon with positive $\delta^{13}\text{C}$ out of the system. Excess organic matter sedimentation dominates the response in atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ during the initial 25 kyr, whereas excess CaCO₃ burial drives a slow reduction in $\delta^{13}\text{C}_{\text{CO}_2}$ thereafter (red line in Fig. 4g). This time-dependent behavior results from the longer relaxation time of the marine CaCO₃ cycle to the intensified marine biology as compared to the relaxation time of the organic matter cycle.

The sedimentary response pushes atmospheric $\Delta^{14}\text{C}_{\text{CO}_2}$ signatures towards more positive values (red line in Fig. 4h). On the one hand, the ocean-sediment interactions draw CO₂ out of the atmosphere which elevates $\Delta^{14}\text{C}_{\text{CO}_2}$ under constant ¹⁴C-production. On the other hand, enhanced organic matter and CaCO₃ sedimentation further contribute to rising atmospheric $\Delta^{14}\text{C}_{\text{CO}_2}$ as excess burial removes relatively old carbon (with low $\Delta^{14}\text{C}$) from the combined ocean-atmosphere system.

2.2.3 Comparison to other studies

Heinze et al. (1991) have increased the marine PO₄ inventory by 30% in the 3-D Hamburg ocean-sediment model. They have found that atmospheric CO₂ decreased by 60 ppm and the Pacific CaCO₃ saturation horizon deepened by roughly 900 m. Sigman et al. (1998) only let low-latitude production rise upon a 30%-increase in whole-ocean PO₄ inventory in the CYCLOPS box model. They find an increase of 32% in low-latitude export production together with a CO₂ reduction of 32 ppm in the ocean-only setting. When CaCO₃ sedimentation is additionally taken into account, the simulated CO₂ decline is only of 19 ppm. As opposed to our results, the sedimentary response in their simulation is to counteract the CO₂ drawdown resulting from strengthened biological pumps. Sigman et al. (1998) find a mean global shoaling

in CaCO₃ saturation depth of 500 m, whereas here we obtain only a modest deepening.

The reason for this discrepancy is mainly the fact that POC burial is not taken into account explicitly by Sigman et al. (1998). The nutrient-feedback is thus not considered. In our simulations, the loss of nutrients through excess POC burial on the one hand draws carbon out of the ocean and on the other hand limits excess CaCO₃ export in such a way that CaCO₃ compensation has only a minor effect opposing the CO₂ reduction. The impact of CaCO₃ compensation on atmospheric CO₂ is seen in a small increase occurring after 30 kyr (Fig. 4a, red line).

The following conclusions emerge. Perturbations in global burial fluxes due to ocean processes are removed on a multi-millennial time scale to re-establish the balance between removal of organic matter, CaCO₃, and opal by burial and corresponding weathering inputs from rivers and winds. As a consequence of this sedimentary compensation, some proxy signals that arise from changes in the internal ocean carbon cycle are partly or completely muted with time (see also next Sect. 3.3). Differences in proxies between distant time periods, for instance between the LGM and the late Holocene, may therefore partly reflect changes in the balance of the weathering-sedimentation cycle.

Kohfeld et al. (2005) compared productivity records for the late Holocene, the Last Glacial Maximum and marine isotope stage 5a to 5d. They concluded that export production was globally higher during the LGM than during either stage 5a-5d or the Late Holocene. They suggest a limited role for the marine biological cycle to explain glacial-interglacial CO₂ variations. Our results indicate that such a conclusion should be viewed with caution, because differences in proxies might also be related to imbalances in the weathering-sedimentation cycle. Potential secular trends arising from these imbalances should also be considered when analyzing whole ocean changes in $\delta^{13}\text{C}$ to infer variations in organic matter pools (Shackleton, 1977; Bird et al., 1994) or in atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ (Leuenberger et al., 1992), as well as when interpreting the recorded evolution of nitrogen and silicate isotope signatures (Deutsch et al., 2004; De La Rocha, 2006).

2.3 Global ocean ventilation

The last set of simulations focusses on the response in atmospheric CO₂, $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ to changes in the rate of deep ocean ventilation. In these idealized experiments we vary deep ocean ventilation in the model by uniformly scaling the prescribed amplitude of both zonal and meridional wind stress in the Southern Ocean (south of 51° S). The time-dependent model response in atmospheric CO₂, $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ is then calculated as the result of the interplay between changes in ocean circulation, marine biogeochemical cycling and sediment diagenesis subsequent to abrupt change in wind boundary conditions. Note that gas-exchange velocities are kept unaffected and feedbacks mediated by SST

Table 1. Bulk numbers characterising the preindustrial steady state of the coupled ocean-sediment Bern3D+C model compared to previous modelling results and independent estimates.

Variable	Units	this study	Heinze et al. (2003)	Ind. estimates
atm. CO ₂	ppm	278.0	277.7	–
POC export production	GtC yr ⁻¹	12.76	8.66	6.5–13.1 ^a
POC rain onto sediments	GtC yr ⁻¹	0.49	0.12	1.7–3.3 ^a
POC accumulation	GtC yr ⁻¹	0.14	0.048	0.12–0.26 ^a
CaCO ₃ export production	GtC yr ⁻¹	1.13	1.64	0.8–1.2 ^b
CaCO ₃ rain onto sediments	GtC yr ⁻¹	0.40	0.42	0.5 ^c
CaCO ₃ accumulation	GtC yr ⁻¹	0.17	0.18	0.1–0.14 ^d
Opal export production	Tmol Si yr ⁻¹	95	185	102–178 ^b
Opal rain onto sediments	Tmol Si yr ⁻¹	64.9	81.3	13–47 ^e
Opal accumulation	Tmol Si yr ⁻¹	1.12	5.5	5.3–8.9 ^e
POC sediment pool	GtC	428	241.9	–
CaCO ₃ sediment pool	GtC	881	1010	–
Opal sediment pool	Tmol Si	9779	19 986	–

^a Sarmiento and Gruber (2006), ^b Jin et al. (2006), ^c Milliman and Droessler (1996), ^d Feely et al. (2004), ^e Treguer et al. (1995)

changes and the response in terrestrial biosphere is not taken into account. The simulation set consists of separate runs in which SO wind stress is gradually varied between 20 % and 180 % relative to the original strength by steps of 20 %.

It has been found in several global ocean models (including ours) that deep mixing in the SO is driven to a large degree by momentum transfer from SH westerly winds to the ocean surface in the latitudinal band spanned by the Drake Passage (e.g., Toggweiler et al., 2006; Menviel et al., 2008; Tschumi et al., 2008). Furthermore, the modeled characteristics of deep ocean water masses are affected on a global scale by changes in SO deep mixing. Accordingly, the SO represents a major route of deep ocean ventilation in these ocean models.

Yet, it remains hard to draw conclusions on the physical mechanisms controlling deep mixing in the real SO. In recent studies, factors other than wind stress have been proposed to play a major role for the control of SO deep mixing. These are for instance sea-ice dynamics and buoyancy forcing (e.g., Fischer et al., 2010; Watson and Naveira-Garabato, 2006), small-scale eddy flows which are not represented explicitly in most global ocean models (e.g., Spence et al., 2009; Böning et al., 2008) or other processes affecting the vertical stratification in the SO (Sigman et al., 2004; Paillard and Parrenin, 2004). The relative importance of the different mechanisms capable of driving glacial-interglacial variations in SO deep mixing remains under debate. For this reason, the amplitude of SO winds in our simulations is regarded as a mere tuning knob for the circulation state in the model.

2.3.1 Ocean circulation response

The two most prominent features in the circulation response to SO wind variations are: (1) changes in the frequency and depth of convective events in the SO and (2) changes in the intensity and depth of the Atlantic meridional overturning circulation (AMOC). Stronger winds intensify deep convection and Ekman upwelling in the SO and strengthen the AMOC (Fig. 5b and Tschumi et al., 2008). Weaker winds induce the opposite response characterized by a more stably stratified Southern Ocean and less intense AMOC.

In case of strongly reduced wind stress (20 % or 40 % of the original amplitude), the AMOC breaks down completely in the model, leaving the SO as the sole region of deep water formation. These latter circulation states are named “off-states” here, whereas the model states with active AMOC are denoted as “on-states”.

As long as the AMOC remains active, i.e. in the on-states, deep ocean ventilation scales quite homogeneously across all ocean basins with the changes in SO wind amplitude (Fig. 5a). Deep ventilation ages below 2000 m vary almost linearly with the strength of SO wind stress. Globally averaged deep ventilation ages increase by 32 years per 10 %-reduction in SO wind strength. The Atlantic deep water age is slightly more sensitive (43 years/10 %-reduction) than in the other basins, owing to the changes in AMOC strength.

In case of the off-states the link between deep ocean ventilation ages and SO wind stress deviates from the quasi-linear relationship (Fig. 5a). In particular in the deep Atlantic, the AMOC shutdown results in a large anomalous drop of roughly 600 years in ventilation age. This model response indicates the change in end-members of Atlantic deep water subsequent to the collapse of AMOC.

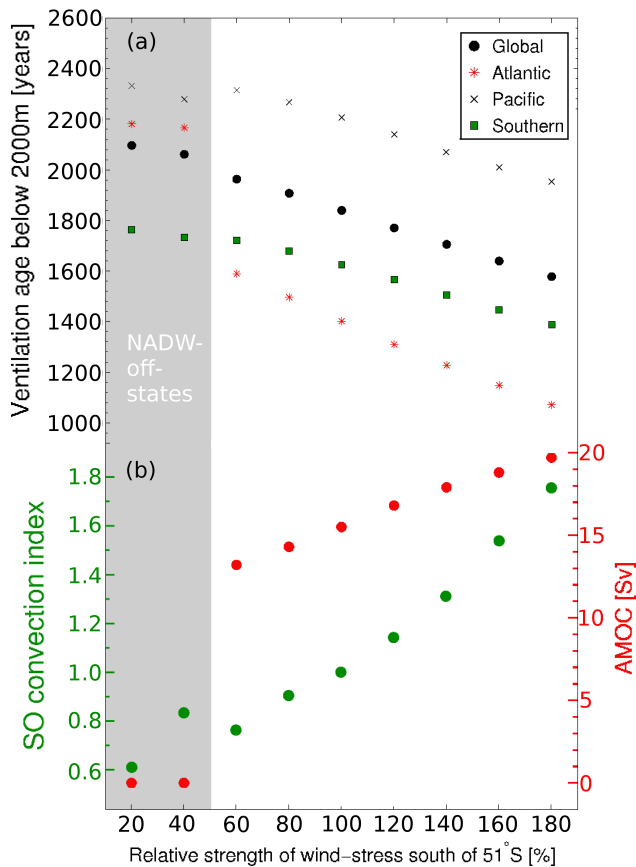


Fig. 5. (a) Ventilation age in different ocean basins as a function of the SO wind stress amplitude. Ventilation age is calculated from simulated ^{14}R -signatures in the ocean below 2000 m depth. ^{14}R is treated here as an ideal age tracer in the sense that it is transported only by ocean circulation and the atmospheric signature is held constant ($^{14}R_{\text{atm}} = 1.0$). Ventilation age is calculated as $\tau_{\text{basin}} = \frac{1}{\lambda} \ln \left(\frac{^{14}R_{\text{atm}}}{^{14}R_{\text{basin}}} \right)$ where $\lambda = 1/8267 \text{ yr}^{-1}$ is the decay constant for ^{14}C and $^{14}R_{\text{basin}}$ is the average ^{14}R -signature in the respective basin. (b) Southern Ocean convection index and Atlantic Meridional Overturning Circulation (AMOC) versus SO wind stress. The SO convection index is proportional to the number of model grid cells in the SO participating in convective mixing.

To sum up, changing wind stress in our model proves itself as an effective method to gradually vary globally averaged deep water age in a range between 1580 years (180 % SO wind stress) and 2100 years (20 % SO wind stress). Major structural shifts in the modeled global circulation pattern occur only if SO wind stress is strongly reduced to 20 % or 40 % where AMOC breaks completely.

2.3.2 Response to sub-millennial processes

Figure 6 shows the time-dependent response for a set of informative carbon cycle parameters. The reaction of atmospheric CO₂ and its isotopic composition to changes in deep ocean ventilation is characterized by a major fast response

unfolding during approximately the first 2 kyr and a minor slow response over several 10 kyr (Fig. 6a, g and h). The fast response arises from the reorganization of the ocean carbon pumps, the long-term response is driven by the slow adaptation of sediment formation to the altered biogeochemical state of the ocean. Atmospheric CO₂, $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ have reached a quasi-equilibrium after 2 kyr in the ocean-only model, whereas in the ocean-sediment model sedimentation processes affect atmospheric carbon on a time scale of several 10 kyr.

Figure 7 shows the correlation between simulated deep ocean ventilation age and the changes in atmospheric CO₂, $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ at the time slice 2 kyr after perturbing SO wind stress forcing. The results unambiguously display that better ventilated deep ocean water masses drive an increase in atmospheric CO₂, whereas $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ decline. Model results of the ocean-sediment model at 2 kyr are not substantially different from those of the ocean-only model since the effect of changes in sedimentation has not yet fully materialized at this early stage of the response.

Atmospheric CO₂ and its isotopic signatures follow an almost linear relationship with deep ocean ventilation age for both without and with sediments as long as there is no AMOC collapse. Linearized sensitivities for $\delta^{13}\text{C}_{\text{CO}_2}$ (0.08 ‰/100 years) and for $\Delta^{14}\text{C}_{\text{CO}_2}$ (10 ‰/100 years) are almost identical in both model settings (Fig. 7b and c). Changes in atmospheric CO₂ are slightly larger (−8 ppm/100 years) at the 2 kyr time slice (Fig. 7a).

