Electrostatics and flexibility drive membrane recognition and early penetration by the antimicrobial peptide dendrimer **bH1**†

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Harish K. Ravi,^a Michaela Stach,^a Thereza A. Soares,^b Tamis Darbre,^a Jean-Louis Reymond^a and Michele Cascella^{*a}

Molecular dynamics simulations of the polycationic antimicrobial peptide dendrimer **bH1** ($\text{Leu}_8(\text{DapLeu})_4(\text{DapPhe})_2\text{DapLys-NH}_2$) binding to membranes suggest that electrostatic interactions with the polyanionic lipopolysaccharide (LPS) and conformational flexibility of the 2,3-diaminopropanoic acid (*Dap*) branching units drive its selective insertion into microbial membranes.

Antibiotic resistant bacteria represent one of the major health care problems today, in particular Gram-negative pathogens such as *Pseudomonas aeruginosa* (PA), which are lethal to cystic fibrosis and immunocompromised patients.¹ Recently, we reported the peptide dendrimer **bH1** (Fig. 1)² as a new type of membrane disrupting antimicrobial peptide (AMP)³ potentially suitable to address antibiotic resistance (Fig. 1). Dendrimer **bH1** shows potent antimicrobial activity against Gram-negative *E. coli* (MIC = 1 μM) and *P. aeruginosa* (MIC = 5 μM) without showing significant hemolysis (MHC > 500 μM).

Most AMPs are polycationic amphiphilic sequences, which disrupt bacterial membranes by electrostatic binding to anionic groups of LPS or phospholipids followed by hydrophobicity driven membrane penetration,³ which may lead to membrane lysis or cell penetration and subsequent interaction with an intracellular target.⁴ To understand the selective interaction of **bH1** with the LPS coated outer membrane of PA *versus* the eukaryotic cell membrane, we performed all atom molecular dynamics (MD) simulations of the dendrimer^{5–8} interacting with membrane models constituted of a 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphocholine (POPC) bilayer as a model of the eukaryotic membrane, or of an asymmetric LPS on top of a 1,2-dipalmitoyl-3-phosphatidyl-ethanolamine (DPPE) layer as a model of the PA outer membrane (structure in ESI†).

Fig. 1 Structural formula of antimicrobial peptide dendrimer **bH1**.

Standard MD protocols were followed using force field parameters and thermally equilibrated coordinates reported for POPC^{9a} and LPS.^{9b}

In both membrane systems **bH1** approached the membrane surface in the first 10 ns of the simulation and established contacts *via* its multiple N-terminal ammonium groups. In **bH1**:POPC, the contacts involved one common terminal dendron Leu_2Dap of **bH1** binding to the phosphate groups of POPC present at the water-membrane interface. The dendrimer-membrane interface did not evolve significantly within the simulated time (500 ns) and the whole dendrimer structure remained fully solvated in the aqueous phase (ESI,† Fig. S4).

In PA the outer phospho-saccharide chains of LPS assemble in a structured multilayer characterized by a high density of negative electrostatic charges, and a strong hydrophilic character. In our MD simulations, **bH1** made contacts with the membrane surface within the first 15 ns and involved the positively charged Leu_2Dap terminal dendron and the solvent-exposed LPS 0- β -D-glucose (0GB) and 0- α -D-glucose (0GA) moieties (Fig. 2E). Further penetration occurred around 18 ns by transient formation of hydrogen bonds between 6- α -D-glucose (6GA) in the LPS and the terminal leucines attached to the same *Dap* unit following a similar

^a Department of Chemistry and Biochemistry, University of Bern, Freiestrasse 3, 3012 Bern, Switzerland. E-mail: michele.cascella@iac.unibe.ch;

Fax: +41 31 631 3994; Tel: +41 31 631 4256

^b Department of Fundamental Chemistry, Federal University of Pernambuco,

Av. Jornalista Anibal Fernandes, s/n° Cidade Universitária 50740-560, Recife, Brazil

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Fig. 2 Interaction of the **bH1** dendrimer with the LPS-DPPE membrane. Upper panels (A–D) – insertion of **bH1** into the membrane as a function of time. **bH1** is shown in van der Waals spheres. Lower panels (E–H) – interactions of *Leu₂Dap* with various residues in the LPS as a function of time. Residues involved in H-bonding are drawn in licorice. Colored lines represent the rest of **bH1**, gray lines represent the rest of the LPS. Blue dashed lines show hydrogen bonds. Panel I: Ramachandran plot for *Dap*. Black squares represent conformations explored during MD of the **bH1**:LPS system. While *Leu₂Dap* is in water (like geometry A in the left panel), it explores extended conformations. Early contact with the LPS (E) induces coiling. Complete insertion into the LPS (F to G) is accompanied by recovery of the extended conformation, kept along the rest of the simulation (H, B, C and D).

