

Energy transfer and upconversion in Tm³⁺-doped β-NaYF₄: comparison between models and theory

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Tm³⁺ ions in close proximity interact non-radiatively in different ways so that, for example, blue ³H₆→¹G₄ excitation (473 nm) results in UV ¹D₂→³H₆ upconverted emission (361 nm) in a β-NaYF₄: 0.3% Tm³⁺ sample, see Fig 1.

The strength of an energy transfer (ET) interaction can be summarized by a single number: the critical radius R_c. This is the distance at which the probability of ET and radiative decay are equal. At shorter distances ET dominates, while at longer distances radiative decay is more important.

A recent model is able to obtain the R_c of different interactions by fitting luminescence decay curves, see Fig 1(b). [1] For the cross-relaxation interaction shown in Fig. 1(a), this model determines R_c = 11.8 Å. An analysis based on the Inokuti-Hirayama model results in R_c = 12.1 Å.

An equation due to T. Kushida determines R_c as a function of the donor state lifetime, refractive index, transition energy, emission and absorption oscillator strengths, and spectral overlap. [2]

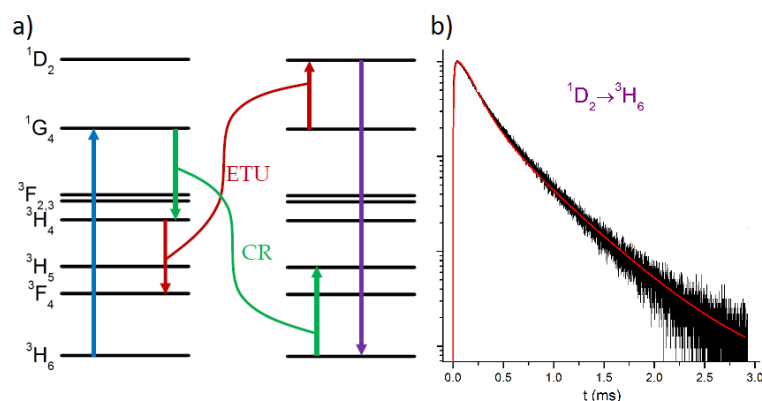


Fig. 1. a) Electronic structure of Tm³⁺ and relevant absorption, emission, and energy transfer processes in β-NaYF₄: 0.3% Tm³⁺. b) ¹D₂→³H₆ luminescence decay curve (361 nm) after ³H₆→¹G₄ excitation (473 nm) with model fit (red line).

These parameters have been determined for a β-NaYF₄: 0.3% Tm³⁺ powder sample and a β-NaGdF₄: Yb³⁺, Tm³⁺ single crystal. [1,3] The predicted critical radius is R_c = 7.6 Å. The disagreement between the model and theory can be explained by the difficulty in measuring the spectral overlap, due to the disorder in the β-NaYF₄ lattice.

[1] Villanueva-Delgado, P.; Krämer, K. W.; Valiente, R., *J. Phys. Chem. C* **119** (2015) 23648.

[2] T. Kushida, *J. Phys. Soc. Jpn.* **34** (1973) 1318.

[3] Villanueva-Delgado, P.; Krämer, *J. Lumin.* (2016) 10.1016/j.jlumin.2016.04.023.