AMOC shutdowns significantly affect atmospheric CO₂ and $\Delta^{14}\text{C}_{\text{CO}_2}$. The simulated changes for off-states clearly deviate from the linear relationship as displayed in Fig. 7a and c. The AMOC shutdown induces an anomalous draw-down in atmospheric CO₂ of ~10 ppm without sediments and of ~20 ppm with sediments. The anomalous rise in $\Delta^{14}\text{C}_{\text{CO}_2}$ is of ~10 ‰ in the ocean-only model and of ~15 ‰ in the ocean-sediment model. Atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ on the other hand is rather insensitive to the shutdown of AMOC. There is no substantial deviation from linearity in the relation between $\delta^{13}\text{C}_{\text{CO}_2}$ with the rate of deep ocean ventilation when AMOC shuts down (Fig. 7b).

The biological pump strongly dominates the net CO₂ response subsequent to wind stress changes in our model (see also Tschumi et al., 2008). The effect of the solubility pump is minor as simulated changes in (potential) deep ocean temperature (below 2000 m) are ~0.5 °C at most (Table 2). This translates into a maximum effect on atmospheric CO₂ of roughly 5 ppm given the sensitivity of CO₂-solubility to temperature in seawater (e.g., Sigman and Boyle, 2000). On the other hand, the imposed changes in ocean ventilation have a strong impact on the marine biological carbon pumps. Enhanced (reduced) ocean ventilation implies increased (diminished) upwelling of nutrients, thus promoting (weakening) biological export from the surface.

Table 2 indicates that organic matter export diagnosed at 2 kyr after the wind stress switch scales roughly linearly with

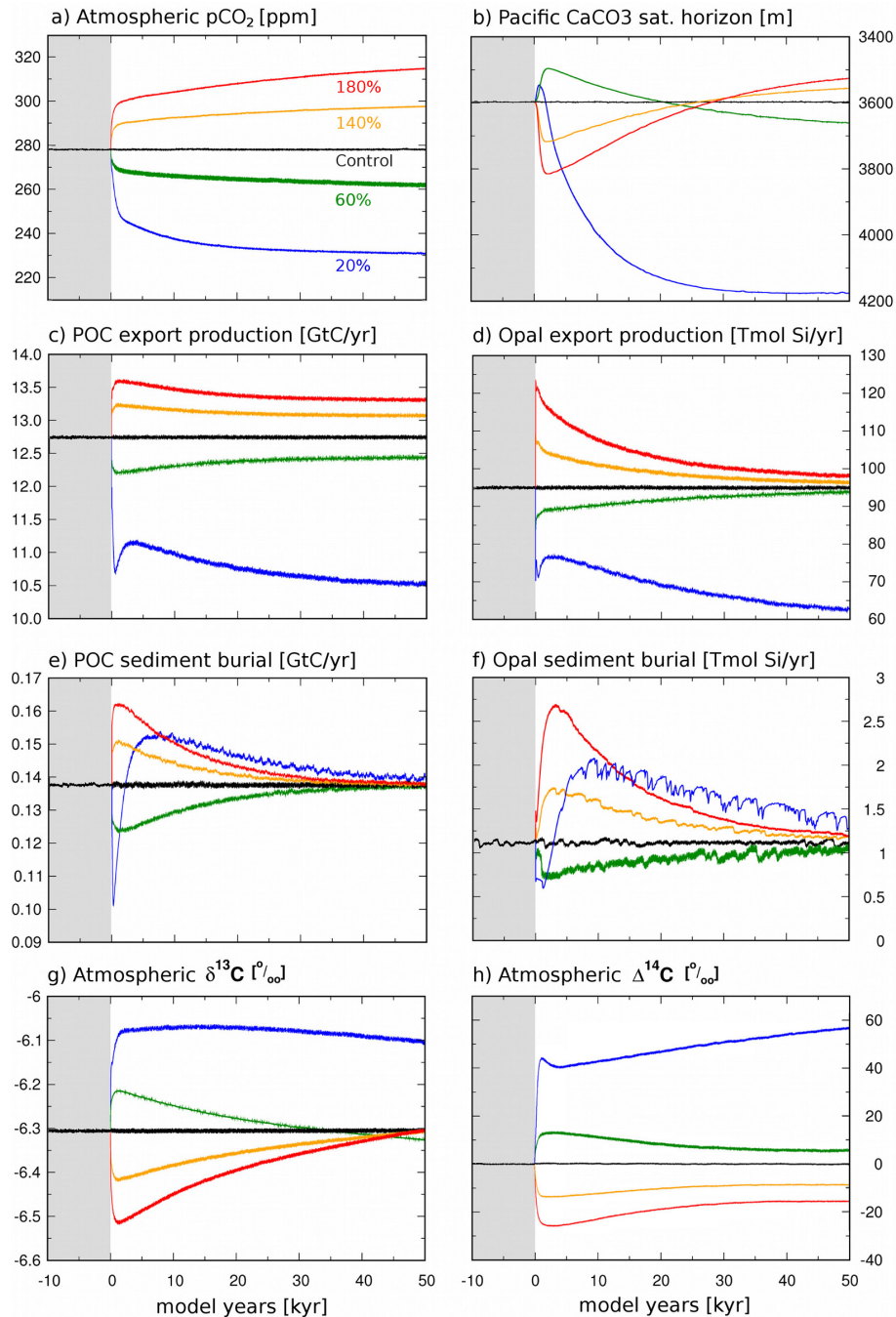


Fig. 6. Ocean ventilation experiments: time-dependent carbon cycle response of the coupled ocean-sediment system to changes in deep ocean ventilation induced by scaling the amplitude of windstress and deep convection in the Southern Ocean.

ventilation in case of the on-states (rising by 2.9 % per 100-year reduction in global deep water age). The marine silica cycle reacts substantially more sensitively to the ocean ventilation changes than the organic matter cycle: Opal export at 2 kyr increases by 8.6 % upon a 100 year-reduction in global deep water age and decreases by 7.6 % per 100 year-increase. The full time-dependent response in POC and opal export is shown in Fig. 6c and d.

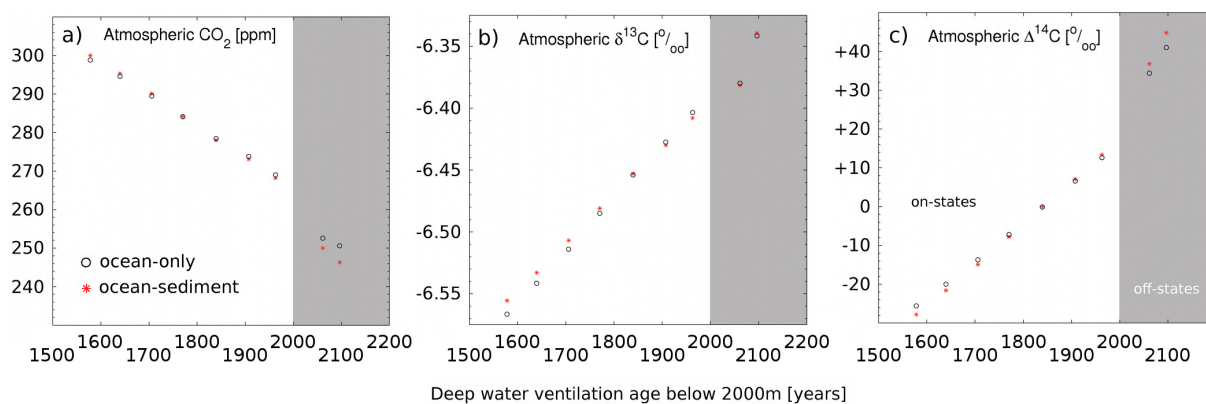
Note that the efficiency of the soft-tissue pump P^* , defined here as the ratio between regenerated and total phosphate in the deep ocean (Ito and Follows, 2005), behaves conversely, i.e. P^* increases as export production decreases parallel to slowed ventilation (Table 3). As long as NADW formation is active, a linear relation is found between atmospheric CO₂ and P^* (Fig. 8), which reflects a maximum impact of the biological pumps of 336 ppm at full efficiency.

Table 2. Ocean ventilation experiments in the ocean-sediment model: simulated changes in the global mean deep ocean temperature, biological production and in the efficiency of the soft-tissue pump at the time slice 2 kyr after the abrupt variation in ocean ventilation.

SO wind stress amplitude	Atm. $p\text{CO}_2$	Mean deep water age ^a	Mean deep water temp. ^a	POM export	CaCO ₃ export	Global mean rain ratio	Opal export	Efficiency of soft-tissue pump ^b
	ppm	years	°C	GtC yr ⁻¹	GtC yr ⁻¹	–	Tmol Si yr ⁻¹	%
20 %	246.2	2097	2.02	11.07	0.99	0.089	76.8	52.0
40 %	250.0	2062	1.51	11.42	1.02	0.089	79.1	48.8
60 %	268.2	1963	1.49	12.14	1.08	0.089	86.4	47.7
80 %	273.1	1908	1.54	12.54	1.11	0.089	92.7	46.7
Control	278.0	1840	1.61	12.76	1.13	0.088	95.5	45.4
120 %	284.2	1770	1.62	12.99	1.14	0.088	98.6	43.5
140 %	290.1	1706	1.69	13.24	1.16	0.088	104.2	41.7
160 %	295.3	1640	1.77	13.46	1.17	0.087	110.7	40.0
180 %	300.0	1578	1.90	13.62	1.18	0.087	116.4	38.4

^aGlobal mean values below 2000 m depth.

^bDefined as the fraction of regenerated PO₄ versus total PO₄ after Ito and Follows (2005), Eq. (4).

**Fig. 7.** Response in atmospheric (a) CO₂, (b) $\delta^{13}\text{C}_{\text{CO}_2}$, and (c) $\Delta^{14}\text{C}_{\text{CO}_2}$ to changes in ocean ventilation. Displayed values are diagnosed for the time slice at 2 kyr after the perturbation in SO wind forcing. At this time carbon cycling has reached a new equilibrium in the ocean-only model (circles), whereas the ocean-sediment model (stars) is still under relaxation at 2 kyr due to the millennial-scale response of ocean-sediment interactions.

This value is close to the 312 ppm found by Ito and Follows (2005) on the basis of a simple theory combining the mass balance for carbon and nutrients in the ocean-atmosphere system and the linearization of carbonate chemistry. The relatively close correspondence between this theoretical value and the sensitivity of atmospheric CO₂ to P^* found in our experiments indicates that the predominant part of the simulated CO₂ response to changes in SO deep water formation is due to changes in the biological pump efficiency.

The simulated relation between atmospheric CO₂ and the soft-tissue pump efficiency P^* deviates from linearity as soon as NADW formation shuts down in response to weak SO winds (Fig. 8). Thus, the anomalous CO₂ reduction of ~10 ppm due to the AMOC collapse cannot be explained with changes in the biological carbon pumps. Further, the

absence of a drop in simulated global mean deep water temperature and in the global rain ratio (Table 2) seems to preclude a dominant role of the solubility and carbonate pumps. We conjecture that the simulated anomalous CO₂ drop in response to collapsed AMOC results primarily from an increase in the deep water DIC disequilibrium component (Marinov et al., 2008).

The simulated interplay between ocean ventilation, marine carbon cycling and atmospheric CO₂ displayed in our wind stress-experiments is in good qualitative agreement with the results of Schmittner et al. (2007), who find a similar response in marine biogeochemistry to changes in the global strength of wind stress. However, the effect of AMOC shut-downs on marine carbon cycling in our experiments cannot be readily compared with the corresponding results of

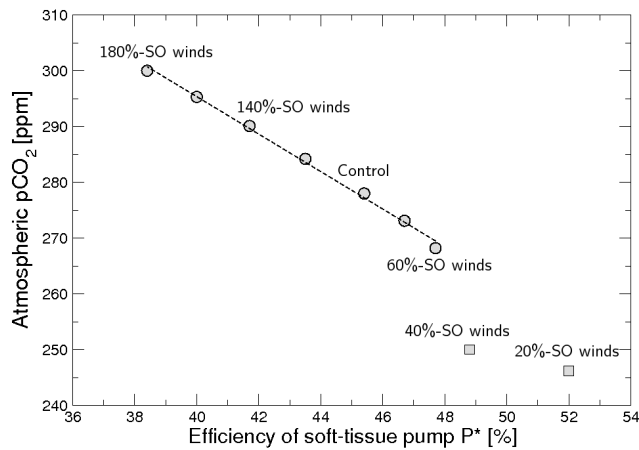


Fig. 8. Ocean ventilation experiments: quasi-linear relation between simulated atmospheric CO₂ and the efficiency of the soft-tissue pump P^* (Ito and Follows, 2005) for the ventilation states with active NADW formation (circles). The linear relation is perturbed when NADW formation is shutdown in response to strongly reduced windstress in the Southern Ocean (rectangles).

Schmittner et al. (2007), since AMOC is suppressed by freshwater forcing in the North Atlantic in their experiments rather than by weak SO winds.

The reorganization of the marine biological cycle is further reflected in the simulated change in the nutrient distribution and $\delta^{13}\text{C}_{\text{DIC}}$ signatures. The total phosphate concentrations decrease and $\delta^{13}\text{C}_{\text{DIC}}$ signatures increase in the sea surface as ocean ventilation is slowed and euphotic zone nutrients are utilized more efficiently (Table 3). After propagation of this signal across the air-sea interface $\delta^{13}\text{C}_{\text{CO}_2}$ is shifted towards less negative values in the atmosphere. The radiocarbon signature in atmospheric CO₂, $\Delta^{14}\text{C}_{\text{CO}_2}$, increases in response to reduced ventilation due to slowed physical transport of radiocarbon to the deep ocean which ultimately reduces ocean uptake of ^{14}C .

A specific consequence of the Southern Ocean ventilation mechanism is that vertical $\delta^{13}\text{C}_{\text{DIC}}$ -gradients in the Southern Ocean should have steepened during glacials (Hodell et al., 2003; Ninnemann and Charles, 2002). Our results reproduce this connection between deep mixing in the Southern Ocean and the vertical gradient in $\delta^{13}\text{C}_{\text{DIC}}$. Reduced deep mixing evidenced by a larger $\Delta^{14}\text{C}_{\text{DIC}}$ difference between the surface and deep Southern Ocean is accompanied by steeper vertical gradients in total phosphate and $\delta^{13}\text{C}_{\text{DIC}}$ (Table 3). According to our results, upon a reduction of 100 years in the difference between surface and deep water age (mean value below 2000 m) in the Southern Ocean, the $\delta^{13}\text{C}_{\text{DIC}}$ -gradient (difference between surface and mean signature below 2000 m in the Southern Ocean) increases by 0.16 ‰ and the vertical phosphate-gradient grows by 0.13 mol m⁻³ (Table 3).

