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High pressure breakdown of antigorite to spinifex-textured olivine and orthopyroxene, SE Spain

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Abstract The prograde, high pressure, transition from antigorite serpentinite to enstatite-olivine rock occurs along a tectonically undisturbed profile at Cerro del Almirez, SE Spain. The reactant assemblage is antigorite + olivine with tremolite rimming precursor diopside. The product assemblage of tremolite + chlorite + enstatite + olivine has a spinifex-like texture with arborescent or radiating olivine elongated parallel to [001] and with radially grown enstatite. Product enstatite is very poor in Al_2O_3 . Due to numerous oriented submicroscopic inclusions of chromian magnetite, product olivine has a brownish pleochroism and a bulk chromium content similar to precursor antigorite. Titanian clinohumite with a fluorine content of 0.45–0.50 wt% persisted beyond the breakdown of antigorite. The partitioning of iron and magnesium amongst the silicate phases is almost identical to that at lower pressures. Average K_d values Mn/Mg and Ni/Mg are 0.17 and 0.70 for antigorite-olivine pairs and 1.83 and 0.22 for orthopyroxene-olivine pairs, respectively. These data are useful in discriminating generations of olivine grown on each other. From the field data a phase diagram topology for a portion of the system $\text{CaO-MgO-SiO}_2\text{-H}_2\text{O}$ is derived. This topology forms the basis for extrapolations into inaccessible P - T regions.

Introduction

Subduction of serpentinitized lithospheric mantle is considered an important process, because its dehydration provides a source of H_2O that may trigger partial melting and affect rheological and seismic properties of mantle rocks. The knowledge of phase relations in subducting serpentinite is important for monitoring these processes. Therefore, in recent years, a great number of experimental studies have been carried out in order to determine high pressure phase relations of hydrous ultramafic systems (Yamamoto and Akimoto 1977; Pawley and Wood 1995, 1996; Wunder and Schreyer 1997; Bose and Ganguly 1995; Ulmer and Trommsdorff 1995). Attempts to model these systems thermodynamically for pressures and temperatures corresponding to various subduction zone environments (Evans and Guggenheim 1988; Pawley and Wood 1995) have been only partly successful and are faced with increasing difficulties, particularly at high pressures, because of the lack of relevant thermodynamic data. In contrast to experimental work and thermodynamic modelling the field evidence for high pressure metamorphism of serpentinite has received relatively little attention. For example, the major sources of H_2O in subducting hydrous mantle, the breakdown reactions of antigorite and talc have been studied repeatedly in experiments and theory but only a part of them have been recovered in the field. The relative stability fields of talc and antigorite parageneses, which are crucial in evaluating the phase relations of subducting serpentinites, are only incompletely known for parts of the lithosphere. Trommsdorff and Evans (1972, 1974) demonstrated that at moderate pressures of up to ~ 10 kbar antigorite breaks down to forsterite + talc at temperatures below 600°C . Depending on pressure this product assemblage is then consumed by various dehydration steps until the ultimate product, enstatite + olivine, is formed. According to experimental studies (Ulmer and Trommsdorff 1995; Wunder and Schreyer 1997) and internally

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consistent thermodynamic data (Berman et al. 1986; Berman 1988) the intermediate dehydration products are absent at high pressures and antigorite breaks down directly to enstatite + olivine. Although antigorite is known to exist to conditions of eclogite facies (Bearth 1967; Evans 1977; Scambelluri et al. 1995), the direct breakdown of antigorite to enstatite + olivine remains to be demonstrated in the field (Evans and Guggenheim 1988, p. 267). This breakdown reaction, however, is considered to be a major source of H₂O for calc-alkaline magmatism (Ulmer and Trommsdorff 1995). It is the purpose of this study to present a case where this reaction can be demonstrated in the field and to derive from the relations of phases coexisting in the field a phase diagram topology relevant for ultramafic rocks.

Geological setting

The studied ultramafic rocks belong to the Nevado-Filábride Complex, the lowermost tectonic complex of the Internal Zones of the Cordilleras Béticas (Fig. 1). They are tectonically emplaced in the upper part (Mulhacén Unit) of this complex. The ultramafics are on the top of a sequence thinned by extensional tectonics. This sequence consists of white micaschists, quartzites and occasional marble lenses, that overlie graphite-bearing micaschists and quartzites which are considered to be of Paleozoic age. Late brittle contacts separate the ultramafic rocks from the other rock types. During the Alpine Orogeny the Mulhacén Unit underwent a Paleogene compressional event, in which *HP* conditions (about 2.0 GPa and 500–600 °C) were reached (Gómez-Pugnaire and Fernández-Soler 1987). Exhumation during the Oligocene was essentially decompressive, with a metamorphic peak at about 0.6 GPa and 600–650 °C (Nijhuis 1964; Gómez-Pugnaire and Fernández-Soler 1987; Bakker et al. 1989). The exhumation path passed through greenschist facies conditions during the Burdigalian.

Subsequently, an important extensional event took place which obliterated most of the previous structures. This event is responsible for the present position of the ultramafic rocks (García Dueñas et al. 1992; González Lodeiro et al. 1996).

The ultramafic rocks from the Almirez area that display a well-developed spinifex-like texture were originally interpreted as the product of near surface crystallization of a crystal-melt mush of harzburgitic composition that was quickly quenched and then partly transformed into antigorite serpentinite (Burgos et al. 1980; Morten and Puga 1984; Puga 1990). A secondary, metamorphic origin for the spinifex-like texture was later proposed by Bodinier et al. (1993) and by Puga et al. (1995, 1997) on basis of the chemical features and by Müntener et al. (1997) on the basis of phase relations.