hydrogen bond acceptor–donor pattern as before (Fig. 2F). Between 20 and 120 ns **bH1** underwent a major conformational change facilitating the insertion of a second *Leu₂Dap* dendron into the LPS, concomitant with deeper penetration of the leucines of the first *Leu₂Dap* dendron to ~ 18 Å from the water–LPS interface, enabling its anchoring to the α -glycero- β -manno-heptose-7-formamide (3H1) and phosphorylated 2-(2-hydroxyethyl)-6-deoxy- β -manno-heptose (PH2) of the LPS, whereby the three anionic phosphate groups of PH2 formed hydrogen bonds with the N-terminal ammonium groups of **bH1** (Fig. 2H).

The contacts of the *Leu₂Dap* dendrons to 3H1 and PH2 were very stable, and no further penetration occurred within our simulation time (300 ns). Rather, the deep rooting of the first *Leu₂Dap* dendron into the LPS induced penetration of the other ones. In particular, after about 100 ns, one of the other three remaining *Leu₂Dap* dendrons started to interact with the outer-core of the LPS, and then followed a similar hydrogen-bonding pattern observed for the initial anchoring *Leu₂Dap* dendron. At the end of the MD simulation this second *Leu₂Dap* dendron had penetrated well inside the outer core of the LPS and its terminal leucines were just 6 Å away from the inner core of the LPS.

Monitoring the insertion of the third and the fourth *Leu₂Dap* dendrons, we observed a hydrogen-bonding pattern with the outer core identical to that of the first two dendrons. Translocation of the various terminal *Leu₂Dap* dendrons of **bH1** from the outer to the inner region of the LPS occurred on average in 100 ns by a diffusive mechanism in which the dendrimer built relatively weak and labile contacts with the various sugar moieties, in contrast to the initial insertion of the *Leu₂Dap* dendrons into the LPS, which required some activation.

The *Dap* residue of the terminal *Leu₂Dap* dendrons underwent significant conformational changes in the course of the MD simulation. Initially the dendrimer was fully hydrated and the dendrons assumed an extended conformation in which the *Dap* residue was present in the allowed β -sheet region of the Ramachandran plot (Fig. 2). Upon localization at the membrane–water interface, this extended conformation was destabilized and *Dap* assumed a coiled conformation typical for a right-handed α -helix. The β -branch of the asymmetric *Leu₂Dap* dendron is longer by one methylene group and conformationally more flexible than the α -branch, and was always the first to penetrate the LPS membrane, followed by *Dap* and the shorter α -branch. After the *Leu₂Dap* dendron had passed the region defined by the 0GB and 0GA saccharides and insertion into the LPS was complete, the secondary structure of *Dap* changed back to an extended conformation similar to that in the aqueous phase. Penetration of **bH1** into the LPS thus seems to be favored by the conformational flexibility of the branching units, which allow local coiling and optimization of the interaction with the LPS at the water–membrane interface.

Electrostatic attraction between the positively charged terminal ends and complementary negatively charged phosphate groups provided the driving force inducing penetration of **bH1** into the LPS during MD. The dendrimer insertion occurred through an entropy-driven process, with sequential binding of the different branches. The first layer of phosphate groups present in the LPS at about 18 Å from the membrane–water interface constituted the main basin of attraction for the terminal *Leu₂Dap* dendrons. The simulated time (300 ns) was sufficient to observe localization of the dendrimer in this area of the LPS. The presence of a second

Fig. 3 **bh1** induced CF-release from large unilamellar vesicles. (a) Phosphatidylglycerol (PG) LUV. (b) Phosphatidylcholine (PC) LUV.

layer of phosphates roughly 26 Å deeper in the LPS may trigger further penetration in longer timescales.

The importance of electrostatic complementarity for the activity of **bh1** was evidenced earlier by the observation that the uncharged, N-acetylated analog of **bh1** shows no antibacterial or membrane disrupting effect.² To further test this interaction, **bh1** was exposed to 5(6)-carboxyfluorescein (CF) loaded large unilamellar vesicles (LUVs) composed of phosphatidylglycerol (PG) as head groups, and induced CF release at low concentration (1–30 $\mu\text{g mL}^{-1}$) demonstrating an interaction and disruption of this membrane. In contrast, LUVs composed of phosphatidylcholine (PC) head groups showed only partial release of CF upon exposure to **bh1** up to 200 $\mu\text{g mL}^{-1}$, showing that **bh1** does not interact with zwitterionic groups (Fig. 3). **bh1** also shows high potency against Gram positive *Bacillus subtilis* (MIC = 1 μM).² Based on the present simulation we predict that **bh1** can interact with negatively charged peptidoglycan elements, like teichoic acids, which are key to the structural stability of the outer cell wall.