2.3.3 The long-term effect of ocean-sediment interactions

The long-term millennial-scale trends in atmospheric CO₂, $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ arise from the slow response in sediment diagenesis to the changing biogeochemical state of the ocean. The sedimentary processes tend to restore in the long run burial rates of CaCO₃, POC and opal towards initial unperturbed values in order to re-establish the balance with the fixed weathering inputs.

The simulated variations in atmospheric CO₂ are amplified by 30 %–60 % within 50 kyr through the net effect of the sedimentary feedbacks subsequent to ventilation changes (Table 4). The overall simulated sedimentary CO₂ response is dominated by the changes in CaCO₃ burial. In fact, the shifts in organic matter burial counteract the simulated CO₂ variations as long as AMOC is maintained (on-states). For instance, accelerated ocean ventilation leads to a net rise in atmospheric CO₂, whereas elevated organic matter burial tends to reduce CO₂ by drawing carbon out of the ocean-atmosphere system. In the on-states, the sedimentary CO₂ response amplification is of 30 %–40 % (Table 4). In case of collapsing AMOC on the other hand, CO₂ changes are reinforced by the sedimentary feedbacks by as much as 60 %, since both carbonate compensation from reduced CaCO₃ export as well as excess POC burial driven by the reduction in deep ocean oxygen availability reinforce the CO₂ drawdown from slowed ventilation.

In contrast to the previous experiment in which the whole-ocean nutrient content is enhanced, the concomitant variations in deep ocean oxygen availability subsequent to ventilation changes weaken the importance of the nutrient-feedback by decoupling the link between shifts in POC export and in POC sediment burial. Enhanced (decreased) ventilation promotes (reduces) remineralization within the sediments through increased (decreased) oxygen concentrations in pore waters. A specific consequence is that the changes in POC export arising from altered deep ocean ventilation are maintained to a substantial degree in the long term (Fig. 6c). The POC burial flux, however, relaxes again to initial conditions to re-establish the balance with weathering input. In conclusion, the variations in oxygen availability in the ocean driven by the ventilation changes significantly decouple the link between POC export and POC burial. This decoupling complicates the reconstruction of past POC export on the basis of sedimentary records.

The sedimentary response in atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ is dominated by the temporal evolution of POC burial. For instance, in the model states with accelerated ocean ventilation, excess POC burial draws organic matter with low $\delta^{13}\text{C}$ (~ -20 ‰) out of the ocean-atmosphere system. As a consequence, the sediment response is to shift $\delta^{13}\text{C}_{\text{CO}_2}$ towards less negative values (Fig. 6g, red and yellow lines). Excess burial of POC and CaCO₃ both remove carbon with low $\Delta^{14}\text{C}$ from the ocean-atmosphere system in response to a better ventilated

Table 3. Ocean ventilation experiments in the ocean-sediment model: simulated changes in ocean biogeochemistry of different ocean regions at 2 kyr.

SO wind stress amplitude	North Atlantic below 2000 m (0°–71° N)					North Pacific below 2000 m (0°–63° N)				
	$\Delta^{14}\text{C}$ [‰]	$\delta^{13}\text{C}$ [‰]	PO ₄ [$\mu\text{mol l}^{-1}$]	PO ₄ ^{reg} [$\mu\text{mol l}^{-1}$]	O ₂ [$\mu\text{mol l}^{-1}$]	$\Delta^{14}\text{C}$ [‰]	$\delta^{13}\text{C}$ [‰]	PO ₄ [$\mu\text{mol l}^{-1}$]	PO ₄ ^{reg} [$\mu\text{mol l}^{-1}$]	O ₂ [$\mu\text{mol l}^{-1}$]
20 %	–260	0.171	2.54	1.31	118.2	–273	0.092	2.61	1.37	106.4
40 %	–258	0.179	2.55	1.23	134.7	–265	0.011	2.60	1.26	129.1
60 %	–195	0.632	1.98	0.88	191.0	–279	0.011	2.74	1.33	117.5
80 %	–188	0.671	1.94	0.85	194.8	–279	0.026	2.73	1.31	121.4
Control	–180	0.711	1.88	0.81	198.9	–278	0.049	2.70	1.28	127.1
120 %	–173	0.734	1.84	0.77	203.3	–276	0.077	2.68	1.22	136.7
140 %	–168	0.745	1.82	0.73	205.9	–274	0.102	2.66	1.17	144.6
160 %	–163	0.754	1.78	0.70	208.5	–273	0.123	2.64	1.13	151.4
180 %	–156	0.774	1.74	0.66	211.9	–272	0.141	2.62	1.10	156.9
SO wind stress amplitude	Southern Ocean below 2000 m (South of 51° S)					Southern Ocean surface (South of 51° S)				
	$\Delta^{14}\text{C}$ [‰]	$\delta^{13}\text{C}$ [‰]	PO ₄ [$\mu\text{mol l}^{-1}$]	PO ₄ ^{reg} [$\mu\text{mol l}^{-1}$]	O ₂ [$\mu\text{mol l}^{-1}$]	$\Delta^{14}\text{C}$ [‰]	$\delta^{13}\text{C}$ [‰]	$\delta^{13}\text{C}_{\text{S-D}}$ [‰]	PO ₄ [$\mu\text{mol l}^{-1}$]	O ₂ [$\mu\text{mol l}^{-1}$]
20 %	–196	0.385	2.27	1.06	160.8	–61	2.350	1.964	0.57	297.3
40 %	–198	0.377	2.30	0.98	178.9	–73	2.198	1.821	0.69	296.9
60 %	–208	0.308	2.40	1.01	175.3	–99	2.049	1.741	1.04	296.7
80 %	–208	0.330	2.39	0.98	180.1	–106	1.944	1.614	1.11	296.5
Control	–207	0.358	2.36	0.94	186.5	–112	1.871	1.513	1.18	295.8
120 %	–205	0.392	2.34	0.88	197.2	–119	1.799	1.407	1.24	295.4
140 %	–204	0.425	2.32	0.82	206.1	–125	1.725	1.301	1.31	295.0
160 %	–202	0.455	2.30	0.77	214.5	–130	1.658	1.203	1.37	294.6
180 %	–199	0.490	2.27	0.72	222.3	–135	1.603	1.113	1.41	294.5

deep ocean. As a result, atmospheric $\Delta^{14}\text{C}_{\text{CO}_2}$ is driven towards slightly less negative values (Fig. 6h, red and yellow lines).

Figure 6b (red and yellow lines) displays that an accelerated ocean ventilation further induces an initial deepening in the CaCO₃ saturation horizon which is offset after 1–2 kyr. The first part of the response results from the removal of carbon from the ocean interior as a result of rejuvenated deep water masses and excess POC sedimentation. The subsequent shoaling is due to the slowly responding effect of carbonate compensation to excess CaCO₃ burial.

The model experiments with collapsing AMOC exhibit the strongest reduction in nutrient supply to the surface and in globally integrated POC export from the surface (blue line in Fig. 6c). Global POC export remains reduced significantly in the long run. However, subsequent to the strong initial decrease, POC export in the North Atlantic eventually increases after about 200 years driven by diffusion of nutrients along isopycnals to the North Atlantic ocean surface. This response in POC export production is qualitatively similar to the results of Chikamoto et al. (2008) where AMOC has been shut down through a freshwater discharge in the

North Atlantic. In our model, we additionally find a long-term decreasing trend in global POC export after 5 kyr which is due to whole-ocean nutrient loss by excess POC burial. This nutrient-feedback has not been taken into account by Chikamoto et al. (2008).

Similarly to POC export, organic matter burial strongly declines during the first 200 years (blue line in Fig. 6e). Thereafter, the burial rate sharply recovers as a result of increasing POC export in the North Atlantic and a strongly reduced deep ocean oxygen supply. After around 2500 years, there is excess POC burial in spite of the reduction in global POC export production. Opal export and opal sediment burial exhibit a similar response as POC export and burial do (Fig. 6d and f): The AMOC shutdown induces major shifts also in the spatial pattern of opal export and sedimentation, with the North Atlantic becoming a region of deep nutrient upwelling similar to the North Pacific.

The reorganization in the marine biological carbon cycle in case of AMOC shutdowns is further reflected in the time-dependent model response for the depth of the CaCO₃ saturation horizon. The AMOC collapse induces an initial shoaling of the CaCO₃ saturation horizon as DIC accumulates in

Table 4. Long-term response in atmospheric CO₂ and amplification by sedimentary feedbacks.

SO wind stress amplitude	Global Deep Water Age [years]	Atmospheric CO ₂ without sediments after 50 kyr [ppm]	Atmospheric CO ₂ with sediments after 50 kyr [ppm]	Sedimentary amplification of the CO ₂ response [%]
20 %	2097	251	234	63 %
40 %	2062	252	236	62 %
60 %	1963	269	265	44 %
80 %	1908	273	271	40 %
Control	1840	278	278	–
120 %	1770	285	287	29 %
140 %	1706	290	294	33 %
160 %	1640	295	302	30 %
180 %	1578	299	308	30 %

the deep from the slowed ventilation and reduced POC burial (Fig. 6b). Consistently, the shoaling is more pronounced in the Atlantic than in the Pacific basin. After 1–2 kyr the saturation horizon starts to deepen again in both basins as deep [CO₃²⁻] rises from the combined effects of reduced CaCO₃ burial and increasing POC burial. The final steady state shift in the CaCO₃ saturation horizon is a deepening of 100–200 m in the Atlantic and of 500–600 m in the Pacific.

3 Discussion

Our modelling approach is associated with a number of conceptual and methodological limitations worth mentioning. First, the model scenarios are idealized and not intended to directly reflect the temporal evolution of paleo-proxy signals as recorded in climatic archives. Second, changes in deep ocean ventilation are generated by varying the magnitude of the wind stress in the Southern Ocean by more than a factor of two. This is a convenient technical tuning mechanism to adjust the deep ocean ventilation age in the model. However, we stress the fact that these variations in SO wind strength exceed the range of realistic changes. The atmospheric CO₂ response to realistic variations has been found to be rather modest (Tschumi, 2009; Menviel et al., 2008).

The model simulations have been tailored to investigate the specific response in atmospheric CO₂ and other paleo-proxies to uniform variations in deep ocean ventilation age. Several possibly important processes to explain the full glacial-interglacial CO₂ variability have not been taken into account. We have neglected, for instance, the well-recorded variations of AMOC (McManus et al., 2004), changes in the terrestrial biosphere, in sea surface temperatures or in sea ice cover. Further, the Bern3D+C is a model of intermediate complexity with coarse resolution and does not resolve eddies and other processes operating on small scales. Marine

biological production is represented as simple prognostic formulations considering the competition between opal and calcite producers and changes in the uptake ratio of silica to phosphate.

When interpreting the model results in the context of carbon cycle changes during the last deglaciation it should be kept in mind that the starting point of all sensitivity experiments is a stationary state corresponding to the preindustrial climate and carbon cycle. This modelling approach has been chosen, as the knowledge about glacial climate is subject to considerable uncertainty. The focus of this study is further laid upon a more qualitative interpretation of the interplay between different carbon cycle processes which depends less critically on the specific choice of initial conditions than the quantitative interpretation of a single paleoclimatic process or signal.

Nevertheless, we have performed an additional sensitivity experiment in order to quantify the uncertainty associated with the initial state of ocean ventilation: Starting from a stationary state with preindustrial carbon cycle boundary conditions but with low ocean ventilation as resulting from 60 %-SO winds, SO wind stress has been increased to 140 %. Comparing the response in atmospheric CO₂, δ¹³C_{CO₂} and Δ¹⁴C_{CO₂} with the corresponding 180 %-SO winds experiment starting from standard winds (red lines in Fig. 6) has shown that the sensitivity to the initial ventilation state of the ocean is negligible as long as there is no structural shift in the large-scale circulation pattern such as a breakdown of AMOC.

In spite of the inherent limitations mentioned above, the results of our study allow for a number of robust conclusions. We argue in the following that the processes represented in our model played a dominant role for reorganizing the global carbon cycle during the early part of the last deglaciation.

3.1 Southern Ocean ventilation hypothesis

The “Southern Ocean ventilation hypothesis” (e.g., Francois et al., 1997) is the starting point for discussion of our modelling results. According to this hypothesis, low glacial CO₂ values are related to a poorly ventilated deep ocean and a stably stratified Southern Ocean. Such a configuration would lead to an accumulation of carbon in the deep ocean during glacial periods relative to interglacials. The rise in atmospheric CO₂ over the glacial transition is driven by the break-up of Southern Ocean stratification and the resumption of modern-scale upwelling, bringing water rich in carbon, silica, and other nutrients and depleted in ¹³C and ¹⁴C to the surface. The characteristics of these upwelled water masses are further communicated within the intermediate and surface ocean by entrainment into the Subantarctic Mode Water and Antarctic Intermediate Water and to the atmosphere by air-sea gas-exchange. During recent years, a large number of observational evidence has been published supporting the view that changes in SO deepwater formation and in deep ocean ventilation have played an important role for glacial-interglacial CO₂ variations (e.g., Fischer et al., 2010; Spero and Lea, 2002; Adkins et al., 2002; Anderson et al., 2009; Muscheler et al., 2004; Brovkin et al., 2007; Hodell et al., 2003; Marchitto et al., 2007; Francois et al., 1997).

The response of our ocean-sediment carbon cycle model to strengthened Southern Hemisphere westerly winds reproduces some of the key mechanisms attributed to the “Southern Ocean ventilation hypothesis”. Invigorated wind stress promotes vertical exchange of water masses in the SO, thereby significantly affecting the physical and biogeochemical properties of deep ocean water on a global scale. Our results highlight in particular a rather close correlation between changes in ventilation age, the amount of biologically sequestered carbon and the $\delta^{13}\text{C}$ -signature in sea water as shown in Fig. 9.