The ultramafic rocks

The ultramafic rocks of Cerro del Almirez (Jansen 1936) form three major bodies that are parts of a thrust sheet covering an area of three km² (Fig. 1). The ultramafic thrust sheet is about 400 m thick and is built up of two rock types. One of them occurs in the upper 200 m of the Cerro del Almirez mountain and is formed of schistose antigorite serpentinite. The second, tectonically lower rock type is more massive and forms various enstatite-forsterite rocks in which the arborescent, radiating, spinifex-like textures dominate (Fig. 2). The two minor ultramafic bodies west of the Cerro del Almirez (Fig. 1) also belong to this rock type. The contact between the two rock types is sharp but irregular and shows no evidence of a tectonic discontinuity.

A localized faint banding is recognised in all ultramafic rocks. The banding results from alternating cm to dm thick layers varying in clinopyroxene content. In places clinopyroxenite up to 1 m thickness occurs. Rodingite boudins, which in places cross the layering, occur over the entire ultramafic body. Titanium clinohumite occurs as knobs or deformed veins together with diopside and olivine in the serpentinites and locally as specks in the enstatite-olivine rocks.

The assemblage of rocks in the serpentinite part of Cerro del Almirez is identical to that of many antigorite serpentinites in the Penninic zone of the Alps (Bearth 1967).

Fig. 1 Outline geological map of the Nevado Filábride Complex with Cerro del Almirez details

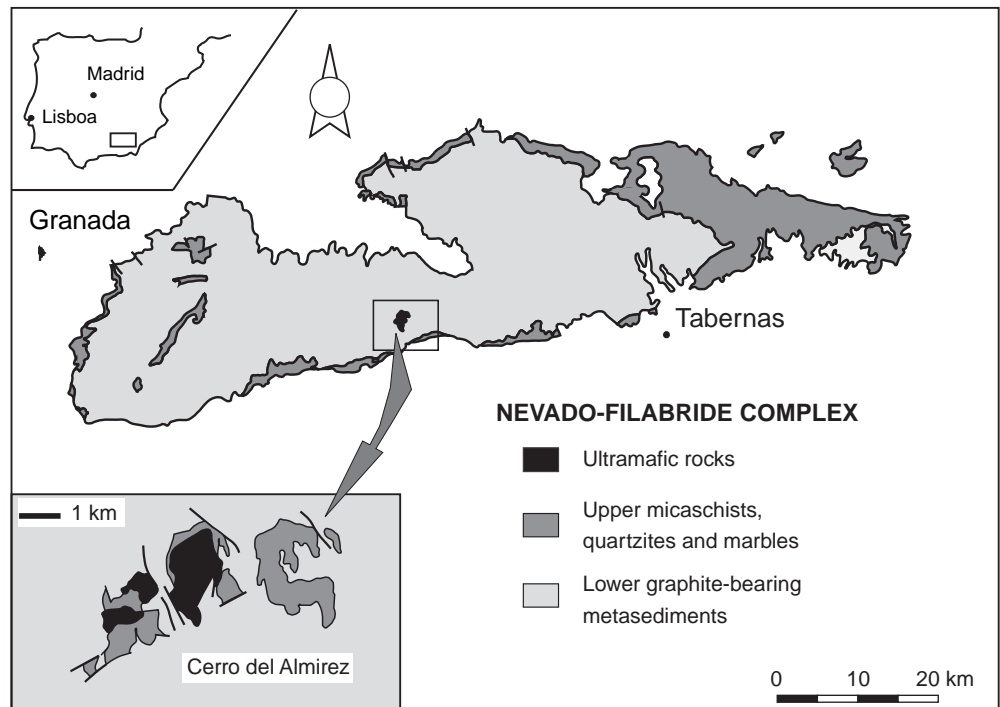


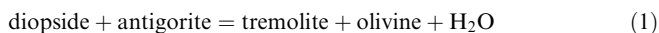
Fig. 2 Spinifex-like, arborescent olivine (*dark*) in an enstatite + chlorite ± tremolite matrix. Field of view 13 × 7 cm



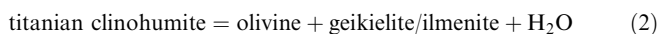
Petrography

Serpentinites

These are made up of a schistose assemblage of antigorite, olivine and diopside, with minor chlorite and discontinuous strings of magnetite. Accessory minerals are pentlandite, pyrrhotite, ilmenite, and titanian clinohumite. Premetamorphic cores of dusty clinopyroxene and of olivine are not uncommon in larger individual crystals. Tremolite associated with olivine is widespread, and frequently rims diopside (compare Plate I A, B in Trommsdorff and Evans 1974), indicating that the reaction



has taken place. Breakdown of OH-titanian clinohumite has been observed locally:



which according to Trommsdorff and Evans (1980) takes place at moderate pressures under the same conditions as reaction (1).

In pyroxenitic layers in the antigorite serpentinites, i.e. antigorite-chlorite-tremolite-diopside rocks, zirconium-rich domains have been found in which zirconolite, zircon and baddeleyite coexist with olivine plus ilmenite. They will be the subject of a separate paper.