In conclusion, the driving force facilitating penetration of the antimicrobial peptide dendrimer **bh1** in the LPS of PA during MD was provided by electrostatic attractions exerted by the phosphate groups localized at the inner regions of the LPS layer on the external side of the membrane, which were established after initial membrane adsorption and sequential insertion of the four terminal Leu_2Dap dendrons, an event requiring local

coiling of the branching *Dap* amino acid to facilitate adaptation to the geometrical restraints at the membrane–water interface. The observation of these coiling events at the interface-crossing point suggests that flexible branching units are crucial to the antimicrobial activity of peptide dendrimers, a finding which may be useful in future designs. Note that in the simulation time presented here the LPS bilayer was still well organized and only marginally perturbed by the dendrimer. It is likely that the antimicrobial function of **bh1** is associated with phenomena occurring at longer timescales, possibly comprising aggregation of multiple **bh1** units into the LPS. A crucial role played by aggregation has been already described for membrane-binding antibacterial linear peptides.¹⁰ Future studies addressing longer MD simulations, also with a higher concentration of **bh1** molecules, may be able to elucidate this other aspect of the biological activity of peptide dendrimers.

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Notes and references

- (a) R. M. Aris, P. H. Gilligan, I. P. Neuringer, K. K. Gott, J. Rea and J. R. Yakaskas, *Am. J. Respir. Crit. Care Med.*, 1997, **155**, 1699–1704; (b) A. Harris, C. Torres-Viera, L. Venkataraman, P. DeGirolami, M. Samore and M. Y. Carmeli, *Clin. Infect. Dis.*, 1997, **28**, 1128–1133; (c) P. Hsueh, P. L. Teng, P. Yang, U. Chen, S. Ho and K. Luh, *J. Clin. Microbiol.*, 1998, **36**, 1347–1351.
- (a) M. Stach, N. Maillard, R. U. Kadam, D. Kalbermatter, M. Meury, M. G. P. Page, D. Fotiadis, T. Darbre and J.-L. Reymond, *MedChemComm*, 2012, **3**, 86–89; (b) T. Darbre and J.-L. Reymond, *Org. Biomol. Chem.*, 2012, **10**, 1483–1492.
- M. Bruschi, G. Pirri, A. Giuliani, S. F. Nicoletto, I. Baster, M. A. Scociapino, M. Casu and A. C. Rinaldi, *Peptides*, 2010, **31**, 1459–1467.
- K. A. Brogden, *Nat. Rev. Microbiol.*, 2005, **3**, 328–350.
- S. V. Lyulin, A. Darinskii and S. V. Lyulin, *Phys. Rev. E*, 2008, **78**, 041801.
- G. M. Pavan, A. Barducci, L. Albertazzi and M. Parrinello, *Soft Matter*, 2013, **9**, 2593–2597.
- P. Welch and M. Muthukumar, *Macromolecules*, 2000, **33**, 6159–6167.
- K. Karatasos, D. B. Adolf and G. R. Davies, *J. Chem. Phys.*, 2001, **115**, 5310–5318.
- (a) A. V. Vargiu, F. Collu, R. Schulz, K. M. Pos, M. Zacharias, U. Kleinekathöfer and P. Ruggerone, *J. Am. Chem. Soc.*, 2011, **28**, 10704–10708; (b) K. N. Kirschner, R. D. Lins, A. Maass and T. A. Soares, *J. Chem. Theory Comput.*, 2012, **8**, 4719–4731.
- (a) R. E. Hancock and D. S. Chapple, *Antimicrob. Agents Chemother.*, 1999, **43**, 1317–1323; (b) K. Matsuzaki, *Biochim. Biophys. Acta*, 1998, **1376**, 391–400; (c) Z. Oren and Y. Shai, *Biopolymers*, 1998, **47**, 451–463; (d) Y. Bai, S. P. Liu, J. G. Li, R. Lakshminarayanan, P. Sarawathi, C. Tang, D. C. Ho, C. Verma, R. W. Bauerman and K. Perushin, *J. Biol. Chem.*, 2012, **287**, 26606–26617; (e) M. M. Domingues, N. C. Santos and M. A. R. B. Castanho, *Curr. Protein Pept. Sci.*, 2012, **7**, 611–619; (f) S. Bobone, Y. Gerelli, M. De Zotti, G. Bocchinfuso, A. Farrotti, B. Orioni, F. Sebastiani, E. Latter, J. Penfold, R. Senesi, F. Formaggio, A. Palleschi, C. Toniolo, G. Fragneto and L. Stella, *Biochim. Biophys. Acta*, 2013, **1828**, 1013–1024; (g) V. P. Zhodanov and F. Hook, *Phys. Rev. E*, 2013, **87**, 04718.