In the following sub-sections we will discuss paleo-data from sediment analyses and from ice cores which indicate that large changes in Southern Ocean ventilation indeed occurred during the early part of the deglaciation, roughly in the interval between 18 ka BP and the onset of the Antarctic Cold reversal and the Bølling/Allerød Northern Hemisphere warming at about 14.6 ka BP (Severinghaus and Brook, 1999). We will then compare the data with our model results to constrain the magnitude of the change in atmospheric CO₂ over the last termination which is driven by a SO ventilation mechanism.

3.1.1 Changes in Southern Ocean opal sedimentation

Anderson et al. (2009) report reconstructed opal fluxes in Southern Ocean sediments which show a rapid increase starting roughly together with the rise in atmospheric CO₂ around 17 ka BP, followed by a deglacial maximum and a long-term

decrease into the Holocene (Fig. 9). Opal flux rates are typically several-fold higher during the transition compared to glacial and late-Holocene values. Anderson et al. (2009) interpret these opal flux changes as variations in Southern Ocean upwelling, concluding that there was a rapid increase in the upwelling of silica and carbon rich deep water to the Southern Ocean surface at the onset of deglaciation (~18 ka BP). They propose a direct link between the increased ventilation of deep water to the deglacial rise in atmospheric CO₂.

The step-wise strengthening of SO wind stress as applied in our ocean ventilation experiments drives an immediate rise in deep upwelling in the Southern Ocean in response to enhanced Ekman pumping and more vigorous deep convection in the model. These simulations may thus serve as model scenarios for an abrupt resumption of upwelling in the SO. Figure 9 shows that our model results for strengthened SO wind stress are in good qualitative agreement with the sediment data of Anderson et al. (2009) in relative magnitude and timing: Increasing SO wind stress by 80 % drives a rapid 30 %-rise in opal export as more silica is supplied to the surface. Additionally, opal sedimentation also increases and reaches a peak value with a delay of several millennia relative to the reinvigoration of deep upwelling in the Southern Ocean. The peak sedimentation flux is two to three times higher than the initial flux. Both opal export and sedimentation slowly decrease towards initial rates with a typical time scale of 20 kyr. Note however that simulated opal burial fluxes are about 10 times smaller than the reconstructed opal flux (Fig. 9). This is likely due to the spatial averaging over the model grid cell and the relatively low simulated global burial rate of opal as compared to available estimates (Table 1).

Anderson et al. (2009) suggest, based on the positive correlation between ²³¹Pa/²³⁰Th and opal flux in their records, that the deglacial maximum in opal flux reflects past changes in opal production and Si supply rather than variable opal preservation. In our model however, a relatively modest increase in global silica export of 30 % enhances the net flux to the sediments in a highly non-linear way as pore water concentrations of silicic acid and thus the preservation of opal in the sediments greatly increases under the enhanced rain of silica particles (Archer et al., 2000a; Ridgwell et al., 2002). The majority of the simulated change in opal sedimentation flux thus reflects increased preservation rather than increased opal production as suggested by Anderson et al. (2009).

Why does the reconstructed opal sedimentation flux relax towards glacial conditions over the course of the Holocene, whereas atmospheric CO₂ and, presumably, Southern Ocean upwelling remains at high interglacial values? From a continued large Southern Ocean upwelling, one might expect a continued silica supply to the surface ocean, and a sustained high export and sedimentation rate. In our simulation, increased opal burial drives a loss of whole-ocean silica. As a result, the simulated export of opal and its sedimentation rate

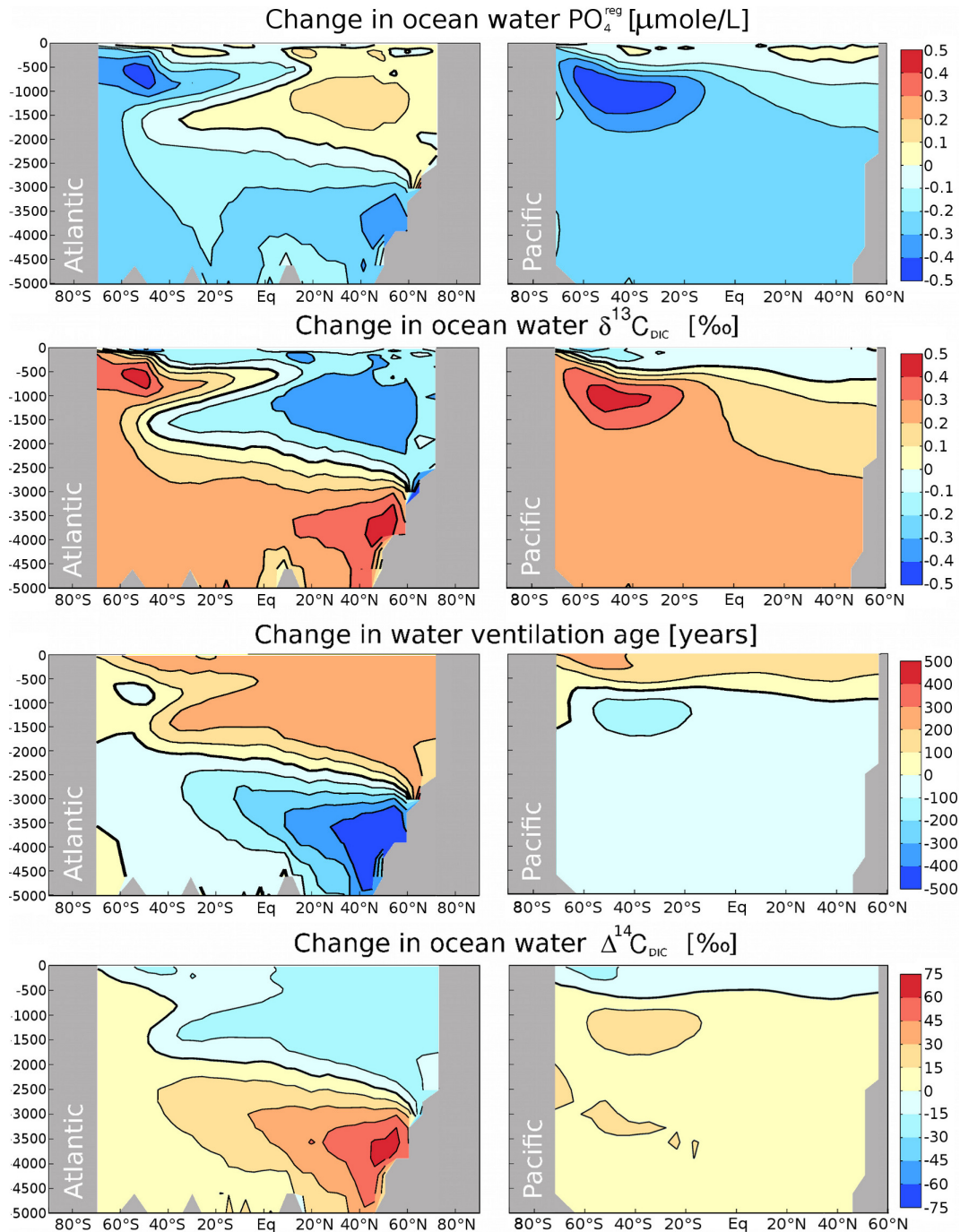


Fig. 9. Model response for $\delta^{13}\text{C}_{\text{DIC}}$, ventilation age and $\Delta^{14}\text{C}_{\text{DIC}}$ to stimulated deep mixing in the Southern Ocean and to more vigorous and deeper AMOC as a result of a 80 %-increase in SO wind stress amplitude. Shown are the simulated differences in Atlantic and Pacific zonal mean sections at 15 ka after the the step-wise perturbation of the model state.

relax towards initial conditions to balance riverine input with the burial flux of silica.

Is the evolution of the opal sedimentation flux directly correlated with opal export and the resumption of upwelling or are there any system time lags? The simulated peak in opal sedimentation is delayed by several millennia compared to the onset of the ventilation in our model. This suggests that

the temporal evolution of the ventilation history may be difficult to disentangle exactly from the temporal evolution of the opal sedimentation flux. Nevertheless, the opal records appear consistent with a change in deep ocean ventilation that occurred during the early deglacial period and that may or may not have been abrupt.

3.1.2 Changes in the stable carbon isotope ratio $\delta^{13}\text{C}$

Figure 10 displays the currently available $\delta^{13}\text{C}_{\text{CO}_2}$ records from Antarctic ice cores covering the period of the last deglaciation (Smith et al., 1999; Lourantou et al., 2010; Elsig, 2009) together with reconstructed atmospheric CO₂ (Monnin et al., 2001) and $\delta^{18}\text{O}_{\text{H}_2\text{O}}$ (Stenni et al., 2001), which is a proxy for local temperature on the ice sheet. The $\delta^{13}\text{C}_{\text{CO}_2}$ records consistently show an early deglacial decrease of 0.3–0.4 ‰ to reach a minimum at around 15 ka BP and a millennial-scale long-term increase by the same magnitude to attain approximately constant values at 7 ka BP (Elsig et al., 2009). However, the records disagree on the details of the $\delta^{13}\text{C}_{\text{CO}_2}$ evolution during the Bølling/Allerød-Younger Dryas climate swings.

Sediment cores from different ocean basins (Indo-Pacific, sub-Antarctic, and South Atlantic) reveal that the occurrence of carbon isotope minima at the beginning of glacial terminations is a widespread feature in planktic foraminifera $\delta^{13}\text{C}$ records (Spero and Lea, 2002). The decrease in the $\delta^{13}\text{C}$ records occurred simultaneously with the initiation of Southern Ocean warming. The data suggests a decrease in surface water $\delta^{13}\text{C}$ in the early deglacial period (~20 ka BP to 16 ka BP), an absolute minimum at the same time as the ice core record of Smith et al. (1999) which is followed by an increase over the rest of the deglacial period into the early Holocene. Taken together, both sediment and ice core data suggest decreasing atmospheric $\delta^{13}\text{C}$ during the first stage of the deglacial period followed by a long-term increase of similar magnitude.

How can this millennial-scale $\delta^{13}\text{C}$ evolution be explained? Our results lend support for the interpretation of Spero and Lea (2002) and Köhler et al. (2005) that the early deglacial $\delta^{13}\text{C}$ decrease reflected renewed Circumpolar Deep Water upwelling with an enhanced transport of low $\delta^{13}\text{C}$. A decrease in atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ of 0.3 ‰ would require a rejuvenation of deep ocean waters by 300 to 400 years according to the $\delta^{13}\text{C}_{\text{CO}_2}$ sensitivity found in our ventilation experiments (Fig. 7b). Subsequently, this negative signal is reduced by roughly 50 % over the following 15 kyr due to excess sedimentation of organic material (Fig. 6g). This would leave an increase of 0.1–0.2 ‰ to be explained by other processes such as carbon uptake by the terrestrial biosphere, surface ocean warming or sea ice retreat.

3.1.3 Changes in CaCO₃ saturation horizon depth

Analyses of the sedimentary bulk composition indicate a period of enhanced preservation of CaCO₃ deep sea sediments between ~15–7 ka BP (Ridgwell et al., 2003; Broecker et al., 2001). Broecker et al. (2001) attribute this sedimentary signal to an early deglacial restoration of the terrestrial biosphere. The resulting removal of carbon from the atmospheric and oceanic reservoirs is made responsible for a deepening of the CaCO₃ saturation horizon between about

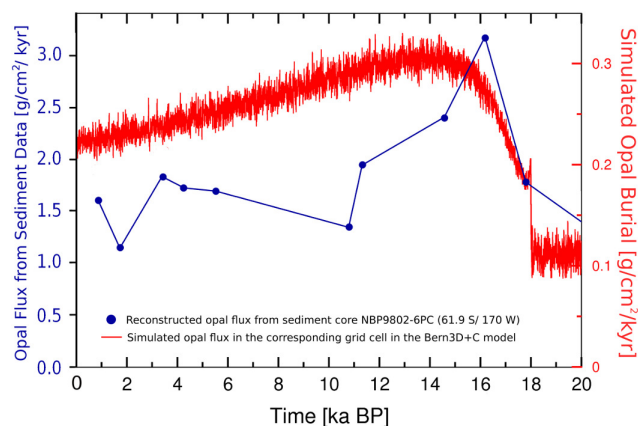


Fig. 10. Reconstructed local opal flux (blue line) at 61.9° S/170° W from Anderson et al. (2009) and simulated opal burial rate in the corresponding model cell (red line) as simulated in the experiment with an abrupt 80 %-increase in SO wind stress at 18 ka BP.

15–10 ka BP. The deepening is followed by a gradual shoaling until the latter part of the Holocene which is attributed to carbonate compensation. Ridgwell et al. (2003) invoke regrowth of coral reefs as additional factor driving the deglacial signal of the Pacific CaCO₃ saturation horizon to simultaneously account for the evolution of atmospheric CO₂.

In response to stimulated upwelling in the Southern Ocean rejuvenating deep waters by 300–400 years, our model predicts a similar temporal evolution for the Pacific CaCO₃ saturation horizon. We simulate a deepening saturation horizon by 300–400 m during the first ~2 kyr and a slow shoaling thereafter to reach initial depths after around 28 kyr (Fig. 6b). The simulated initial deepening is a result of the flushing of the old carbon-rich deep water mass from the Pacific basin and of excess organic matter sedimentation (Fig. 4b). This is consistent with deglacial CaCO₃ and organic matter flux reconstructions from the North Pacific (e.g., Jaccard et al., 2005; Galbraith et al., 2007) as well as with CO₃²⁻ reconstructions based on benthic Zn/Ca measured in the deep equatorial Pacific (Marchitto et al., 2005). However, the timing of our model results does not match the observations very precisely, maybe due to the unrealistic abrupt change in model boundary conditions. Nevertheless, the agreement is sufficient in order to indicate that increasing deep ocean ventilation and early deglacial excess sedimentation of organic matter may have significantly contributed to the recorded carbonate ion signal in the Pacific ocean in addition to the changes in terrestrial biosphere and growth of coral reefs as proposed by Broecker et al. (2001) and Ridgwell et al. (2003).