Enstatite-olivine rocks

These rocks in the lower part of the sequence (Fig. 3) occur in two varieties: the predominant variety is characterized by a spinifex-like texture, whereas the second variety has a granular texture. The mineralogical composition of both rock types is essentially the same. The assemblage consists of olivine, enstatite, chlorite, tremolite and chromian magnetite. Accessory minerals, in some instances also in major quantities, are pyrite, pyrrhotite, pentlandite, ilmenite and titanian clinohu-

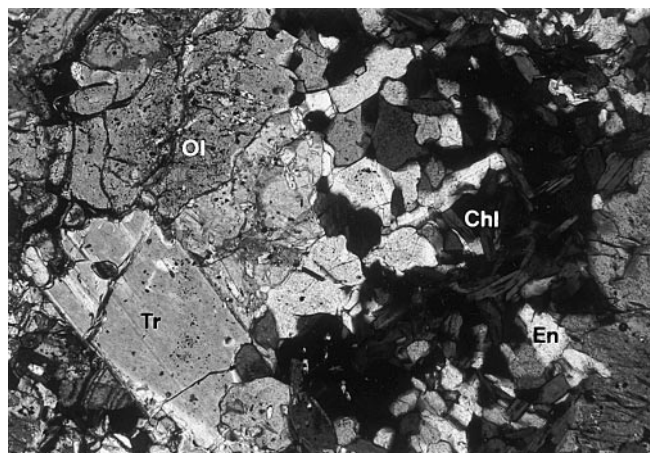


Fig. 3 The product assemblage enstatite-olivine-chlorite-tremolite. Olivine is cut *perpendicular to elongation*. Mineral abbreviations after Kretz (1983). Field of view 1.35 × 0.93 mm

mite. In contrast to the antigorite serpentinites chlorite is quite abundant.

Olivine in the enstatite-olivine rocks occurs in three generations (Fig. 4) that overgrow each other. The first, serpentinite stage, generation forms irregular shaped clear cores within elongate cm-size porphyroblastic spinifex-like olivine which constitutes a second generation. This second, enstatite-olivine stage, generation has a brownish pleochroism and is dusty with small inclusions of chlorite and smaller chromian magnetite. The brownish colour results from numerous oriented submicroscopic inclusions of chromian magnetite, in some cases concentrated in lamellae parallel to (001) and (100). Second generation olivine in places overgrows folded strings of magnetite which are typical for the precursor serpentinite (similar to Plate IC, Trommsdorff and Evans 1974). The third generation of olivine is texturally late, in part granoblastic and minor in size. In the spinifex-like rocks brownish olivines dominate in size and quantity and are up to 10 cm in length. The elongation direction is [001]. Enstatite occurs in two gener-

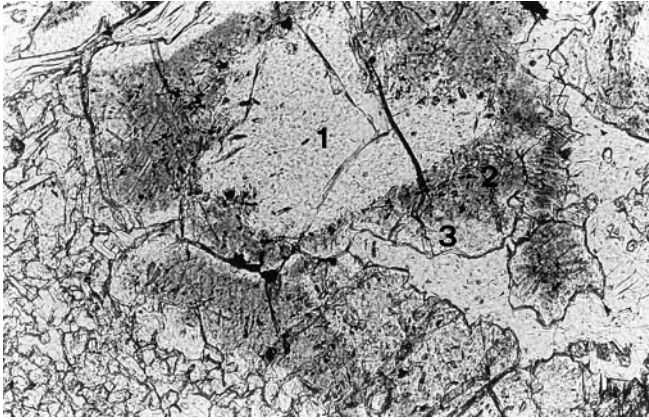


Fig. 4 Three generations of olivine. The *clear core* is inherited from antigorite serpentinite, generation (1); the *dusty zone* is spinifex-like olivine with numerous inclusions of chlorite and chromian magnetite (2) and the *clear rim* is recrystallized olivine belonging to the granoblastic generation (3). Field of view 1.20×0.83 mm

ations. The first of these is radiating (Fig. 5), with a grain size of typically 1 cm, and frequently contains tiny inclusions of chlorite. The second generation of enstatite is represented by small granoblastic grains. Simultaneous crystallization of both spinifex-like olivine and radiating enstatite is suggested by frequent parallel intergrowth of these minerals and by the occurrence of idioblastic microinclusions of enstatite in the cores of brownish olivine (Fig. 6). Tremolite occurs as individual crystals and as radial aggregates and in more pyroxenitic layers comprises up to 10% of the rock. In addition to its occurrence as inclusions, chlorite forms fine grained interstitial aggregates associated with chromian magnetite and enstatite.

Titanian clinohumite is found only locally as granoblasts several mm in size, typically associated with ilmenite, chlorite and chromian magnetite. Breakdown of titanian clinohumite to wormy intergrowths of olivine plus ilmenite is observed in some samples.

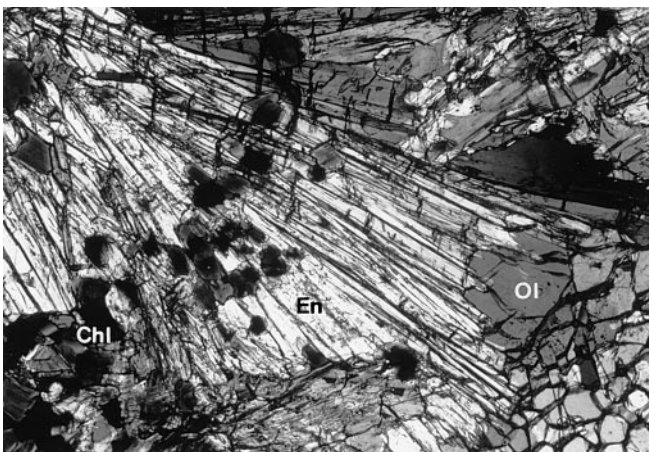


Fig. 5 Radiating enstatite in spinifex-like rock. Field of view 2.7×1.9 mm

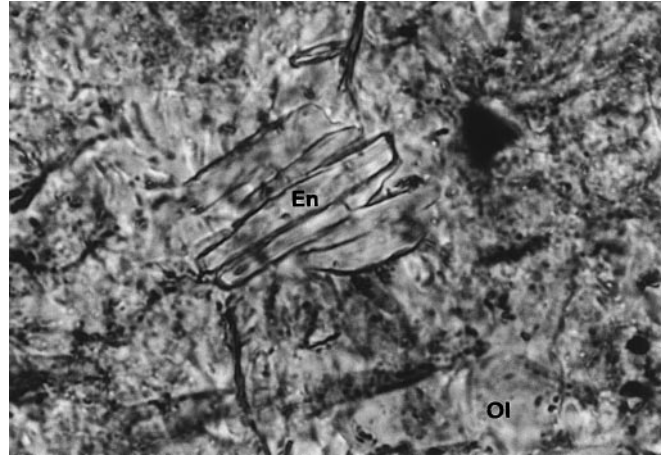


Fig. 6 "Baby enstatite" in centre of elongate olivine. Field of view 0.14×0.10 mm

Among secondary minerals talc is typically replacing enstatite and can reach considerable quantities although it is largely absent from the unaltered rocks. A later local product is secondary chrysotile close to the borders of the ultramafic bodies. No antigorite replacing the enstatite-olivine assemblage has been observed (cf. Puga 1990), but some antigorite bordering chlorite in retrograde talc-rich rocks has been found.

The granoblastic variety of enstatite-olivine rock is considered to be derived from the spinifex-like variety through deformation. It is concentrated in shear bands within the latter rocks in which many relicts of brownish olivine are observed. Radial enstatite on the other hand survived the deformation and remains surrounded by granoblastic olivine and enstatite (Fig. 5).

Mineral compositions

The Table summarizes typical microprobe analyses of the metamorphic minerals from Cerro del Almiraz. Although the overall variations in mineral composition of the various rock types are small there are some notable differences.

Olivine

In terms of iron-magnesium ratios *olivine* varies from Fo 89 to 92, whereas the other minerals show maximum variations in X_{Mg} of 0.01. The order of preference of Fe for the different minerals is similar to the values given by Trommsdorff and Evans (1972, 1974). There is a tendency that all the Mg-silicates are somewhat less Fe-poor relative to olivine than in Trommsdorff and Evans. For Fo 90 the composition of coexisting minerals, expressed as X_{Mg} , are (Fig. 7): enstatite 0.90; antigorite 0.93; chlorite 0.94; diopside 0.96 and tremolite 0.965. Retrograde talc has a constant X_{Mg} of 0.98.

Table 1. Representative analyses of minerals from antigorite serpentinites (Al-95-20), and olivine-enstatite rocks (Al 95-4d, Al 95-R2). [X_{Mg} is Mg atoms/(Mg + Fe + Mn + Ni) atoms]

Sample	Olivine enstatite rock (Al95-4d)					Olivine-enstatite rock (Al95-R2)					Tremolite-bearing serpentinite (Al95-20)					Average relative error (2σ) in %		
	Weight % (core)	Ol	Ol (brown)	Opx	Tr	Chl	Mag	TiCl	Ol	Opx	Chl	Ol	Atg	Tr	Di		Mag	
SiO ₂	40.7	40.1		58.8	58.6	33.1	<0.03	36.4	40.5	57.4	33.7	41.6	41.2	57.8	54.8	<0.03	<0.03	0.6 (at 40% level)
TiO ₂	<0.02	<0.02		<0.02	<0.02	<0.02	0.49	4.87	<0.02	0.05	<0.02	<0.02	<0.02	<0.02	<0.02	<0.02	<0.02	8.2 (at 0.05% level)
Cr ₂ O ₃	<0.03	<0.03		0.17	<0.03	2.35	6.90	0.11	<0.03	0.15	1.06	<0.03	0.49	<0.03	<0.03	<0.03	2.10	4.4 (at 0.4% level)
Al ₂ O ₃	<0.03	<0.03		0.25	0.19	13.7	0.21	<0.03	<0.03	0.21	13.0	<0.03	3.89	0.78	<0.03	<0.03	0.03	2.0 (at 4% level)
Fe ₂ O ₃ ^a						59.8										67.3		
FeO	9.57	8.93		6.18	1.34	3.49	30.7	8.99	10.23	7.18	3.87	6.58	3.65	1.18	0.79	29.0		3.0 (at 7% level)
NiO	0.35	0.44		0.07	0.09	0.29	0.38	0.31	0.42	0.06	0.22	0.24	0.18	0.05	<0.05	0.21		9.6 (at 0.2% level)
MnO	0.23	0.15		0.19	0.06	<0.03	0.15	0.10	0.07	0.13	0.03	0.36	0.05	0.06	0.09	0.28		3.8 (at 0.35% level)
MgO	49.0	50.1		35.6	23.9	34.3	1.02	47.6	48.7	34.7	35.1	51.5	38.1	23.9	18.1	1.14		1.4 (at 20% level)
CaO	<0.02	<0.02		0.11	13.1	<0.02	<0.02	<0.02	<0.02	0.05	<0.02	<0.02	<0.02	<0.02	13.3	25.9	<0.02	1.2 (at 10% level)
Nb ₂ O	<0.03	<0.03		<0.03	0.31	<0.03	<0.03	<0.03	<0.03	<0.03	<0.03	<0.03	<0.03	0.33	<0.03	<0.03		10.6 (at 0.2% level)
K ₂ O					0.08	<0.02					<0.02							12.0 (at 0.1% level)
H ₂ O calc					2.20	12.64		1.44			12.64			12.01	2.19			
F							100.2 ^b	0.45										
Σ	99.9	99.7		101.4	99.9	99.9	100.1	100.1	99.9	99.9	99.6	100.3	99.6	99.6	99.7	100.2		1.6 (at 0.45% level)