3.1.4 Changes in radiocarbon signatures

Is an increase in deep ocean ventilation consistent with marine radiocarbon data? To generate the full amplitude of

the early deglacial 35 ppm increase in atmospheric CO₂, a decrease in the deep ocean ventilation age of about 300–400 years is required according to the model-derived linear relationship between CO₂ and ventilation age (Fig. 7a). Corresponding simulated changes in deep ocean $\Delta^{14}\text{C}_{\text{DIC}}$ are an increase of 40–45 ‰ in the deep North Atlantic of 10–15 ‰ in the deep North Pacific and in the deep Southern Ocean (Table 4). Atmospheric $\Delta^{14}\text{C}$ would drop by 30–40 ‰ (Fig. 7c). The simulated atmosphere–deep ocean $\Delta^{14}\text{C}$ -difference would thus be reduced by 70–80 ‰ in the deep Atlantic and by 40–50 ‰ in the deep North Pacific and the deep Southern Ocean.

A decrease in deep ocean ventilation age of 300–400 years during the early deglacial period seems broadly consistent with available proxy records. However, the locally large changes between 18 ka BP and 14.5 ka BP in the benthic-planktonic age-difference of ~ 1000 years in the Atlantic sector of the SO (Skinner et al., 2010) or of ~ 700 years in the deep North Pacific (Galbraith et al., 2007) are not reproduced in our model. Overall, our results are in line with the model-based interpretation of the glacial–interglacial changes in past atmospheric $\Delta^{14}\text{C}$ by Köhler et al. (2006b) where a ~ 40 ‰-contribution is attributed to changes in ocean circulation, compared to the 30–40 ‰ from our results.

A difficulty in the interpretation of marine $\Delta^{14}\text{C}$ data in terms of ventilation age is that the $\Delta^{14}\text{C}$ signature at any given time does not just reflect the ventilation age, but also the time history of the atmospheric $\Delta^{14}\text{C}$. The life time of radiocarbon with respect to radioactive decay is 8267 years, implying a considerable memory to earlier $\Delta^{14}\text{C}$ variations. Atmospheric $\Delta^{14}\text{C}$ decreased by about 400 ‰ over the last transition (Hughen et al., 2004; Beck et al., 2001). A riddle presents the as yet unexplained drop of about 190 ‰ between 17.5 to 14.5 ka BP (Broecker and Barker, 2007). This drop is almost four times larger than the scenario-based decrease of 50 ‰ and apparently also finds no correspondence in deep ocean radiocarbon change (Broecker and Barker, 2007). We are cautious to draw conclusions at this stage and believe that further evaluations are required using transient simulations of the coevolution of atmospheric CO₂, $\Delta^{14}\text{C}$, and ^{14}C productivity.

4 Conclusions

Can we estimate the contribution of the Southern Ocean ventilation mechanism to the last deglacial CO₂ rise on the basis of our model results? CO₂ rose by about 35 ppm between the LGM and 15 ka BP (Fig. 10). If this CO₂ increase is entirely attributed to rejuvenating deep ocean waters, it would translate into a global reduction in deep water age of roughly 350 years, according to the model-derived linear relationship between CO₂ and ventilation age (Fig. 7a). Due to carbonate compensation in response to this ventilation change, CO₂ would increase by another ~ 10 ppm over

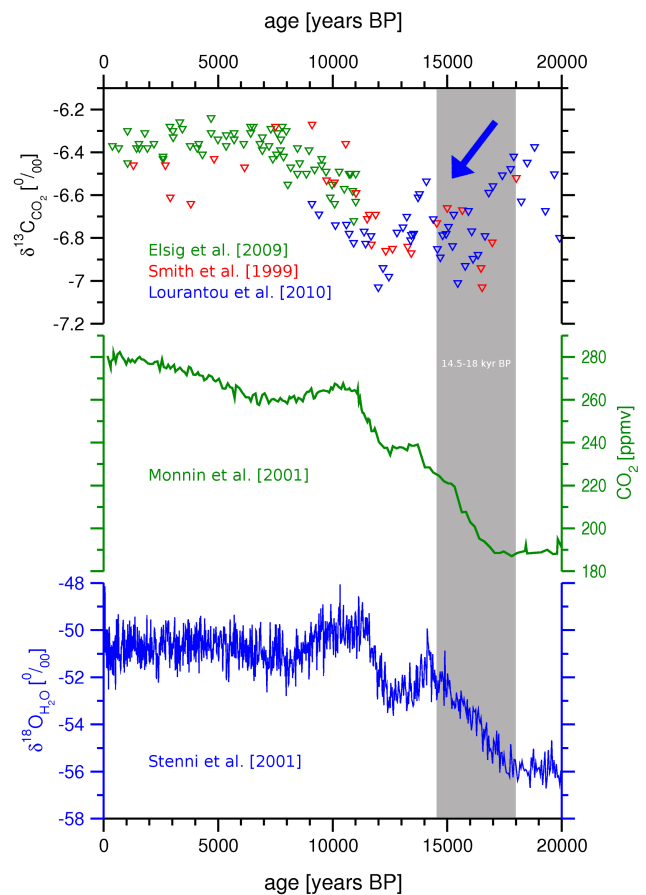


Fig. 11. Paleo-records from Antarctic ice cores: $\delta^{13}\text{C}$ -data is from Elsig et al. (2009) (green triangles), Smith et al. (1999) (red triangles) and Lourantou et al. (2010) (blue triangles). The blue arrow indicates the recorded drop in $\delta^{13}\text{C}_{\text{CO}_2}$ (~ -0.4 ‰) starting around 18 ka BP. CO₂-data is from Monnin et al. (2001) and $\delta^{18}\text{O}$ -data from Stenni et al. (2001).

the following 15 kyr. The early deglacial CO₂ increase might also reflect positive contributions from surface water temperature changes (Tschumi et al., 2008) and an estimated negative contribution from terrestrial uptake due to CO₂ fertilization on ice-free land (Joos et al., 2004). The two effects might have roughly cancelled. Taken together, our model results suggest that the Southern Ocean ventilation mechanism has contributed about half to the reconstructed total glacial–interglacial CO₂ increase of 90–100 ppm.

Our interpretation is further supported by a number of prominent paleo-proxy signals that are adequately reproduced in our simulations. As a response to an abrupt rejuvenation of deep ocean watermasses by 300–400 years stimulated by more vigorous deep convection in the Southern Ocean, we simulate a decline in $\delta^{13}\text{C}_{\text{CO}_2}$ by ~ 0.3 ‰ during the first 2 kyr followed by a gradual rise as observed in the paleo-records. Taken at face value, the ice core data suggest a slope of 12 ppm per 0.1 ‰ change in $\delta^{13}\text{C}_{\text{CO}_2}$ (35 ppm per

0.3 ‰) during the early deglacial period, whereas the model-derived slope amounts to 13 ppm per 0.1 ‰. The model response also agrees well with increased opal sedimentation rates in the Southern Ocean during the early deglaciation (Anderson et al., 2009) and is further consistent with the reconstructed evolution of the Pacific CaCO₃ saturation horizon (Broecker et al., 2001). We finally predict a drop of 30–40 ‰ in atmospheric $\Delta^{14}\text{C}_{\text{CO}_2}$ as another result of the invigorated deep ocean ventilation.

Although the simulated model response fits rather well with these specific paleo-proxy records, a perfect match should not be expected. Modeled carbon cycle sensitivities might be subject to considerable uncertainty, in particular due to unsettled questions related to the diffusivity of the oceans (Archer et al., 2000b). Further, it turned out difficult to reproduce the exact temporal evolution of $\delta^{13}\text{C}$ and $\Delta^{14}\text{C}$, as recorded in sediment cores in the model. These signals might sensitively depend upon regional and temporal details of the deglacial reorganization of global ocean circulation and may be further affected by the variations of AMOC (McManus et al., 2004) not considered here. Nonetheless, based on the multi-proxy comparison between our simulated response and the available paleo-data, we conclude that accelerating deep ocean ventilation linked to physical processes in the SO has been the dominant driver of early deglacial changes in global carbon cycling. Early deglacial ocean ventilation changes have continued to significantly contribute to variations in atmospheric CO₂ until the latter part of the Holocene, since it takes millennia for marine sedimentation to re-establish equilibrium with weathering input.

The glacial-interglacial CO₂ variations appear to be driven by a complex spatio-temporal pattern of processes involving relaxation time-scales of years to millennia. Our analysis of the time-dependent response in the coupled ocean-sediment system highlights that the benefit of conventional “time slice” approaches to study these processes might be rather limited. Our model results suggest that proxy signals arising from deglacial ventilation changes are sustained on a multi-millennial time-scale until the balance between carbon, nutrient and isotope input through rivers and winds and the loss through sedimentation is re-established. Thus, differences in these proxies between the LGM and the late Holocene time slice may partly reflect imbalances in the weathering-sedimentation cycles. This may also be relevant for proxies not included here such as silica or nitrogen isotopes.

We have identified a feedback between the marine nutrient inventory and sedimentation of organic matter. This mechanism emerges from the fact that marine biological productivity in the long run is limited by the riverine input of nutrients into the ocean. Our results indicate that this nutrient-sediment feedback might be important for glacial-interglacial changes in global carbon cycling. For instance, invigorated ocean circulation promotes export production and excess sediment burial of organic matter. The implications for

the combined ocean-atmosphere carbon cycle are similar to those of a postulated early deglacial restoration of the terrestrial biosphere (Broecker et al., 2001).

In this study, we have analyzed the time-dependent response in atmospheric CO₂ and its isotopic signatures only to the specific mechanisms of deep ocean ventilation changes and biological pump reorganizations. To improve the process-related understanding of the CO₂, $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$ records over the last deglaciation, further mechanisms such as sea surface temperature changes, coral reef growth, changes in the carbon stock on land and the spatio-temporal details of the various regional oceanic reorganizations should be considered in coupled 3-D global carbon cycle models. Simultaneous model-data comparison on the basis of multiple proxy records wisely chosen seems to be a promising approach towards a better understanding of the dynamical interplay between the multiple factors driving glacial-interglacial CO₂ fluctuations. A particular challenge is to identify and simulate the physical mechanisms responsible for the changes in Southern Ocean ventilation.

Appendix A

Sediment diagenesis model

The sediment diagenesis component used in this study is built on previous work by Heinze et al. (1999) and Gehlen et al. (2006) where the basic modelling concept is based on Archer et al. (1993). The spatial model domain is restricted to the diagenetical zone of the sediments which is assumed here to be the top 10 cm of the surface sediments. Any solid material leaving this domain disappears into the subjacent diagenetically consolidated zone. The interfaces of the 10 model layers are defined at depths of 0, 0.3, 0.6, 1.1, 1.6, 2.1, 3.1, 4.1, 5.1, 7.55, and 10.0 cm. The fraction of pore water volume (porosity) decreases from 0.95 to 0.75, following observations from Ullman and Aller (1982). Model geometry remains fixed during integration.

The vertical movement of solid material in the sediment column is calculated on the basis of the equilibrium: burial = rain redissolution. When the particulate rain onto the sediment is larger than the redissolution of solid material within the diagenetical zone, the entire column is shifted downwards and sediment burial occurs. Conversely, whenever redissolution is larger than rain, gaps of solid material are produced within the model geometry. These are filled with clay from below. Lateral movements in the sediments are not considered.

Modeled tracers are the four solid components (CaCO₃, opal, POM and clay) and the six pore water substances (DIC, total alkalinity, phosphate, nitrate, oxygen and silicic acid). CaCO₃ is only considered in the form of calcite. The less stable aragonite form is assumed to be only of secondary importance for the global sediments (Bernier and Honjo, 1981).

Table A1. Parameters for the sediment component of the Bern3D+C model along with standard values and references.

Symbol	Parameter	Value	Reference
	Number of time steps per year	48	this study
ρ	Density of solid sediment components	2.6 g cm ⁻³	Heinze et al. (1999)
M_{POM}	Molar weight of POM	32.74 g mol ⁻¹	Heinze et al. (1999)
M_{CaCO_3}	Molar weight of CaCO ₃	100.0 g mol ⁻¹	Heinze et al. (1999)
M_{opal}	Molar weight of opal	67.2 g mol ⁻¹	Heinze et al. (1999)
M_{clay}	Molar weight of clay	430.51 g mol ⁻¹	Heinze et al. (1999)
	Terrestrial clay flux	1 g m ⁻² yr ⁻¹	Heinze et al. (1999)
r_{CaCO_3}	Rate constant for CaCO ₃ redissolution	1000 l mol ⁻¹ yr ⁻¹	Heinze et al. (1999)
r_{opal}	Rate constant for opal redissolution	20 l mol ⁻¹ yr ⁻¹	this study
r_{oxy}	Rate constant for oxydation	50 l mol ⁻¹ yr ⁻¹	Heinze et al. (1999)
r_{denit}	Rate constant for denitrification	50 l mol ⁻¹ yr ⁻¹	Gehlen et al. (2006)
$[\text{Si}]_{\text{sat}}$	Saturation concentration for pore water silicic acid	800 μmol l ⁻¹	Heinze et al. (2003)
D	Pore water diffusion coefficient	8. × 10 ⁻⁶ cm ² s ⁻¹	Heinze et al. (1999)
D_{B}	Bioturbation diffusion coefficient	15. × 10 ⁻³ cm ² yr ⁻¹	Heinze et al. (1999)
$[\text{O}_2]_{\text{crit}}$	Critical oxygen concentration for denitrification	1 μmol l ⁻¹	Gehlen et al. (2006)
$\mathcal{R}_{\text{PO}_4}$	Redfield coefficient for phosphate	1	Anderson and Sarmiento (1994)
\mathcal{R}_{O_2}	Redfield coefficient for oxygen	170	Anderson and Sarmiento (1994)
$\mathcal{R}_{\text{NO}_3}$	Redfield coefficient for nitrate	16	Anderson and Sarmiento (1994)
\mathcal{R}_{C}	Redfield coefficient for carbon	117	Anderson and Sarmiento (1994)
$\mathcal{R}_{\text{denit}}$	Redfield coefficient for denitrification	97.6	Gehlen et al. (2006)

All the solid components are taken to have equal densities ($\rho = 2.6 \text{ g cm}^{-3}$).