Ions calculated on the basis of 3 cations (olivine, spinel), 6 oxygens (orthopyroxene), 23 oxygens and Fe³/Fe^{total} = 0 (amphibole), 36 oxygens (chlorite), 147 oxygens (antigorite) and 13 cations (titanian clinohumite)

Si	1.000	0.980		1.997	7.986	6.290	0.000	3.982	0.995	1.987	6.400	1.004	31.896	7.905	1.992	0.000		
Ti	0.000	0.000		0.001	0.002	0.003	0.014	0.400	0.000	0.001	0.000	0.000	0.006	0.001	0.000	0.004		
Cr	0.001	0.000		0.005	0.000	0.353	0.207	0.010	0.000	0.004	0.159	0.000	0.298	0.003	0.001	0.063		
Al	0.000	0.000		0.010	0.031	3.054	0.010	0.000	0.000	0.009	2.912	0.000	3.551	0.125	0.000	0.001		
Fe ³							1.705									1.928		
Fe ²	0.196	0.182		0.176	0.152	0.554	0.974	0.822	0.210	0.208	0.613	0.133	2.365	0.135	0.024	0.923		
Ni	0.007	0.009		0.002	0.010	0.044	0.012	0.027	0.008	0.002	0.034	0.005	0.112	0.005	0.001	0.007		
Mn	0.005	0.003		0.006	0.007	0.000	0.005	0.009	0.002	0.004	0.004	0.007	0.031	0.007	0.003	0.009		
Mg	1.789	1.823		1.800	4.853	9.701	0.058	7.748	1.784	1.790	9.937	1.850	43.903	4.861	0.983	0.064		
Ca	0.000	0.000		0.004	1.907	0.001	0.000	0.000	0.000	0.002	0.004	0.000	0.000	1.943	1.007	0.001		
Na	0.002	0.000		0.000	0.082	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.026	0.087	0.000	0.000		
K					0.015	0.000												
OH					2.000	16.000		1.047			16.000		62.000	2.000				
F								0.156										
X_{Mg}	0.896	0.904		0.907	0.966	0.942	0.055	0.900	0.890	0.893	0.939	0.927	0.946	0.971	0.972	0.064		

^a Calculated assuming stoichiometry

^b Sum includes 0.56 wt% V₂O₅

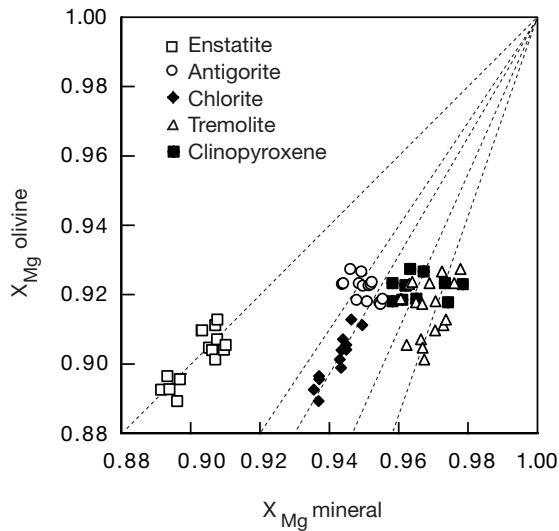


Fig. 7 Partitioning of Mg between olivine and other silicates [X_{Mg} Mg atoms/(Mg + Fe + Mn + Ni) atoms]. Dotted lines are eye ball fits

The generations of metamorphic olivine are distinguishable on the basis of their forsterite contents. Olivine in the serpentinites has Fo 0.92 ± 0.007 . In enstatite-olivine rocks the brown elongate generation spans a range from Fo 89 to 91.5 with the clear cores clustering around Fo 89.5. The neoblast generation is distinctly more magnesian with Fo 91–91.5.

Antigorite

This is quite similar to typical antigorite from Alpine ultramafic rocks (Trommsdorff and Evans 1972) with the exception of trivalent cations which are distinctly higher and range up to over 4 wt% Al_2O_3 and to 0.58 wt% Cr_2O_3 . For example, antigorite Al 95–20 (Table) contains 28 mol% of chlorite component with a composition of those reported for Al 95–4d. The capacity of antigorite to accommodate trivalent cations appears to be maximized at high temperature in the stability field of the tremolite-olivine- antigorite assemblage and near its breakdown. In view of the quite high pressures of over 1.5 GPa at Cerro del Almirez the antigorite breakdown is also at distinctly higher temperatures (≥ 650 °C) than in the Bergell aureole (Trommsdorff and Connolly, 1996, 530 °C at 0.35 GPa.)

Orthopyroxene

This mineral occurs only in the lower section of the Almirez ultramafics and is $En_{90}Fs_{10}$. The radial enstatite reaches a maximum of 0.6 wt% Al_2O_3 in its core and the later, granoblastic, generation has < 0.1 wt% Al_2O_3 , comparable to enstatite-chlorite-forsterite schists of the Central Alps. The zonation of the enstatite in terms of

trivalent cations with higher contents in the core and lower ones in the rims is considered to be retrograde in origin.

Clinopyroxene and tremolite

Both these are close to the pure endmember compositions with low contents of trivalent cations and with a tendency to accommodate some alkalis (diopside up to 0.6 wt% and tremolite up to 0.10 wt% Na_2O ; K_2O in tremolite < 0.10 wt%). In tremolite coexisting with olivine, enstatite and titanian clinohumite traces of fluorine have been detected (< 0.05 wt%).

Chlorite

Chlorite in the spinifex-like rocks is comparable to that in enstatite-chlorite-olivine schists (Trommsdorff and Evans 1974) with high magnesium contents, Cr_2O_3 up to 3 wt% and Al_2O_3 from 12 to 15 wt%.

Titanian-clinohumite

This mineral in the serpentinites is close to the pure (OH)-end-member. In the spinifex-like rocks it typically contains about 0.5 wt% of fluorine and 4.8–5 wt% TiO_2 which translates into $X_{Ti} = 0.4$, where $X_{Ti} = Ti$ p.f.u. of 13 cations; and $X_F = 0.1$, where $X_F = F/(F + OH)$ and $X_{Ti} + X_F = 0.5$. The measured titanian clinohumites are quite homogeneous in their F/Ti ratio and contain systematically about 0.1 wt% Cr_2O_3 .

Chromian magnetite

Chromian magnetite in the spinifex-like rocks is titanium poor (Fenoll Hach-Ali et al. 1991), low in Al_2O_3 (< 2 wt%) and MgO (< 1.5 wt%). The Cr_2O_3 reaches 12 wt% equivalent to 16 wt% of chromite component. The composition compares well with chromian magnetites of tremolite-chlorite-enstatite-olivine rocks from the Central Alps (Evans and Frost 1975, p. 966, analysis n. 4).

Compared to these chromian magnetites the submicroscopic inclusions in olivine seem to contain more Cr_2O_3 and similarly low titanium (semiquantitative energy dispersive transmission electron microscope analysis).

The chemical discrimination of different mineral generations on the basis of minor elements is most obvious in the case of the olivine. This is shown in Fig. 8 where the cation proportions of Ni and Mn are plotted against each other. Olivine from serpentinites is lowest in Ni and highest in Mn; the opposite is true for the large dusty spinifex-like olivine. The clear cores of spinifex-like olivine plot nicely in an intermediate position between these two extremes and the recrystallized genera-

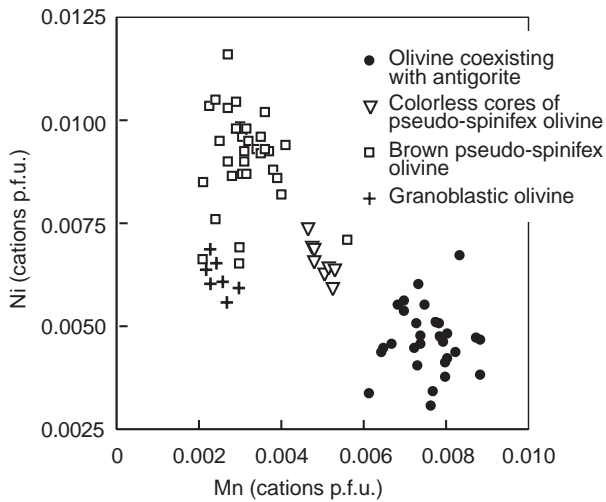


Fig. 8 Ni versus Mn for olivine generations of the serpentinite and the spinifex-like rocks

tion of granoblastic clear olivine has, compared to the spinifex-like one, lower Ni contents. Olivine that coexists with secondary talc spans the range of all generations of the spinifex-like rocks. In sample Al 95-4D a threefold zoned single olivine grain shows clearly the variations (Fig. 9) as discussed above for the whole range of data. The average distribution coefficients (K_D) for Mn/Mg and Ni/Mg for antigorite/olivine pairs are typically 0.17 and 0.70 respectively, and for orthopyroxene/olivine pairs in the spinifex-like rocks 1.83 and 0.22 respectively.

The breakdown of antigorite

The breakdown of antigorite at Cerro del Almirez can be described by an overall mass-balance of the type

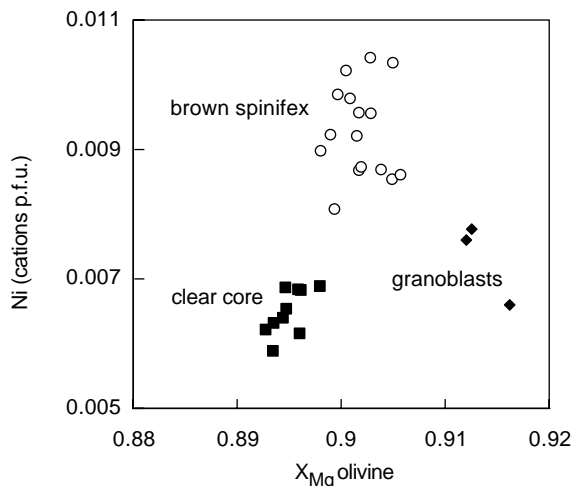
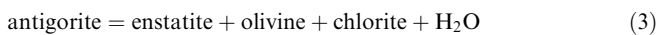
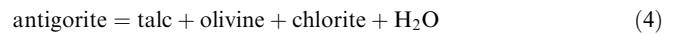
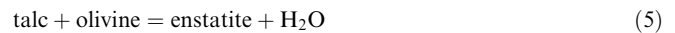


Fig. 9 X_{Mg} versus Ni (cations per formula unit) for three olivine generations in one grain of spinifex-like olivine [X_{Mg} Mg atoms / (Mg + Fe + Mn + Ni) atoms]. (Sample Al-95-4d)

which occurs entirely within the stability field of tremolite plus olivine. Because the product assemblage can also be attained through two intermediate steps (Trommsdorff and Evans 1974, p. 64)



and



arguments and evidence for reaction (3) are here listed in detail. They are as follows:

1. The boundary between serpentinite and pseudo-spinifex chlorite-enstatite-olivine rocks is sharp, but irregular, with no intermediate tectonic discontinuity. The two rock types grade into each other over a distance of no more than a few tens of metres. No intermediate talc zone is observed.