Considered processes occurring during sediment diagenesis are the dissolution of opal and CaCO₃ as well as the oxydation and denitrification of POM. The mixing of sediments by burrowing animals such as clams and worms (bioturbation) is parametrized as a diffusion of solid components with closed boundaries at the sediment-water interface and at the bottom of the bioturbated zone.

The mathematical formulation of the sediment model consists of two sets of dynamical equations, one for the four solid components and one for the six pore water substances, along with adequate boundary conditions. The two sets of equations are coupled to each other through the four reaction rates of the sediment-pore water processes (redissolution / remineralization). All of these reactions follow first order kinetics with spatially uniform reaction rate constants. The model parameters of the sediment component are listed in Table A1 along with their standard values and references.

A1 Redissolution/remineralization

The rate of opal dissolution R_{opal} [mol Si m⁻³ yr⁻¹] is proportional to the amount of opal in the sediment, c_{opal} , the pore water undersaturation with respect to silicic acid, $([\text{Si}]_{\text{sat}} - [\text{Si}])$, and the rate constant, r_{opal} :

$$R_{\text{opal}} = r_{\text{opal}} \cdot c_{\text{opal}} \cdot \Phi \cdot (1 - \Phi) \cdot \frac{\rho}{M_{\text{opal}}} \cdot ([\text{Si}]_{\text{sat}} - [\text{Si}]) \cdot H([\text{Si}]_{\text{sat}} - [\text{Si}]). \quad (\text{A1})$$

M_{opal} stands for the molar weight of opal, ρ for the density, Φ for porosity and c_{opal} corresponds to the weight-fraction of opal sediments. $H(x)$ is the Heaviside step function, preventing spontaneous precipitation of solid opal in case of pore water supersaturation. The silicic acid saturation concentration, $[\text{Si}]_{\text{sat}}$, is a spatially constant model parameter set to 800 μmol l⁻¹.

Since $[\text{Ca}^{2+}]$ is fairly constant and has a relatively high background concentration, CaCO₃ preservation is primarily governed by the concentration of carbonate ions in the pore water. The rate of CaCO₃ dissolution R_{CaCO_3} [mol C m⁻³ yr⁻¹] is thus linearly related to the pore water undersaturation with respect to carbonate ions, $([\text{CO}_3]_{\text{sat}} - [\text{CO}_3])$, the weight-fraction of CaCO₃, c_{CaCO_3} , and the rate constant, r_{CaCO_3} :

$$R_{\text{CaCO}_3} = r_{\text{CaCO}_3} \cdot c_{\text{CaCO}_3} \cdot \Phi \cdot (1 - \Phi) \cdot \frac{\rho}{M_{\text{CaCO}_3}} \cdot ([\text{CO}_3]_{\text{sat}} - [\text{CO}_3]) \cdot H([\text{CO}_3]_{\text{sat}} - [\text{CO}_3]), \quad (\text{A2})$$

$$[\text{CO}_3]_{\text{sat}} = \frac{[\text{Ca}^{2+}]}{K'_{\text{sp}}}$$

The solubility product of calcite K'_{sp} is calculated after Mucci (1983) with a pressure dependence following Millero (1995). The calcium ion concentration $[Ca^{2+}]$ is proportional to salinity.

The rate of organic matter oxydation R_{oxy} [$mol\ C\ m^{-3}\ yr^{-1}$] is limited by the availability of oxygen in the pore water, $[O_2]$, and by the amount of POC, c_{POC} , in the sediment:

$$R_{oxy} = r_{oxy} \cdot c_{POC} \cdot \Phi \cdot (1 - \Phi) \cdot \frac{\rho}{M_{POC}} \cdot \frac{\mathcal{R}_C}{\mathcal{R}_{O_2}} \quad (A3)$$

$$\cdot [O_2] \cdot H([O_2]).$$

The rate constant for oxydation, r_{oxy} , is set to $501\ mol^{-1}\ yr^{-1}$. \mathcal{R}_C and \mathcal{R}_{O_2} are the Redfield ratios for carbon and oxygen, respectively.

Organic matter denitrification only occurs in suboxic conditions, i.e. below a critical oxygen concentration in the pore water $[O_2]_{crit}$. The reaction rate R_{denit} [$mol\ C\ m^{-3}\ yr^{-1}$] is proportional to the POC weight-fraction, c_{POC} , the availability of nitrate, $[NO_3^-]$, and the rate constant r_{denit} :

$$R_{denit} = r_{denit} \cdot c_{POC} \cdot \Phi \cdot (1 - \Phi) \cdot \frac{\rho}{M_{POC}} \cdot \frac{\mathcal{R}_C}{\mathcal{R}_{denit}} \quad (A4)$$

$$\cdot [NO_3^-] \cdot H([NO_3^-]) \cdot H([O_2]_{crit} - [O_2]).$$

\mathcal{R}_{denit} is the Redfield coefficient for denitrification. The reaction rate, r_{denit} , is set to $501\ mol^{-1}\ yr^{-1}$.

Inert clay is not subject to redissolution in the sediments:

$$R_{clay} = 0. \quad (A5)$$

A2 Solid components

The four solid sediment components are subject to bioturbation, advection and remineralization/redissolution. They satisfy the following equations:

$$\frac{\partial c_{CaCO_3}}{\partial t} = D_B \frac{\partial^2}{\partial z^2} (c_{CaCO_3}) - \frac{\partial}{\partial z} (w c_{CaCO_3}) \quad (A6)$$

$$- R_{CaCO_3} \cdot \frac{M_{CaCO_3}}{\rho} \cdot \frac{1}{1 - \Phi},$$

$$\frac{\partial c_{opal}}{\partial t} = D_B \frac{\partial^2}{\partial z^2} (c_{opal}) - \frac{\partial}{\partial z} (w c_{opal}) - R_{opal} \quad (A7)$$

$$\cdot \frac{M_{opal}}{\rho} \cdot \frac{1}{1 - \Phi},$$

$$\frac{\partial c_{POC}}{\partial t} = D_B \frac{\partial^2}{\partial z^2} (c_{POC}) - \frac{\partial}{\partial z} (w c_{POC}) \quad (A8)$$

$$- (R_{oxy} + R_{denit}) \cdot \frac{M_{POC}}{\rho} \cdot \frac{1}{1 - \Phi},$$

$$\frac{\partial c_{clay}}{\partial t} = D_B \frac{\partial^2}{\partial z^2} (c_{clay}) - \frac{\partial}{\partial z} (w c_{clay}). \quad (A9)$$

c_S corresponds to the dry weight fraction of CaCO₃, opal, POC or clay. D_B is the bioturbation coefficient, w is the vertical advection velocity and R_x are the reaction rates. The variable porosities are taken into account when solving Eqs. (A6)–(A9).

Below the bioturbated zone, the sediment composition remains unchanged:

$$\frac{\partial c_S}{\partial z} \Big|_{z \geq 10\text{cm}} = 0, \quad \text{for each solid component } S. \quad (A10)$$

At the sediment-water interface ($z=0$ cm) the vertical advection velocity is given by the sum of the volume rain rates, $Rains_S \cdot M_S/\rho$ [$cm^3\ cm^{-2}\ yr^{-1}$], divided by the solid sediment fraction, $(1 - \Phi(0))$:

$$w(0) = \frac{1}{1 - \Phi(0)} \sum_{\text{solid components}} Rains_S \cdot M_S/\rho. \quad (A11)$$

Vertical advection velocities, $w(z)$, decrease with depth due to remineralization and redissolution as well as due to increasing solid sediment fractions:

$$w(z) < w(0), \quad \text{for } z > 0. \quad (A12)$$

The velocity $w(z)$ is given by the remaining volume flux at depth z divided by the solid sediment fraction, $(1 - \Phi(z))$:

$$w(z) = \frac{1}{1 - \Phi(z)} \left(\sum_{\text{solid components}} Rains_S \cdot M_S/\rho \quad (A13)$$

$$- \int_{z'=0}^z \sum_{\text{solid components}} R_S(z') \cdot M_S(z')/\rho \cdot dz' \right).$$

The burial flux of the solid component S at the base of the bioturbated zone ($z=10$ cm), F_S^{burial} [$mol\ cm^{-2}\ yr^{-1}$], is given by:

$$F_S^{burial} = w(10) \cdot c_S(10) \cdot \frac{\rho}{M_S} \cdot (1 - \Phi(10)). \quad (A14)$$

In case total redissolution is larger than total rain in the sediment column, gaps of solid material are produced within the fixed model geometry. These are filled with clay in order to ensure mass conservation:

$$c_{opal} + c_{cal} + c_{org} + c_{clay} = 1. \quad (A15)$$

A3 Pore water solutes

The concentrations of pore water solutes are subject to sediment remineralization and redissolution as well as to pore water diffusion. Here, the diffusion in a porous medium is represented by multiplying the uniform pore water diffusion coefficient D by the porosity Φ . Experimentally derived diffusion coefficients do not deviate much from these values

(Ullman and Aller, 1982). The dynamical equations for the pore water solutes are:

$$\frac{\partial \text{Si}}{\partial t} = \frac{\partial}{\partial z} \left(\Phi D \frac{\partial \text{Si}}{\partial z} \right) + R_{\text{opal}}/\Phi, \quad (\text{A16})$$

$$\frac{\partial \text{O}_2}{\partial t} = \frac{\partial}{\partial z} \left(\Phi D \frac{\partial \text{O}_2}{\partial z} \right) - \frac{\mathcal{R}_{\text{O}_2}}{\mathcal{R}_C} \cdot R_{\text{oxy}}/\Phi, \quad (\text{A17})$$

$$\begin{aligned} \frac{\partial \text{NO}_3^-}{\partial t} = & \frac{\partial}{\partial z} \left(\Phi D \frac{\partial \text{NO}_3^-}{\partial z} \right) + \frac{\mathcal{R}_{\text{NO}_3^-}}{\mathcal{R}_C} \\ & \cdot R_{\text{oxy}}/\Phi - \frac{\mathcal{R}_{\text{denit}}}{\mathcal{R}_C} \cdot R_{\text{denit}}/\Phi, \end{aligned} \quad (\text{A18})$$

$$\begin{aligned} \frac{\partial \text{PO}_4}{\partial t} = & \frac{\partial}{\partial z} \left(\Phi D \frac{\partial \text{PO}_4}{\partial z} \right) + \frac{\mathcal{R}_{\text{PO}_4}}{\mathcal{R}_C} \\ & \cdot (R_{\text{oxy}} + R_{\text{denit}})/\Phi, \end{aligned} \quad (\text{A19})$$

$$\begin{aligned} \frac{\partial \text{DIC}}{\partial t} = & \frac{\partial}{\partial z} \left(\Phi D \frac{\partial \text{DIC}}{\partial z} \right) \\ & + (R_{\text{calc}} + R_{\text{oxy}} + R_{\text{denit}})/\Phi, \end{aligned} \quad (\text{A20})$$

$$\begin{aligned} \frac{\partial \text{ALK}}{\partial t} = & \frac{\partial}{\partial z} \left(\Phi D \frac{\partial \text{ALK}}{\partial z} \right) - \frac{\mathcal{R}_{\text{NO}_3^-} + 2 \cdot \mathcal{R}_{\text{PO}_4}}{\mathcal{R}_C} \\ & \cdot R_{\text{oxy}}/\Phi + \frac{\mathcal{R}_{\text{denit}} - 2 \cdot \mathcal{R}_{\text{PO}_4}}{\mathcal{R}_C} \cdot R_{\text{denit}}/\Phi \\ & + 2 \cdot R_{\text{cal}}/\Phi. \end{aligned} \quad (\text{A21})$$

The boundary condition for pore water solutes at the ocean-sediment interface is:

$$C|_{z=0\text{cm}} = C_{\text{bottom water}}, \quad \text{for each solute tracer } C. \quad (\text{A22})$$

Below the bioturbated zone the solute concentrations are assumed to remain constant:

$$\left. \frac{\partial C}{\partial z} \right|_{z \geq 10\text{cm}} = 0, \quad \text{for each solute tracer } C. \quad (\text{A23})$$

The pH-equilibrium in the pore water is given by:

$$\frac{[\text{CO}_2] [\text{CO}_3^{=}]}{[\text{HCO}_3^-]^2} = \frac{K'_1}{K'_2}. \quad (\text{A24})$$

The apparent dissociation constants K'_1 and K'_2 are calculated after Mehrbach et al. (1973) with pressure dependence following Millero (1995).

Acknowledgements. This study received support by the Swiss National Science Foundation, by the Swiss Staatssekretariat für Bildung und Forschung (#C07.0068; COST Action 735), and by the European Commission through the FP7 projects Past4Future (grant no. 243908), CARBOCHANGE (grant no. 264879), and EPOCA (grant no. 211384). This is publication no. A 343 of the Bjerknes Centre for Climate Research. The comments by T. F. Stocker are greatly appreciated.