2. Pseudospinifex olivine in places overgrows folded strings of magnetite which are typical for the precursor serpentinite.

3. Parallel growth of arborescent olivine and radiating enstatite indicates simultaneous crystallization. If enstatite grew from talc, olivine would be consumed (reaction 5) and not produced.

4. Chlorite occurs as inclusions in pseudospinifex olivine and enstatite and is far more abundant than in the serpentinites. The clear relic olivine cores are free of chlorite inclusions.

5. The elongation of pseudospinifex olivine is parallel to [001] and not parallel to [010] as observed in the talc-olivine spinifex-like rocks (Evans and Trommsdorff 1974).

6. The submicroscopic inclusions in pseudospinifex olivine are chromian magnetites poor in titanium. The quantity of chromium in olivine plus microinclusions is approximately equivalent to that of the precursor antigorite. Antigorite does not accommodate any titanium.

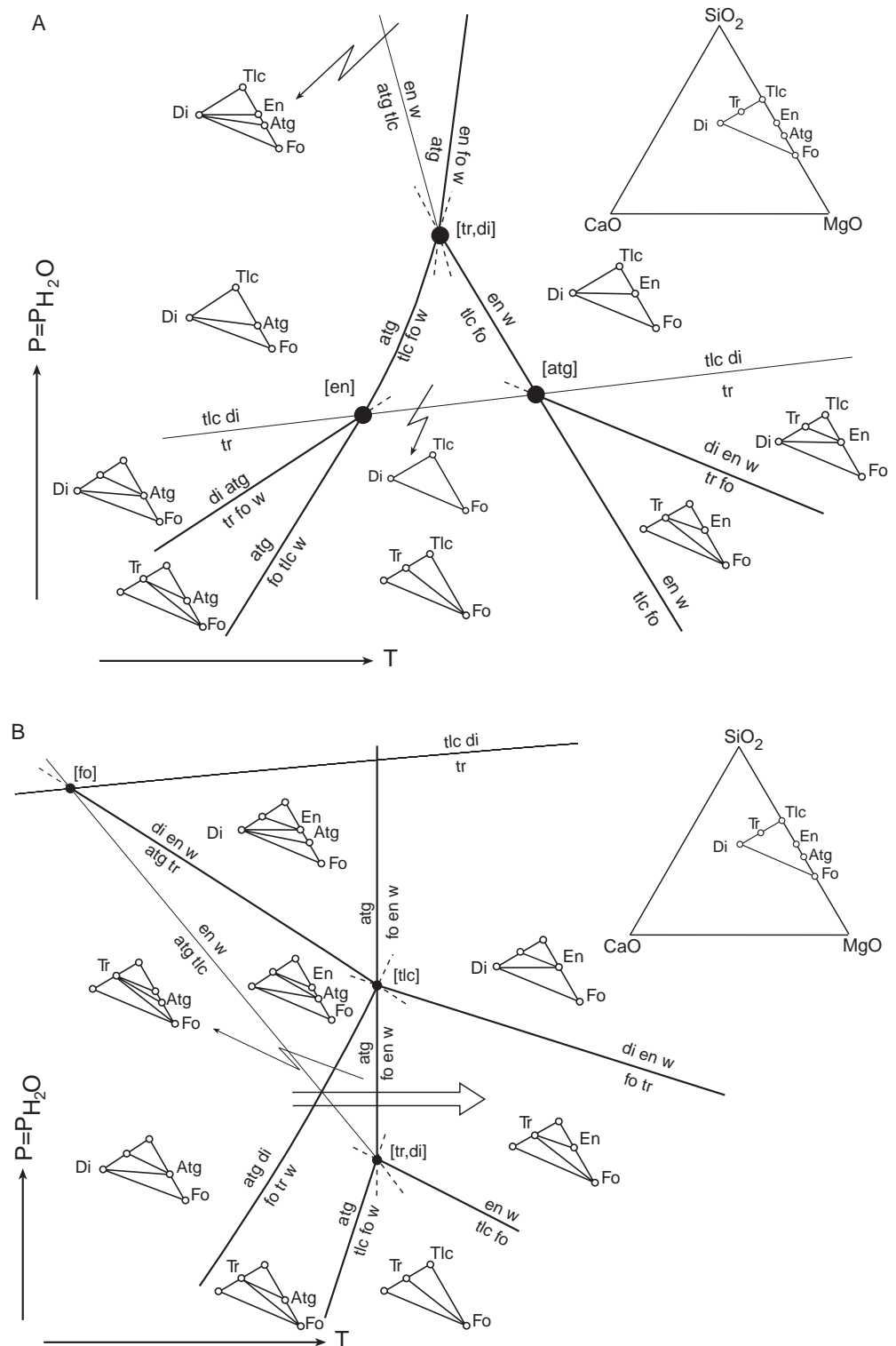
7. Nickel content in spinifex-textured olivine is higher than in olivine of the serpentinites which is opposite to what is observed in olivine + talc rocks growing from serpentinite. The reason for this is that talc incorporates relatively high quantities of Ni but enstatite does not.

8. Manganese is concentrated in olivine of the serpentinites, but in enstatite of the spinifex-like rocks. Thus Mn is low in spinifex-like olivine which is not the case in olivine of talc + olivine rocks.

Evaluation of the stable phase diagram topology for CaO-MgO-SiO₂-H₂O

Because of unresolved differences in the experimental data on antigorite breakdown [Evans et al. 1976; Ulmer and Trommsdorff 1995 (natural antigorite); Wunder and Schreyer 1997 (synthetic, pure antigorite)] and uncertainties in the various thermodynamic data bases due to an undefined structural state of antigorite and lack of an appropriate model for the high pressure and high temperature volumetric properties of tremolite (Comodi

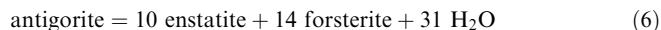
Fig. 10A, B Phase diagram topologies for a portion of the system CaO-MgO-SiO₂-H₂O, defined by the *apices* diopside, talc and forsterite and projected from H₂O. *Thin lines* are for bulk compositions that do not encounter the join forsterite-antigorite. **A** Topology that was not encountered at Cerro del Almirez. **B** Topology fitting the phase relations of Cerro del Almirez with the prograde sequence encountered shown by an arrow



et al. 1991; Berman 1988; Holland and Powell 1990) and talc (Bose and Ganguly 1995, cf. Pawley et al. 1995) it is impossible to derive a relevant phase diagram topology for the system merely on the basis of experimental and thermodynamic data. However, the evidence discussed in this paper permits derivation of the stable phase re-

lations among natural antigorite, forsterite, enstatite, talc, tremolite and diopside. These phase relations are constrained by the prograde sequence of isograd reactions encountered at low to moderate pressures (Evans et al. 1976), and those encountered at high pressure at Cerro del Almirez. Taking the sequence derived by

Evans et al. (1976) and stability of the *equilibrium* involved in reaction (3)



as given, two topologies are possible for the system as shown in Fig. 10 A and B. In both topologies the invariant point [tr, di] is stable (abbreviations after Kretz 1983) at which antigorite, olivine, talc and enstatite coexist with an H₂O-fluid. In topology A additional invariant points [en] and [atg] are stable and [fo] and [tlc] are metastable. The inverse is true for topology B. However, only topology B satisfies the phase relations observed at Cerro del Almirez, because tremolite coexists with the assemblage given by reaction (3). The alternative topology requires, instead of tremolite, stability of diopside + talc which has not been observed.

The prograde sequence of Cerro del Almirez must occur between the invariant points [tr, di] and [tlc] as indicated by an arrow on Fig. 10B. These points fall between temperatures of 640 and 720 °C and 1.5–2.2 GPa (Point [tr, di]) and 1.8–2.2 GPa (Point [tlc]), taking into consideration the various data bases and experimental determinations.

Summary and conclusions

The situation encountered at Cerro del Almirez allowed us to demonstrate for the first time in the field that the breakdown of antigorite to orthopyroxene plus olivine has taken place. The reactants developed distinctive textures with arborescent olivine elongated parallel to [001] and radial enstatite. Jackstraw (criss-cross) textured olivine elongated parallel to [001] and in association with enstatite has been described from the Norwegian Caledonides by Bakke and Korneliussen (1986) where the authors assumed a serpentinite protolith but did not discuss the reaction path to the final orthopyroxene plus olivine rock. Elongation of olivine parallel to [001] is also typical for spinifex textures developed in komatiites (Dickey 1972). Olivine growing according to the breakdown of antigorite to olivine plus talc in some cases is elongated parallel to [010] and tabular parallel to (100) (Evans and Trommsdorff 1974). Snoke and Calk (1978) suggested that the habit of olivine may be an indicator of the physical conditions of crystallization. For all the occurrences of elongate olivine textures the presence of abundant fluid seems critical, independent of whether olivine growth was metamorphic (Evans and Trommsdorff 1974; Snoke and Calk 1978) or magmatic (Grove et al. 1996; Parman et al. 1997). For the breakdown of antigorite at Cerro del Almirez the presence of an abundant, H₂O dominated, fluid phase is assumed in view of the 12 wt% H₂O liberated by this reaction.

The topology derived here from field relationships for high pressure antigorite rocks is strictly valid only for the fluid and the mineral compositions encountered at Cerro del Almirez. In view of the fact that these minerals

are close to the Mg-endmember compositions the pressure/temperature shift in dehydration equilibria due to impurities is trivial with respect to an ideal, pure system. On the other hand, considerable pressure shifts have been calculated for the *intersections* of such equilibria which determine the topology of the phase diagram (Evans and Guggenheim 1988). It may, however, be noted that the topology relevant for Cerro del Almirez is also obtained for the pure system with the data base of Berman (1988, modified 1996).

Many arguments and evidence have been given in this paper for the breakdown of antigorite to enstatite plus olivine. However, unless the breakdown is mappable it may not be easy to find proof for this reaction in other field areas. In most of the metamorphic enstatite-forsterite rocks of the Central Alps enstatite can be found which contains aligned inclusions of tiny olivine (Trommsdorff and Evans 1974, Plate I, E) which indicates growth of enstatite from precursor olivine-talc or olivine-Mg-amphibole rocks. Additional arguments for the alternative breakdown reaction are to be found if bulk compositions are considered that are more siliceous than normal ultramafic rocks (see Fig. 10B); where antigorite and enstatite can coexist.

The persistence of titanian clinohumite in the stability field of enstatite plus olivine at Cerro del Almirez is consistent with its occurrence in high pressure peridotites as well as with recent experimental investigations (Weiss 1997). Although beyond the scope of this paper it may be stated on the basis of the observed phase relations for the Cerro del Almirez rocks that maximum metamorphic pressures were in excess of 1.5 GPa and temperatures in excess of 640 °C. This is consistent with field evidence cited in the geological introduction.

Evidence from nature, as presented here, may be considered as a starting point for the modelling of phase relations in hydrous ultramafic rocks for pressure temperature regimes that are otherwise inaccessible.

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