Edited by: J. Singarayer

References

- Adkins, J., McIntyre, K., and Schrag, D.: The salinity, temperature and $\delta^{18}\text{O}$ content of the glacial deep ocean, *Science*, 298, 1769–1773, 2002.
- Anderson, L. and Sarmiento, J.: Redfield ratios of remineralization determined by nutrient data analysis, *Global Biogeochem. Cy.*, 8, 65–80, 1994.
- Anderson, R., Ali, S., Bradtmiller, L., Nielsen, S., Fleisher, M., Anderson, B., and Burckle, L.: Wind-Driven Upwelling in the Glacial Southern Ocean and the Deglacial Rise in Atmospheric CO₂, *Science*, 323, 1443–1448, 2009.
- Archer, D. and Maier-Reimer, E.: Effect of deep-sea sedimentary calcite preservation on atmospheric CO₂ concentration, *Nature*, 367, 292–294, 1994.
- Archer, D., Lyle, M., Rodgers, K., and Froelich, P.: What Controls Opal Preservation in Tropical Deep Sea Sediments?, *Paleoceanography*, 8, 7–21, 1993.
- Archer, D., Peltzer, E., and Kirchman, D.: A timescale for dissolved organic carbon production in equatorial Pacific surface waters, *Global Biogeochem. Cy.*, 11, 435–452, 1997.
- Archer, D., Winguth, A., Lea, D., and Mahowald, N.: What caused the glacial/interglacial atmospheric $p\text{CO}_2$ cycles?, *Rev. Geophys.*, 38, 159–189, 2000a.
- Archer, D., Eshel, G., Winguth, A., Broecker, W., Pierrehumbert, R., Tobis, M., and Jacob, R.: Atmospheric $p\text{CO}_2$ sensitivity to the biological pump of the cean, *Global Biogeochem. Cy.*, 14, 1219–1230, 2000b.
- Aumont, O. and Bopp, L.: Globalizing results from ocean in situ iron fertilization studies, *Global Biogeochem. Cy.*, 20, GB2017, doi:10.1029/2005GB002591, 2006.
- Beck, J., Warren, J., Richards, D., Edwards, R., Smart, B. S. P., Donahue, D., Herrera-Osterheld, S., Burr, G., Calsoyas, L., Jull, A., and Biddulph, D.: Extremely Large Variations of Atmospheric ¹⁴C Concentration During the Last Glacial Period, *Science*, 292, 2453–2458, doi:10.1126/science.1056649, 2001.
- Berger, W. and Keir, R.: Glacial-Holocene Changes in atmospheric CO₂ and the deep-sea record, in: *Climate Processes and Climate Sensitivity*, edited by: Hansen, J. E. and Takahashi, T., 303–317, 1984.
- Berner, R. and Honjo, S.: Pelagic Sedimentation of Aragonite: Its Geochemical Significance, *Science*, 211, 940–942, doi:10.1126/science.211.4485.940, 1981.
- Bird, M., Lloyd, J., and Farquhar, G.: Terrestrial carbon storage at the LGM, *Nature*, 371, 566, doi:10.1038/371566a0, 1994.
- Böning, C., Disper, A., Visbeck, M., Rintoul, S., and Schwarzkopf, F.: The response of the Antarctic Circumpolar Current to recent climate change, *Nat. Geosci.*, 1, 864–869, 2008.
- Boyle, E.: Vertical oceanic nutrient fractionation and glacial/interglacial CO₂ cycles, *Nature*, 331, 55–56, 1988.
- Bradtmiller, L., Anderson, R., Sachs, J., and Fleisher, M.: A deeper respired carbon pool in the glacial equatorial Pacific Ocean, *Earth Planet. Sc. Lett.*, 299, 417–425, 2010.
- Broecker, W.: Ocean chemistry during glacial time, *Geochim. Cosmochim. Acta*, 46, 1689–1706, 1982.
- Broecker, W. and Barker, S.: A 190‰ drop in atmosphere's $\Delta^{14}\text{C}$ during the “Mystery Interval” (17.5 to 14.5 kyr), *Earth Planet. Sc. Lett.*, 256, 90–99, 2007.

- Broecker, W. and Peng, T.: The role of CaCO₃ compensation in the glacial to interglacial atmospheric CO₂ change, *Global Biogeochem. Cy.*, 1, 15–29, 1987.
- Broecker, W. and Peng, T.-H.: *Tracers in the Sea*, Eldigo, Palisades, N. Y., 1982.
- Broecker, W., Lynch-Stieglitz, J., Clark, E., Hajdas, I., and Bonani, G.: What caused the atmosphere's CO₂ content to rise during the last 8000 years?, *Geochem. Geophys. Geosys.*, 2(10), 1062–1074, doi:10.1029/2001GC000177, 2001.
- Broecker, W., Clark, E., Barker, S., Hajdas, I., Bonani, G., and Moreno, E.: Radiocarbon age of late glacial deep water from the equatorial Pacific, *Paleoceanography*, 22, PA2206, doi:10.1029/2006PA001359, 2007.
- Brovkin, V., Ganopolski, A., Archer, D., and Rahmstorf, S.: Lowering of glacial atmospheric CO₂ in response to changes in ocean circulation and marine biogeochemistry, *Paleoceanography*, 22, PA4202, doi:10.1029/2006PA001380, 2007.
- Chikamoto, M., Matsumoto, K., and Ridgeway, A.: Response of deep-sea CaCO₃ sedimentation to Atlantic meridional overturning circulation shutdown, *J. Geophys. Res.*, 113, G03017, doi:10.1029/2007JG000669, 2008.
- Chikamoto, M., Matsumoto, K., and Yamanaka, Y.: Influence of Export Ratio Changes in Atmospheric CO₂ and Sedimentary Calcite Preservation, *J. Oceanogr.*, 65, 209–221, 2009.
- Crusius, J., Pedersen, T., Kienast, S., and Keigwin, L.: Influence of northwest Pacific productivity on North Pacific Intermediate Water oxygen concentrations during the Bolling-Allerod interval (14.7–12.9 ka), *Geology*, 32, 633–636, 2004.
- Curry, W., Labeyrie, J. D. L., and Shackleton, N.: Changes in the Distribution of $\delta^{13}\text{C}$ of Deep Water ΣCO_2 Between the Last Glaciation and the Holocene, *Paleoceanography*, 3, 317–341, 1988.
- De La Rocha, C.: Opal-based isotopic proxies of paleoenvironmental conditions, *Global Biogeochem. Cy.*, 20, GB4S09, doi:10.1029/2005GB002664, 2006.
- Deusch, C., Sigman, D., Thunell, R., Meckler, A., and Haug, G.: Isotopic constraints on glacial/interglacial changes in the oceanic nitrogen budget, *Global Biogeochem. Cy.*, 18, GB4012, doi:10.1029/2003GB002189, 2004.
- Doney, S., Glover, D., and Najjar, R.: A new coupled, one-dimensional biological-physical model for the upper ocean: Applications to the JGOFS Bermuda Atlantic Time-series Study (BA TS) site, *Deep-Sea Res. Pt. II*, 43, 591–624, 1996.
- Duplessy, J. C., Shackleton, N. J., Fairbanks, R. G., Labeyrie, L., Oppo, D., and Kallel, N.: Deepwater source variations during the last climatic cycle and their impact on the global deepwater circulation, *Paleoceanography*, 3, 343–360, 1988.
- Edwards, N. R. and Marsh, R.: Uncertainties due to transport-parameter sensitivity in an efficient 3-d ocean-climate model, *Clim. Dynam.*, 3, 67–94, 2005.
- Edwards, N. R., Willmott, A. J., and Killworth, P. D.: On the role of topography and wind stress on the stability of the thermohaline circulation, *J. Phys. Oceanogr.*, 28, 756–778, 1998.
- Elsig, J.: New insights into the global carbon cycle from measurements of CO₂ stable isotopes: methodological improvements and interpretation of a new EPICA Dome C ice core $\delta^{13}\text{C}$ record, Ph.D. thesis, Climate and Environmental Physics, Physics Institute, University of Bern, 2009.
- Elsig, J., Schmitt, J., Leuenberger, D., Schneider, R., Eyer, M., Leuenberger, M., Joos, F., Fischer, H., and Stocker, T.: Stable isotope constraints on Holocene carbon cycle changes from an Antarctic ice core, *Nature*, 461, 507–510, 2009.
- Falkowski, P.: Evolution of the nitrogen cycle and its influence on the biological sequestration of CO₂ in the ocean, *Nature*, 387, 272–275, 1997.
- Feely, R., Sabine, C., Lee, K., Berelson, W., Kleypas, J., Fabry, V., and Millero, F.: Impact of Anthropogenic CO₂ on the CaCO₃ System in the Oceans, *Science*, 305, 362–366, doi:10.1126/science.1097329, 2004.
- Fischer, H., Schmitt, J., Lüthi, D., Stocker, T., Tschumi, T., Parekh, P., Joos, F., Köhler, P., Völker, C., Gersonde, R., Barbante, C., Floch, M. L., Raynaud, D., Barnola, J.-M., and Wolff, E.: The role of Southern Ocean processes on orbital and millennial CO₂ variations – a synthesis, *Quaternary Sci. Rev.*, 29, 193–205, 2010.
- Francois, R., Altabet, M., Yu, E., Sigman, D., Bacon, M., Frank, M., Bohrmann, G., Bareille, G., and Labeyrie, L.: Contribution of Southern Ocean surface-water stratification to low atmospheric CO₂ concentrations during the last glacial period, *Nature*, 389, 929–935, 1997.
- Freeman, K. H. and Hayes, J. M.: Fractionation of carbon isotopes by phytoplankton and estimates of ancient CO₂ levels, *Global Biogeochem. Cy.*, 6, 185–198, 1992.
- Galbraith, E., Jaccard, S., Pedersen, T., Sigman, D., Haug, G., Cook, M., Southon, J., and Francois, R.: Carbon dioxide release from the North Pacific abyss during the last deglaciation, *Nature*, 449, 890–893, 2007.
- Gehlen, M., Bopp, L., Emprin, N., Aumont, O., Heinze, C., and Ragueneau, O.: Reconciling surface ocean productivity, export fluxes and sediment composition in a global biogeochemical ocean model, *Biogeosciences*, 3, 521–537, doi:10.5194/bg-3-521-2006, 2006.
- Heinze, C., Maier-Reimer, E., and Winn, K.: Glacial *p*CO₂ Reduction by the World Ocean: Experiments With the Hamburg Carbon Cycle Model, *Paleoceanography*, 6, 393–430, 1991.
- Heinze, C., Maier-Reimer, E., Winguth, A., and Archer, D.: A global oceanic sediment model for long-term climate studies, *Global Biogeochem. Cy.*, 13, 221–250, 1999.
- Heinze, C., Hupe, A., Maier-Reimer, E., Dittert, N., and Ragueneau, O.: Sensitivity of the marine biospheric Si cycle for biogeochemical parameter variations, *Global Biogeochem. Cy.*, 17, 1086, doi:10.1029/2002GB001943, 2003.
- Hodell, D., Venz, K., Charles, C., and Ninnemann, U.: Pleistocene vertical carbon isotope and carbonate gradients in the South Atlantic sector of the Southern Ocean, *Geochem., Geophys., Geosys.*, 4(1), 1004, doi:10.1029/2002GC000367, 2003.
- Hughen, K. S., Lehman, J. S., Overpeck, J., Marchal, O., Her-ring, C., and Turnbull, J.: ¹⁴C Activity and Global Carbon Cycle Changes over the Past 50'000 Years, *Science*, 303, 202–207, 2004.
- Ito, T. and Follows, M.: Preformed phosphate, soft tissue pump and atmospheric CO₂, *J. Mar. Res.*, 63(4), 813–839, doi:10.1357/0022240054663231, 2005.
- Jaccard, S., Haug, G., Sigman, D., Pedersen, T., Thierstein, H., and Röhl, U.: Glacial/Interglacial Changes in Subarctic North Pacific Stratification, *Science*, 308, 1003–1006, 2005.

- Jaccard, S., Galbraith, E., Sigman, D., Haug, G., Francois, R., Pedersen, T., Dulski, P., and Thierstein, H.: Subarctic Pacific evidence for a glacial deepening of the oceanic respired carbon pool, *Earth Planet. Sc. Lett.*, 277, 156–165, 2009.
- Jin, X., Gruber, N., Dunne, J., Sarmiento, J., and Armstrong, R.: Diagnosing the contribution of phytoplankton functional groups to the production and export of particulate organic carbon, CaCO₃, and opal from global nutrient and alkalinity distributions, *Global Biogeochem. Cy.*, 20, GB2015, doi:10.1029/2005GB002532, 2006.
- Joos, F., Gerber, S., Prentice, I., Otto-Bliesner, B., and Valdes, P.: Transient simulations of Holocene atmospheric carbon dioxide and terrestrial carbon since the Last Glacial Maximum, *Global Biogeochem. Cy.*, 18, GB2002, doi:10.1029/2003GB002156, 2004.
- Kalnay, E., Kanamitsu, M., Kistler, R., Collins, W., Deaven, D., Gandin, L., Iredell, M., Saha, S., White, G., Woollen, J., Zhu, Y., Leetmaa, A., Reynolds, R., Chelliah, M., Ebisuzaki, W., Higgins, W., Janowiak, J., Mo, K. C., Ropelewski, C., Wang, J., Jenne, R., and Joseph, D.: The NCEP/NCAR 40-Year Reanalysis Project, *B. Am. Meteorol. Soc.*, 77(3), 437–471, 1996.
- Keigwin, L.: Glacial-Age Hydrography of the Far Northwest Pacific Ocean, *Paleoceanography*, 13, 323–339, 1998.
- Keigwin, L. and Schlegel, M.: Ocean ventilation and sedimentation since the glacial maximum at 3 km in the western North Atlantic, *Geochem. Geophys. Geosy.*, 3(6), 1034, doi:10.1029/2001GC000283, 2002.
- Knox, F. and McElroy, M.: Changes in atmospheric CO₂ – Influence of the marine biota at high-latitude, *J. Geophys. Res.*, 89, 4629–4637, 1984.
- Kohfeld, K. E., Le Quééré, C., Harrison, S. P., and Anderson, R. F.: Role of marine biology in glacial-interglacial CO₂ cycles, *Science*, 308, 74–78, 2005.
- Köhler, P., Fischer, H., Munhoven, G., and Zeebe, R.: Quantitative interpretation of atmospheric carbon records over the last glacial termination, *Global Biogeochem. Cy.*, 19(4), GB4020, doi:10.1029/2004GB002345, 2005.
- Köhler, P., Fischer, H., Schmitt, J., and Munhoven, G.: On the application and interpretation of Keeling plots in paleo climate research - deciphering $\delta^{13}\text{C}$ of atmospheric CO₂ measured in ice cores, *Biogeosciences*, 3, 539–556, doi:10.5194/bg-3-539-2006, 2006a.
- Köhler, P., Muscheler, R., and Fischer, H.: A model-based interpretation of low-frequency changes in the carbon cycle during the last 120,000 years and its implications for the reconstruction of atmospheric $\Delta^{14}\text{C}$, *Geochem., Geophys., Geosy.*, 7, Q11N06, doi:10.1029/2005GC001228, 2006b.
- Kwon, E., Primeau, F., and Sarmiento, J.: The impact of remineralization depth on the air-sea carbon balance, *Nat. Geosci.*, 2, 630–635, 2009.
- Leuenberger, M., Siegenthaler, U., and Langway, C.: Carbon Isotope Composition of Atmospheric CO₂ during the Last Ice-Age from an Antarctic Ice Core, *Nature*, 357, 488–490, 1992.
- Levitus, S. and Boyer, T.: World Ocean Atlas 1994 Volume 4: Temperature, NOAA Atlas NESDIS, US Department of Commerce, Washington, DC, 117 pp., 1994.
- Levitus, S., Burgett, R., and Boyer, T.: World Ocean Atlas 1994 Volume 3: Salinity, NOAA Atlas NESDIS, US Department of Commerce, Washington, DC, 99 pp., 1994.
- Lourantou, A., Lavric, J., Köhler, P., Barnola, J.-M., Paillard, D., Michel, E., Raynaud, D., and Chappellaz, J.: Constraint of the CO₂ rise by new atmospheric carbon isotopic measurements during the last deglaciation, *Global Biogeochem. Cy.*, 24(2), GB2015, doi:10.1029/2009GB003545, 2010.
- Lüthi, D., Floch, M. L., Bereiter, B., Blunier, T., Barnola, J.-M., Siegenthaler, U., Raynaud, D., Jouzel, J., Fischer, H., Kawamura, K., and Stocker, T.: High-resolution carbon dioxide concentration record 650,000–800,000 years before present, *Nature*, 453, 379–382, 2008.
- Maier-Reimer, E., Kriest, I., Segschneider, J., and Wetzel, P.: The Hamburg Ocean Carbon Cycle Model HAMOCC5.1 – Technical Description Release 1.1, Tech. rep., Max Planck Institute for Meteorology, Hamburg, Germany, 2005.
- Marchitto, T., Lynch-Stieglitz, J., and Hemming, S.: Deep Pacific CaCO₃ compensation and glacial-interglacial atmospheric CO₂, *Earth Planet. Sc. Lett.*, 231, 317–336, 2005.
- Marchitto, T., Lehman, S., Ortiz, J., Flückiger, J., and van Geen, A.: Marine radiocarbon evidence for the mechanism of deglacial atmospheric CO₂ rise, *Science*, 316, 1456–1459, 2007.
- Margalef, R.: Life-forms of phytoplankton as survival alternatives in an unstable environment, *Oceanol. Acta*, 1, 493–509, 1978.
- Marinov, I., Gnanadesikan, A., Sarmiento, J. L., Toggweiler, J. R., Follows, M., and Mignone, B. K.: Impact of oceanic circulation on biological carbon storage in the ocean and atmospheric $p\text{CO}_2$, *Global Biogeochem. Cy.* 22(3), GB3007, doi:10.1029/2007GB002958, 2008.
- Martin, J.: Glacial-Interglacial CO₂ change: The iron hypothesis, *Paleoceanography*, 5, 1–13, 1990.
- Martin, J. and Fitzwater, S.: Iron deficiency limits phytoplankton growth in the north-east Pacific subarctic, *Nature*, 331, 341–343, 1988.
- Matsumoto, K., Sarmiento, J., and Brzezinski, M.: Silicic acid leakage from the Southern Ocean: a possible explanation for glacial atmospheric $p\text{CO}_2$, *Global Biogeochem. Cy.*, 16(3), 1031–1054, doi:10.1029/2001GB001442, 2002.
- Matsumoto, K., Hashioka, T., and Yamanaka, Y.: Effect of temperature-dependent organic carbon decay on atmospheric $p\text{CO}_2$, *J. Geophys. Res.*, 112, G02007, doi:10.1029/2006JG000187, 2007.
- McCorkle, D., Heggie, D., and Veeh, H.: Glacial and Holocene Stable Isotope Distributions in the Southeastern Indian Ocean, *Paleoceanography*, 13, 20–34, 1998.
- McManus, J., Francois, R., Gherardi, J.-M., Keigwin, L., and Brown-Leger, S.: Collapse and rapid resumption of Atlantic meridional circulation linked to deglacial climate changes, *Nature*, 428, 834–837, 2004.
- Mehrbach, C., Culbertson, C., Hawley, J., and Pytkowicz, R.: Measurement of the apparent dissociation constant of carbonic acid in seawater at atmospheric pressure, *Limnol. Oceanogr.*, 18, 897–907, 1973.
- Menviel, L., Timmermann, A., Mouchet, A., and Timm, O.: Climate and marine carbon cycle response to changes in the strength of the southern hemispheric westerlies, *Paleoceanography*, 23(4), PA4201, doi:10.1029/2008PA001604, 2008.
- Millero, F.: Thermodynamics of the carbon dioxide system in the oceans, *Geochim. Cosmochim. Acta*, 59, 661–677, 1995.

- Millimann, J. and Droxler, A.: Neritic and pelagic carbonate sedimentation in the marine environment: ignorance is not bliss, *Geol. Rundsch.*, 85, 496–504, 1996.
- Monnin, E., Indermühle, A., Dällenbach, A., Flückiger, J., Stauffer, B., Stocker, T., Raynaud, D., and Barnola, J.: Atmospheric CO₂ concentrations over the Last Glacial Termination, *Science*, 291, 112–114, 2001.
- Mook, W. G.: ¹³C in atmospheric CO₂, *Neth. J. Sea Res.*, 20, 211–223, 1988.
- Mucci, A.: The solubility of calcite and aragonite in seawater at various salinities, temperatures, and on atmosphere total pressure, *Am. J. Sci.*, 283, 780–799, 1983.
- Müller, S., Joos, F., Edwards, N., and Stocker, T.: Water mass distribution and ventilation time scales in a cost-efficient, three-dimensional ocean model, *J. Climate*, 19, 5479–5499, 2006.
- Muscheler, R., Beer, J., Wagner, G., Laj, C., Kissel, C., Raisbeck, G., Yiou, F., and Kubik, P.: Changes in the carbon cycle during the last deglaciation as indicated by the comparison of ¹⁰Be and ¹⁴C records, *Earth Planet. Sc. Lett.*, 238, 325–340, 2004.
- Najjar, R. G., Orr, J., Sabine, C. L., and Joos, F.: Biotic-HOWTO, Internal OCMIP Report, Tech. rep., LSCE/CEA Saclay, Gif-sur-Yvette, France, 15 pp., 1999.
- Ninnemann, U. and Charles, C.: Changes in the mode of Southern Ocean circulation over the last glacial cycle revealed by foraminiferal stable isotopic variability, *Earth Planet. Sc. Lett.*, 201, 383–396, 2002.
- Orr, J., Najjar, R. G., Sabine, C. L., and Joos, F.: Abiotic-HOWTO, Internal OCMIP Report, Tech. rep., LSCE/CEA Saclay, Gif-sur-Yvette, France, 25 pp., 1999.
- Paillard, D. and Parrenin, F.: The Antarctic ice sheet and the triggering of deglaciations, *Earth Planet. Sc. Lett.*, 227, 263–271, 2004.
- Parekh, P., Joos, F., and Müller, S.: A modeling assessment of the interplay between aeolian iron fluxes and iron-binding ligands in controlling carbon dioxide fluctuations during Antarctic warm events, *Paleoceanography*, 23(4), PA4202, doi:10.1029/2007PA001531, 2008.
- Peacock, S., Lane, E., and Restrepo, J.: A possible sequence of events for the generalized glacial-interglacial cycle, *Global Biogeochem. Cy.*, 20(2), GB2010, doi:10.1029/2005GB002448, 2006.
- Primeau, F.: Characterizing Transport between the Surface Mixed Layer and the Ocean Interior with a Forward and Adjoint Global Ocean Transport Model, *J. Phys. Oceanogr.*, 35, 545–564, 2005.
- Ridgwell, A., Watson, A., and Archer, D.: Modeling the response of the oceanic Si inventory to perturbation, and consequences for atmospheric CO₂, *Global Biogeochem. Cy.*, 16(4), 1071–1097, doi:10.1029/2002GB001877, 2002.
- Ridgwell, A., Watson, A., Maslin, M., and Kaplan, J.: Implications of coral reef buildup for the controls on atmospheric CO₂ since the Last Glacial Maximum, *Paleoceanography*, 18(4), 1083–1093, doi:10.1029/2003PA000893, 2003.
- Sarmiento, J. and Gruber, N.: Ocean biogeochemical dynamics, chap. Chapter 8: Carbon Cycle, Princeton University Press, 318–358, 2006.
- Sarmiento, J. and Toggweiler, J.: A new model for the role of the oceans in determining atmospheric *p*CO₂, *Nature*, 308, 621–624, 1984.
- Schmittner, A. and Galbraith, E.: Glacial greenhouse-gas fluctuations controlled by ocean circulation changes, *Nature*, 456, 373–376, 2008.
- Schmittner, A., Brooke, E., and Ahn, J.: Ocean Circulation: Mechanisms and Impacts, vol. 173, chap. Impact of the ocean's overturning circulation on atmospheric CO₂, AGU Geophysical Monograph Series, 209–246, 2007.
- Severinghaus, J. and Brook, E.: Abrupt Climate Change at the End of the Last Glacial Period Inferred from Trapped Air in Polar Ice, *Science*, 286, 930–934, 1999.
- Shackleton, N.: Carbon-13 in *uvigerina*: Tropical rainforest history and the equatorial Pacific carbonate dissolution cycles, in: *The Fate of Fossil Fuel CO₂ in the Ocean*, edited by: Andersen, N. and Malahoff, A., Plenum, New York, 401–428, 1977.
- Siegenthaler, U. and Muennich, K. O.: Carbon-13/carbon-12 fractionation during carbon dioxide transfer from air to sea, *SCORE 16: Carbon Cycle Modelling*, edited by: Bolin, B., Wiley, Chichester, 249–257, 1981.
- Siegenthaler, U. and Oeschger, H.: Biospheric CO₂ emissions during the past 200 years reconstructed by deconvolution of ice core data, *Tellus B*, 39, 140–154, 1987.
- Siegenthaler, U. and Wenk, T.: Rapid atmospheric CO₂ variations and ocean circulation, *Nature*, 308, 624–626, 1984.
- Sigman, D. and Boyle, E.: Glacial/interglacial variations in atmospheric carbon dioxide, *Nature*, 407, 859–869, 2000.
- Sigman, D., McCorkle, D., and Martin, W.: The calcite lysocline as a constraint on glacial/interglacial low latitude production changes, *Global Biogeochem. Cy.*, 12, 409–428, 1998.
- Sigman, D., Jaccard, S., and Haug, G.: Polar ocean stratification in a cold climate, *Nature*, 428, 59–63, 2004.
- Skinner, L., Fallon, S., Waelbroeck, C., Michel, E., and Barker, S.: Ventilation of the Deep Southern Ocean and Deglacial CO₂ Rise, *Science*, 328, 1147–1151, doi:10.1126/science.1183627, 2010.
- Smith, H., Fischer, H., Wahlen, M., Mastroianni, D., Beck, D., and Beck, B.: Dual modes of the carbon cycle since the Last Glacial Maximum, *Nature*, 400, 248–250, 1999.
- Spence, P., Saenko, O., Eby, M., and Weaver, A.: The Southern Ocean Overturning: Parameterized versus Permitted Eddies, *J. Phys. Oceanogr.*, 39, 1634–1651, 2009.
- Spero, H. and Lea, D.: The Cause of Carbon Isotope Minimum Events on Glacial Terminations, *Science*, 296, 522–525, 2002.
- Stenni, B., Masson-Delmotte, V., Johnsen, S., Jouzel, J., Longinelli, A., Monnin, E., Rothlisberger, R., and Selmo, E.: An Oceanic Cold Reversal During the Last Deglaciation, *Science*, 293, 2074–2077, doi:10.1126/science.1059702, 2001.
- Talley, L., Reid, J., and Robbins, P.: Data-based meridional overturning streamfunctions for the global oceans, *J. Climate*, 16, 3213–3226, 2003.
- Toggweiler, J. R., Russell, J., and Carson, S.: Midlatitude westerlies, atmospheric CO₂, and climate change during ice ages, *Paleoceanography*, 21(2), PA2005, doi:10.1029/2005PA001154, 2006.
- Treguer, P., Nelson, D., DeMaster, A., Leynard, A., and Queguiner, B.: The silica balance in the world ocean: A reestimate, *Science*, 268, 375–379, 1995.
- Tschumi, T.: Modeling the ocean's contribution to past and future changes in the global carbon cycle, Ph.D. thesis, Climate and Environmental Physics, Physics Institute, University of Bern, 2009.

- Tschumi, T., Joos, F., and Parekh, P.: How important are Southern Hemisphere wind changes for low glacial carbon dioxide? A model study, *Paleoceanography*, 23(4), PA4208, doi:10.1029/2008PA001592, 2008.
- Ullman, W. and Aller, R.: Diffusion coefficients in nearshore marine environments, *Limnol. Oceanogr.*, 27, 552–556, 1982.
- Volk, T. and Hoffert, M.: Ocean carbon pumps: Analysis of relative strengths and efficiencies in ocean driven atmospheric $p\text{CO}_2$ changes, vol. 32 of *Geophys. Monogr. Ser.*, edited by: Sundquist, E. T. and Broecker, W. S., AGU, Washington, DC, 99–110, 1985.
- Wallmann, K.: Phosphorous imbalance in the global ocean?, *Global Biogeochem. Cy.*, 24(4), GB4030, doi:10.1029/2009GB003643, 2010.
- Watson, A. and Naveira-Garabato, A.: The role of Southern Ocean mixing and upwelling in glacial-interglacial atmospheric CO₂ change, *Tellus B*, 58(15), 73–87, 2006.
- Zeng, N.: Glacial-Interglacial Atmospheric CO₂ Change – The Glacial Burial Hypothesis, *Adv. Atmos. Sci.*, 20, 677–693, 2003.